



**TLMC<sub>5</sub>**  
**Quinto Taller Latinoamericano  
de Materiales de Carbono**



# A small history of carbon materials from paint to scaffold biomaterials

(04.11.2024 – Monday – 09:45 – 10:45h )

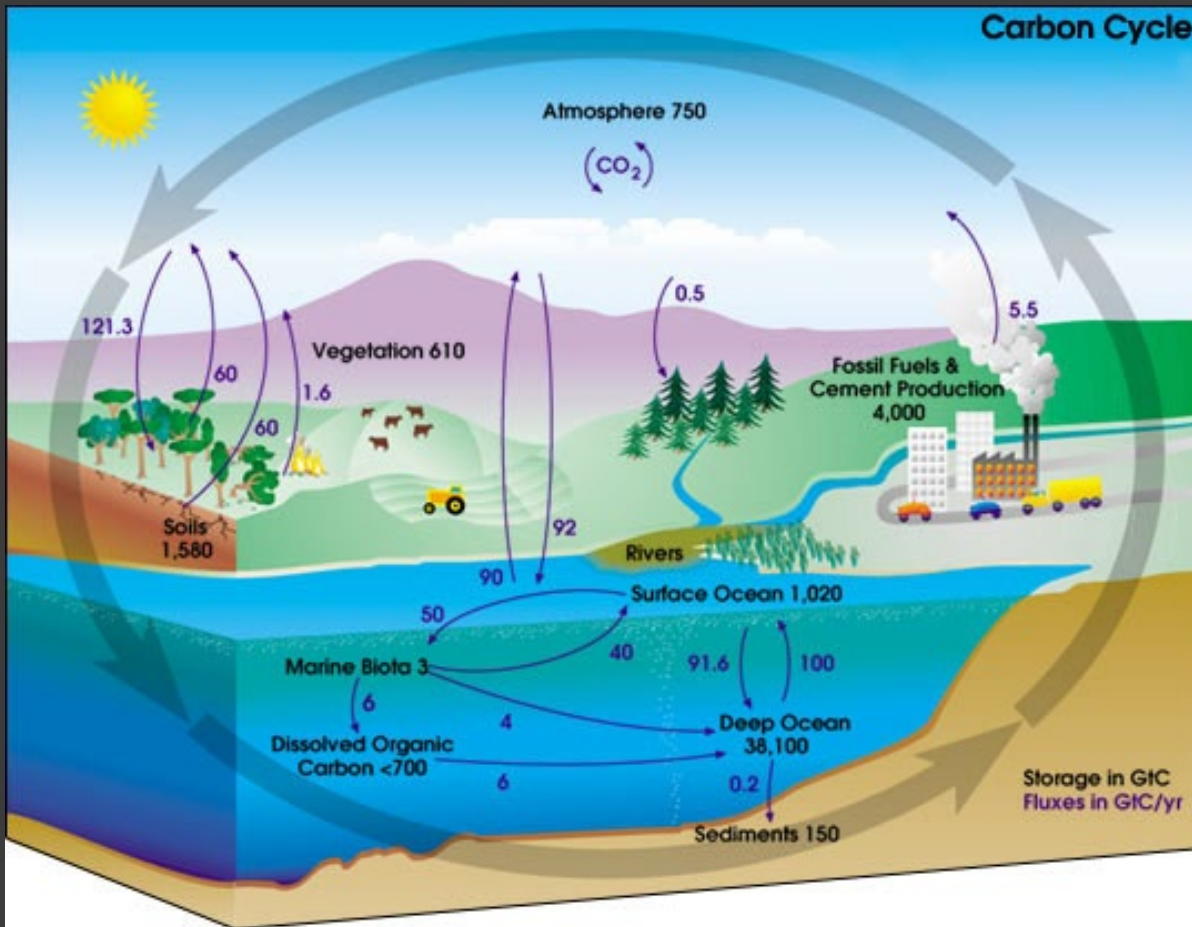
**Prof. Dr. Guilherme F. B. Lenz e Silva, BSc, CSP, MBA, Ph.D, SMAIChE**  
**Dept. Metallurgy & Materials Engineering / University of São Paulo – Brazil**

**2024**



# Everyday presence ...

## The good, ...



- Photosynthesis
- Earth natural temperature control and maintenance (live planet !)



... the bad and the ugly !



<https://bangkok.unesco.org/content/unesco-bangkok-marks-beat-plastic-pollution-mission-sense-urgency>

**Plastic and tyres pollution !**

# Carbon: origins and technological evolution

The word “carbon” came from Latin word: carbo that means: **charcoal**



15,000 BC – North of Spain



First black colour paint !

# Carbon: origins and technological evolution

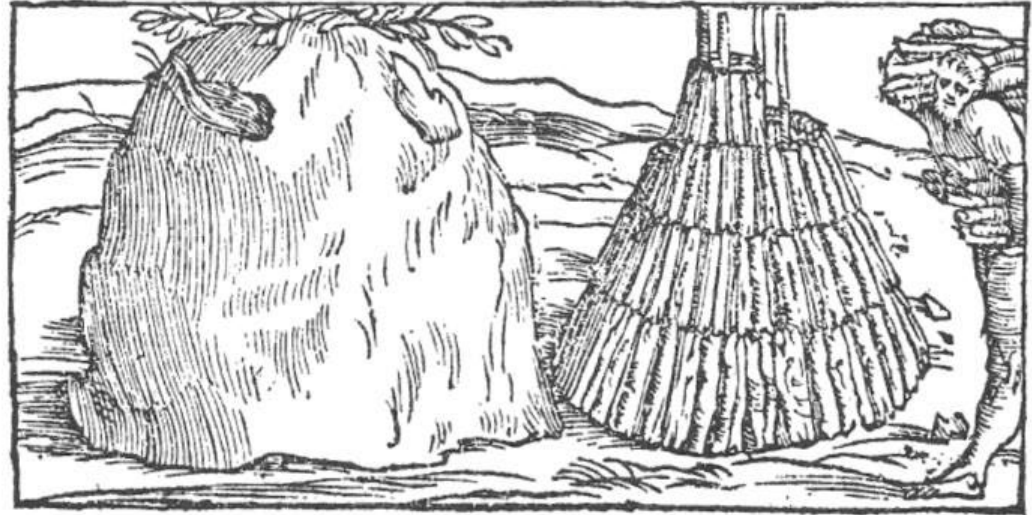
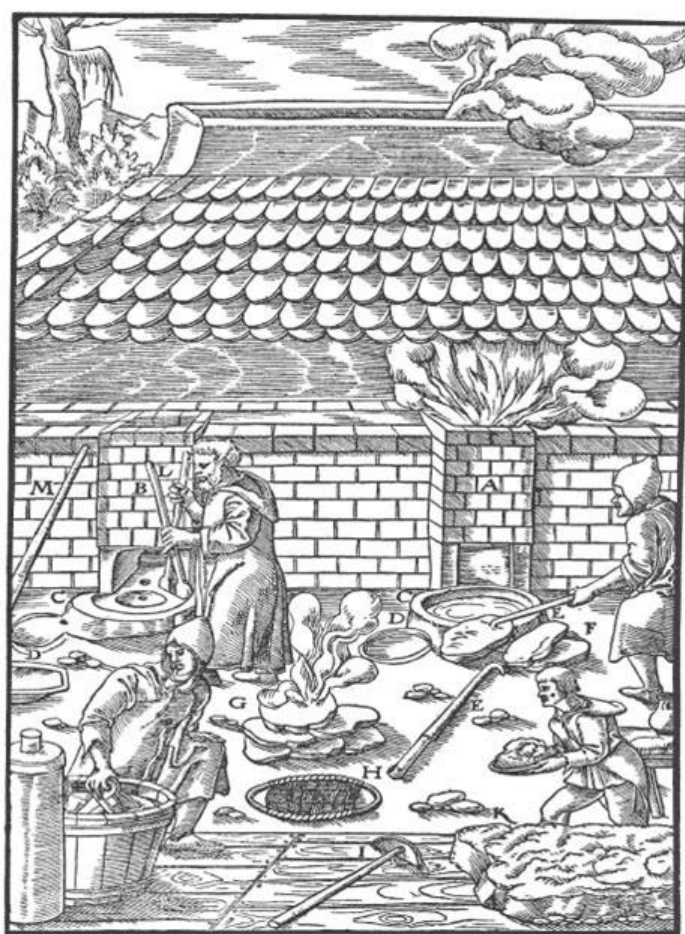
Essential for iron-carbon  
Development and Technology

The Iron Pillar of Delhi  
(3<sup>rd</sup> ~4<sup>th</sup> Century)



# Charcoal and the Middle age

The word “carbon” came from Latin word: carbo that means charcoal



**Middle age: charcoal production for iron making**

Source: The Pyrotechnia, VANNOCCIO BIRINGUCCIO; Year: 1540 edition

# Carbon: and the pre-industrial revolution



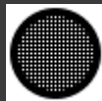
**The first blast furnace on coke**

**Abraham Darby I,  
10 of January of 1709**



# Chemistry evolution ... The first two allotropes: graphite and diamond

*John Dalton. 1808; (Symbol of Carbon)  
New System of Chemical Philosophy.*



**Antoine-Laurent Lavoisier,**  
*named the carbon element (C)*



**Diamond: 1,200 BC (from  
Hindu manuscripts).**



Source: <http://www.mineralminers.com/html/diamins.stm>

# MARVEL

ADAMANTIUM ! ...

... it's came from ADAMAS

The origin of diamond word came from the Greek word: “Adamas” – that means: **invincible !**



## WOLVERINE

Born with super-human senses and the power to heal from almost any wound, Wolverine was captured by a secret Canadian organization and given an unbreakable skeleton and claws. Treated like an animal, it took years for him to control himself. Now, he's a premiere member of both the X-Men and the Avengers. [less](#)

Real Name	James Howlett
Height	5'3"
Weight	(Without Adamantium skeleton) 195 lbs., (with Adamantium skeleton) 300 lbs.

<http://marvel.com/characters/66/wolverine>

<http://www.slashfilm.com/new-logan-photo-effectively-shows-why-the-wolverine-sequel-will-be-r-rated/>

## Emma Frost



January Jones



X-man:  
First Class

Tahyna Tozzi

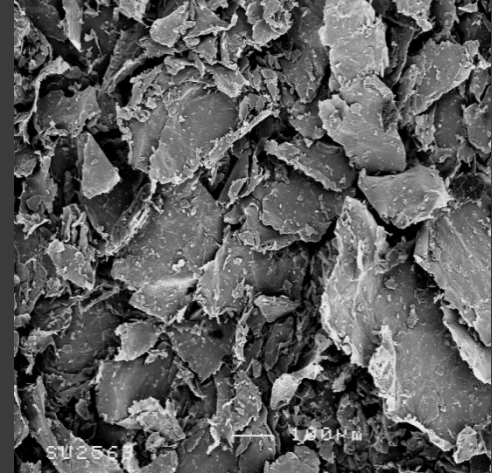


X-man:  
Origins (2009)

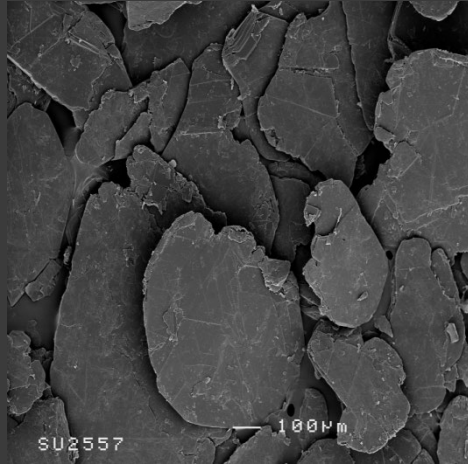


# Carbon materials, from the SEM point of view

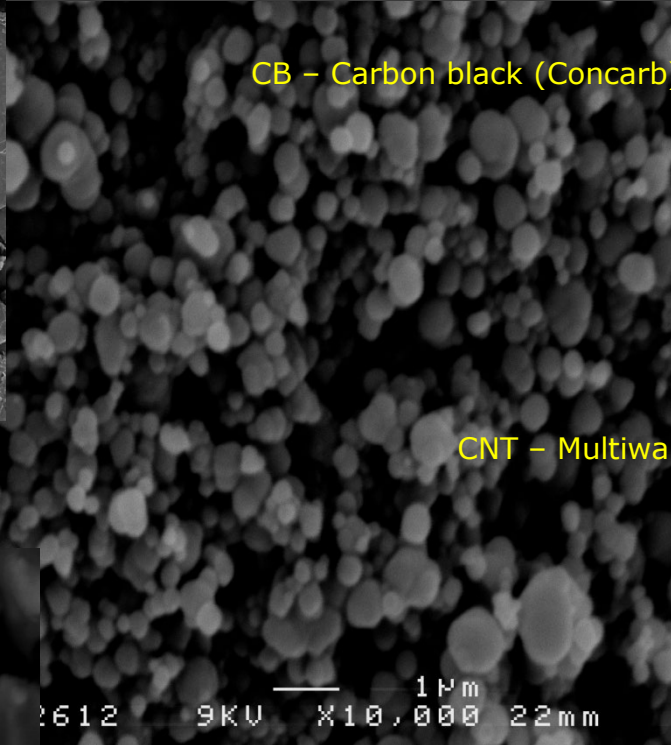
TG – Treated graphite (BSP 80)



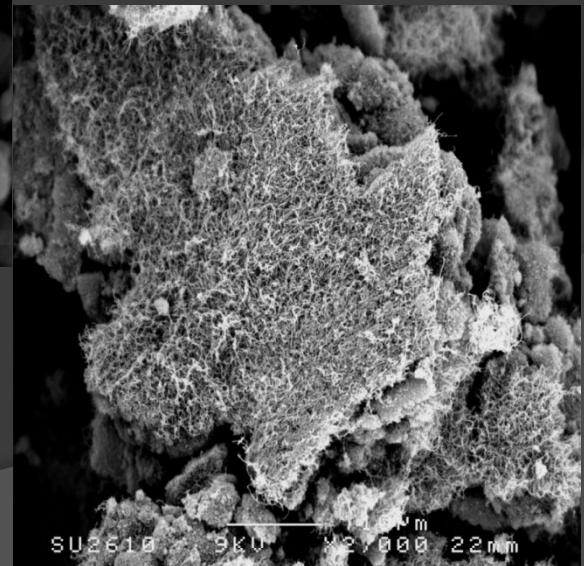
FG - Brazilian flake graphite ( N. Grafite)



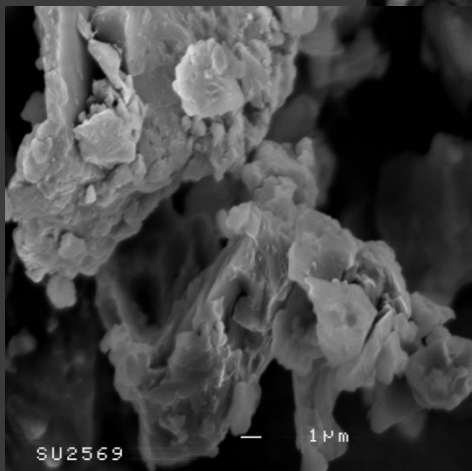
CB – Carbon black (Concarb)



CNT – Multiwalled carbon nanotubes (Sigma – Aldrich)

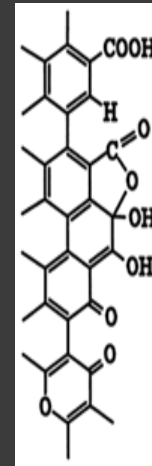
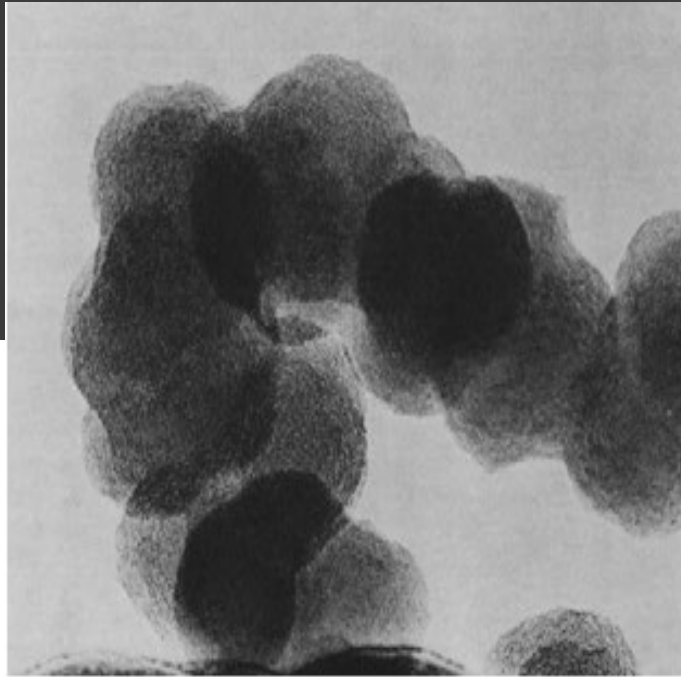


SP – Solid pitch  
(Carbores-P / Rutgers)



Prof. Guilherme Lenz images SEM –  
JEOL 6400 / Sorby Centre –  
University of Sheffield

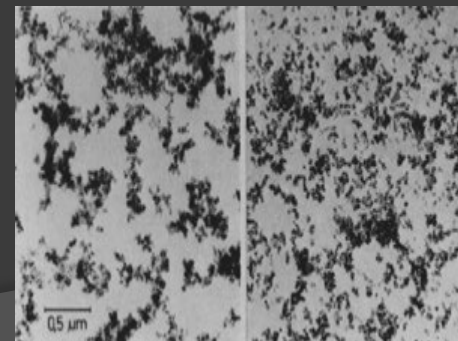
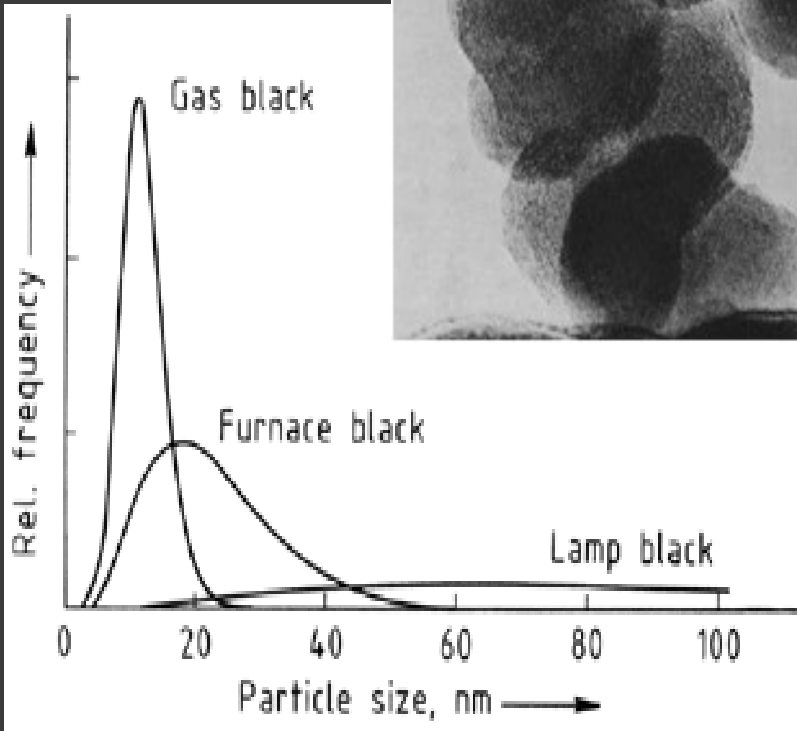
# Carbon blacks



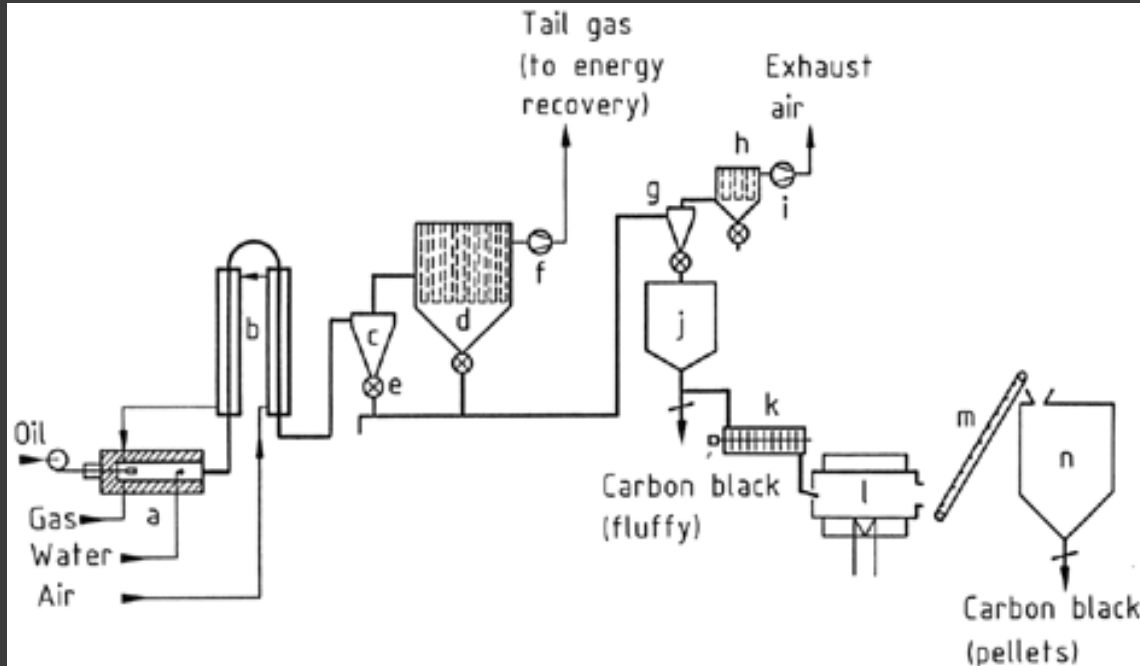
## Applications:

- Pigments
- Structural reinforcement of rubber
- Refractories
- Paints (Black pigment and electrostatic /conductive) and coatings

...

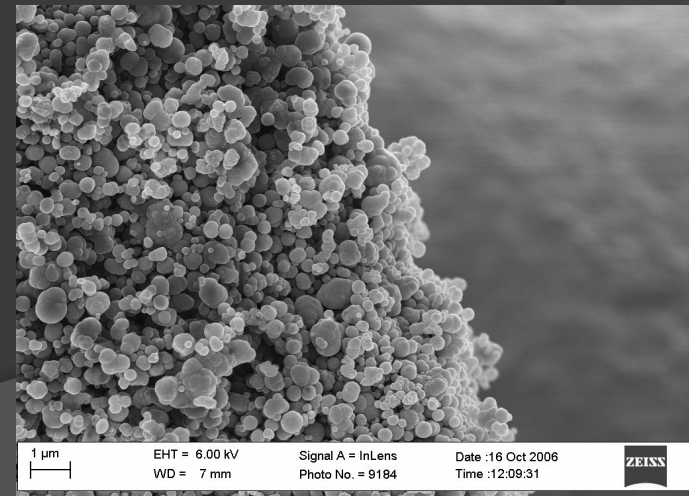


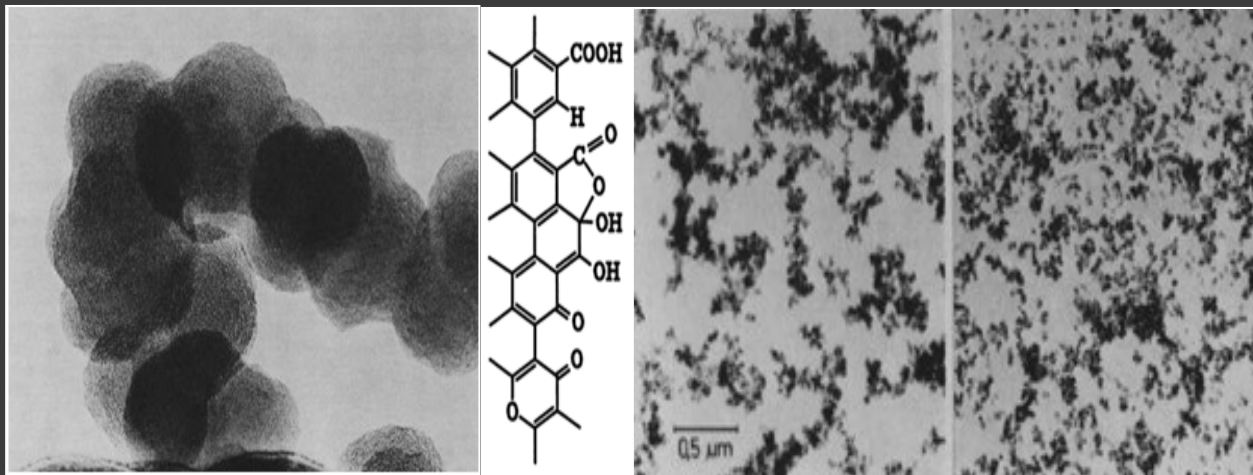
# Carbon black production: from oil and/or gas incomplete combustion



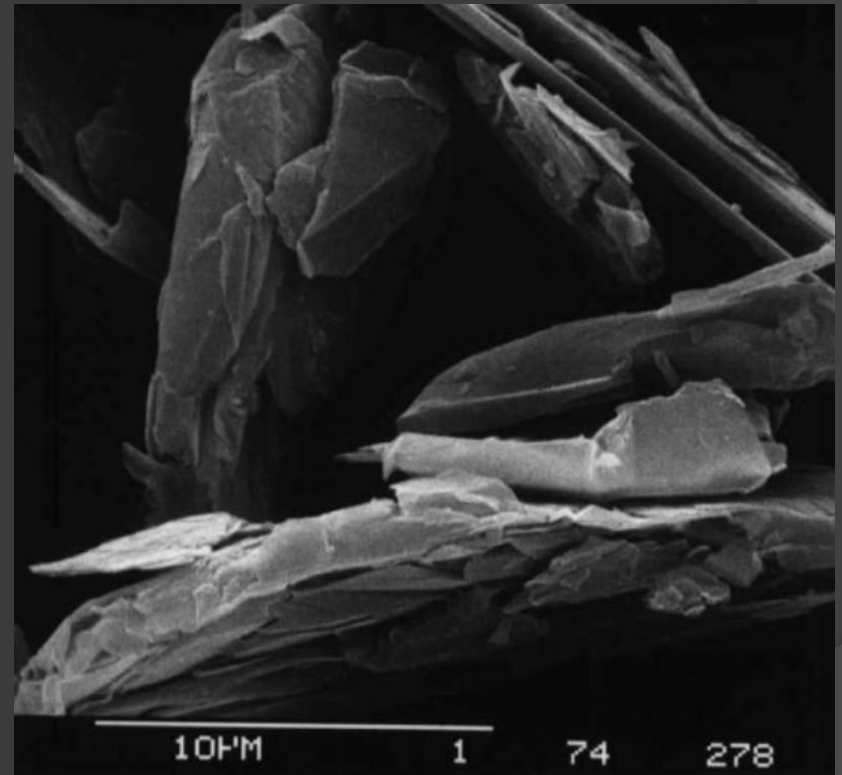
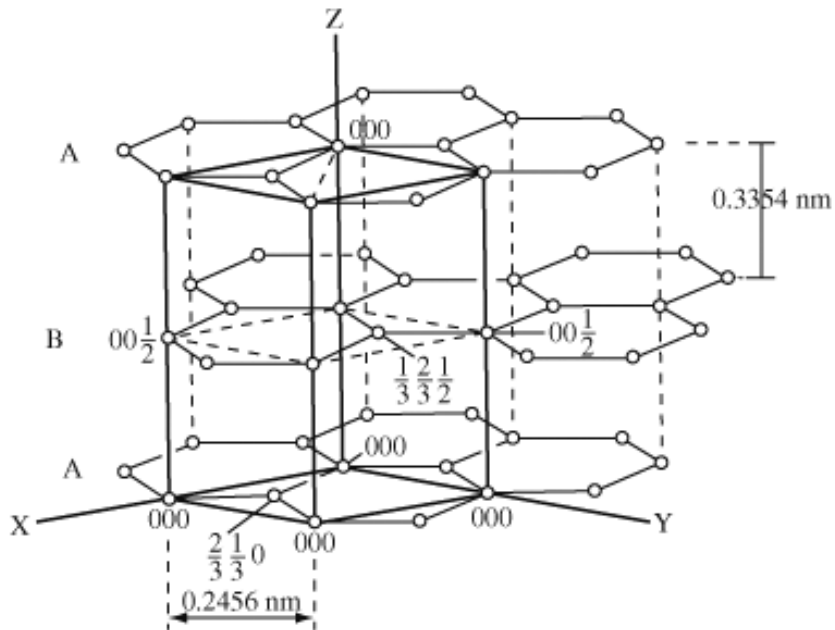
70% of world total production goes to tyres production

Up to 16 different types of carbon black on a tyre formulations.





# Natural Flake Graphite



Qingdao Kropfmuehl Graphite Co., Ltd



Application	Type	C content	Flake size	Comments
Refractories				
Magnesia graphite	F	85 - 90%	150 - 710 µm	Ash <2%, often up to 10%.
Alumina graphite	F	Min. 85%	150 - 500 µm	
Crucibles	AF	80 - 90%	+150 µm	Bulk density 48 - 54 g per 100 cm <sup>3</sup>
Expanded graphite	F	Min. 90%	200 - 1700 µm	
Foundry additive	AFV	40 - 70%	53 - 75 µm	Sulphides and other readily fusible minerals undesirable
Foundry core and mould washes	AF	70 - 90%	~75 µm	
Brake / clutch linings	AFV	Min. 98%	<75 µm	
Bearings	FV	90 - 93%	+150 µm	
Lubricants	AFV	98 - 99%	53 - 106 µm	Free from sulphides, abrasive material and metallic contaminants.
Dry cell batteries	A	Min. 88%	85% <75 µm	No metallic impurities and S less than 0.5%
Alkaline batteries	AF	Min. 98%	5 - 75 µm	No impurities such as Cu, Co, Sb and As.
Recarburizing steel	A	98 - 99%	~5 µm	
Carbon brushes	AFV	95 - 99%	<53 µm	< 1% ash / silica. No abrasive or metallic contaminants.
Electrical	FV	93 - 95%	+150 µm	
Pencils	AF	95 - 97%	+150 µm	Free of gritty impurities
Conductive coatings	A	50 - 55%	<75 µm	May contain up to 25% silica, but free from gritty impurities.
Packing paints	FV	85 - 90%	<150 µm	
Polishes	AFV	85 - 90%	<150 µm	
Drilling mud (lubricating)	F	80%+	N/A	About 4lb per barrel of mud
Explosives (control of burning rate)	AF	65%	<150 µm	Free from sulphides and acids. Low moisture content
Nuclear reactors (moderators & reflectors)	F	93 - 95%	N/A	Free from high neutron absorbing elements, e.g. boron
Boilers (scale prevention)	F	50%+	N/A	



British Geological Survey



TECHNICAL REPORT WG/92/30  
Mineralogy and Petrology Series

Industrial Minerals Laboratory Manual  
**FLAKE GRAPHITE**

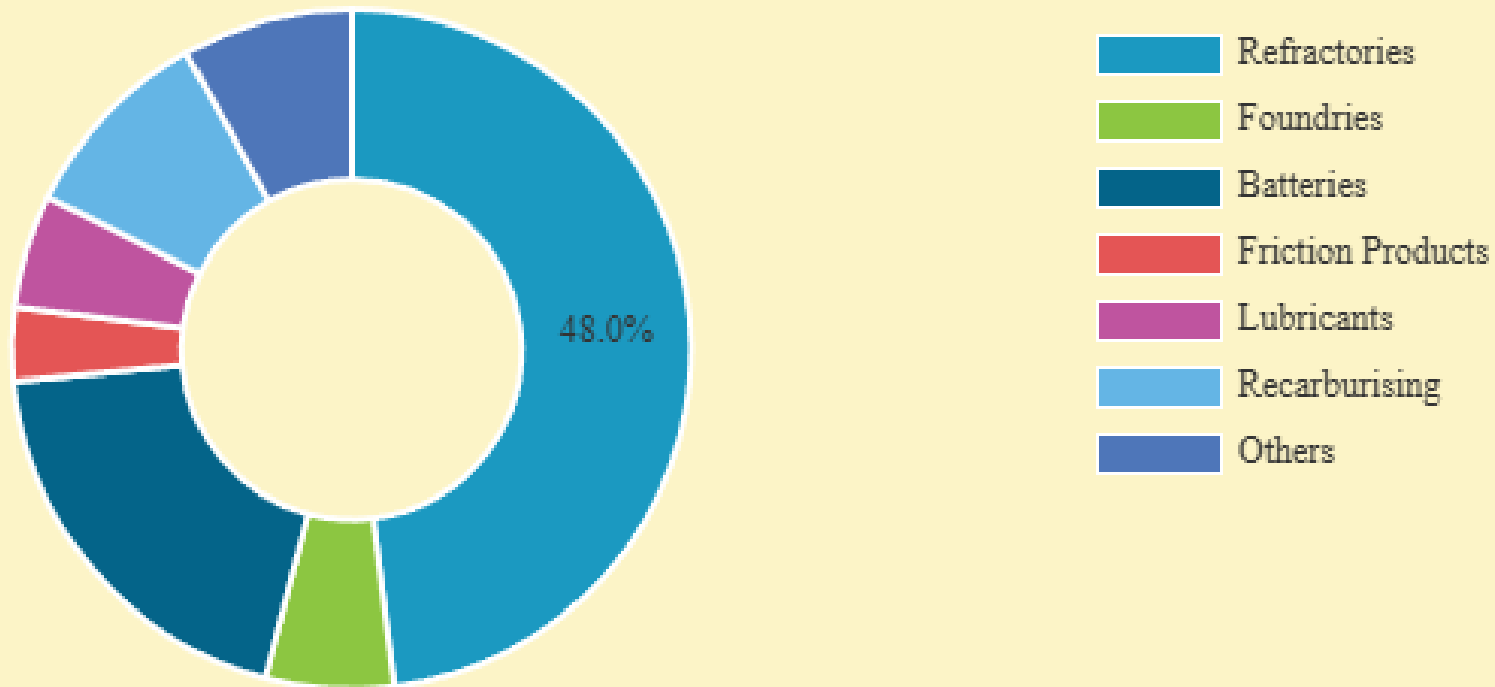
C J Mitchell



	Flake	Vein	Amorphous
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<b>Description</b>	Crystalline flakes; coarse >150 µm; fine <150 µm	Coarse crystals Mostly >4 cm	Microcrystalline <70 µm
<b>Origin</b>	Syngenetic; regional metamorphism	Epigenetic; regional metamorphism	Syngenetic; contact and / or regional metamorphism
<b>Ore</b>	5 - 30% graphite; strata-bound, tabular or lenses	98%+ graphite; veins and fractures	Seams, often folded and faulted
<b>Product grade</b>	75 - 97% graphite	98 - 99.9% graphite	60 - 90% graphite
<b>Main uses</b>	Refractories, brake linings, lubricants and batteries	Carbon brushes, brake linings and lubricants	Refractories, steel industry, paint and batteries
<b>Major producers</b>	China, Brazil, India, Madagascar, Germany, Norway, Canada and Zimbabwe	Sri Lanka.	China, S Korea, Czechoslovakia, Austria and N Korea

## Global Graphite Market Share, By Application, 2023



[www.fortunebusinessinsights.com](http://www.fortunebusinessinsights.com)

# GRAPHITE MARKET



**\$7.32 Billion** 2023      **\$7.80 Billion** 2024      **\$13.35 Billion** 2032

**CAGR 6.9%**  
2024-2032

## DRIVERS

Increasing Demand from the Refractory Industry

## TRENDS

Trend of Battery Powered Vehicles



### BY APPLICATION, 2023

Refractories 48.0%

Foundries | Batteries | Friction Products | Lubricants  
Recarburising | Others

### BY PRODUCT

Synthetic  
Natural



## CHINA, BY APPLICATION, 2023

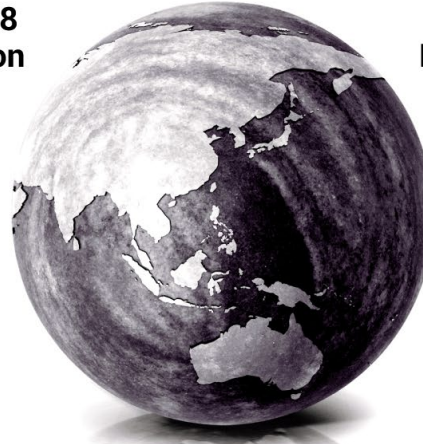
Refractories 48.3%

Foundries | Batteries | Friction Products | Lubricants  
Recarburising | Others

# ASIA PACIFIC

**\$3.78 Billion**  
2022

**\$4.08 Billion**  
2023



North America | Europe | Latin America | Middle East & Africa

## KEY PLAYERS



## INDUSTRY DEVELOPMENT

MARCH 2022

EagleGraphite announced its partnership with British Columbia to develop silicon-modified battery anodes.



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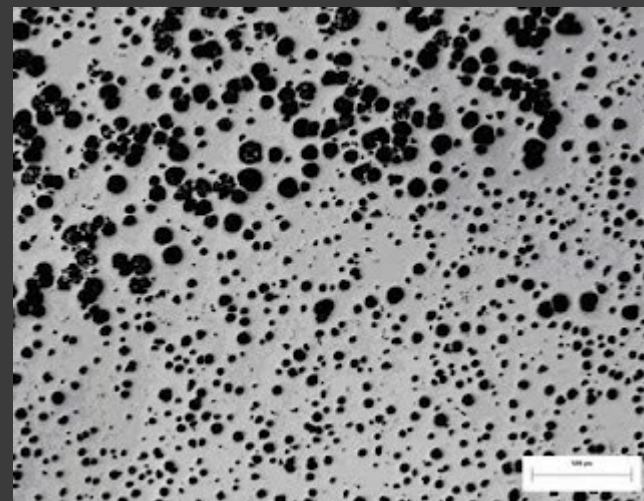
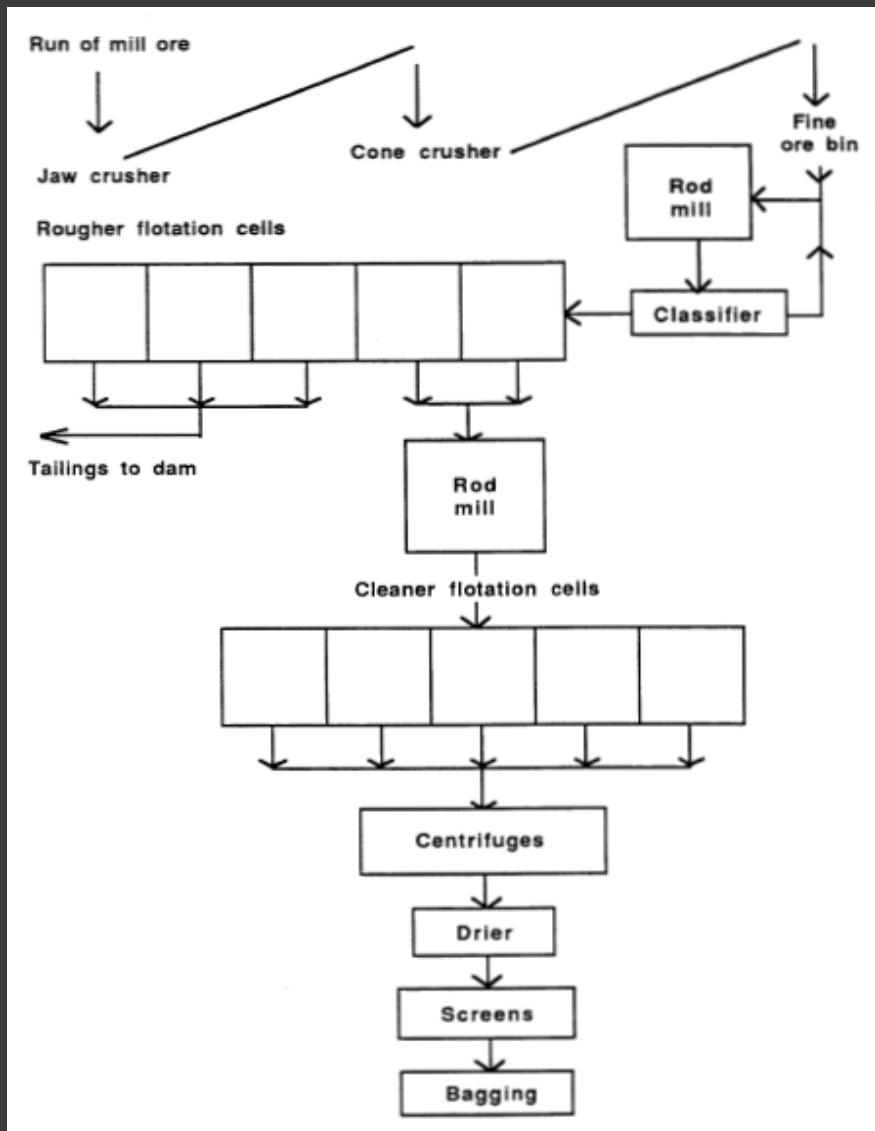
**TLMC<sup>5</sup>**  
Quinto Taller Latinoamericano  
de Materiales de Carbono



### LIST OF KEY COMPANIES PROFILED:

- » AMG (Germany)
- » [Asbury Carbons](#) (U.S.)
- » Eagle Graphite (Canada)
- » [Grafitbergbau Kaisersberg GmbH](#) (Austria)
- » [Imerys S.A.](#) (France)
- » Stoker Concast Pvt. Ltd. (India)
- » BTR NEW Material Group Co., Ltd. (China)
- ★ [Nacional de Grafite](#) (Brazil)
- » [SGL Carbon](#) (Germany)
- » Mineral Commodities Ltd. (Australia)
- » Superior Graphite (U.S.)
- » Tirupati Carbons & Chemicals Pvt. Ltd. (India)

# Natural flake graphite processing flowchart (Mine Lynx – Zimbabwe)

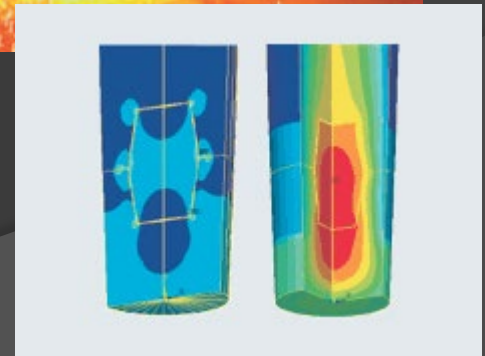
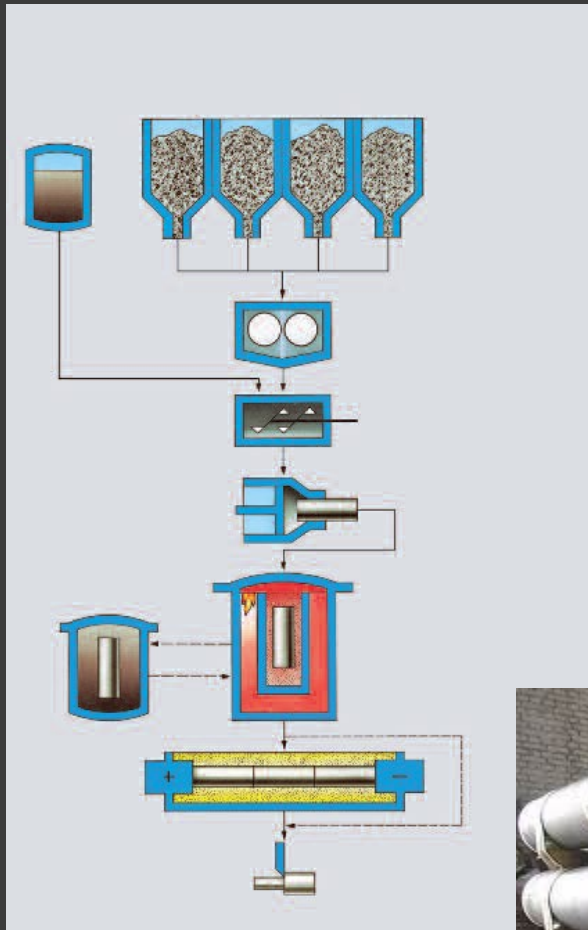


Flotation of natural graphite (Madagascar)



<http://ets.gallois.pagesperso-orange.fr/index.htm>

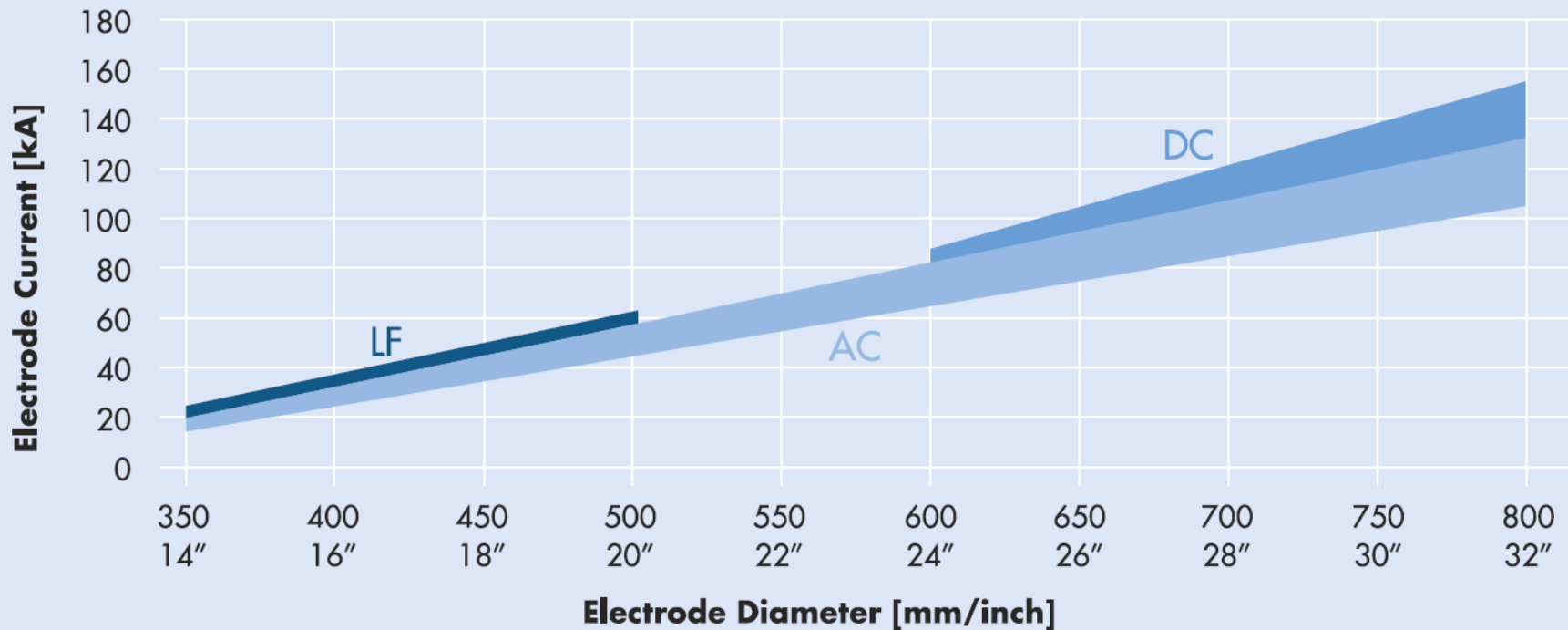
# Electrodes for metal reduction and melting/scrap furnaces



# Electrodes for different size equipment

## Electrical current vs. electrode diameter

### Current-Carrying Capacity



# Carbon cathodes for metallic aluminum production from alumina reduction



**Carbon blocks:** calcined anthracite treated at 1,800 °C

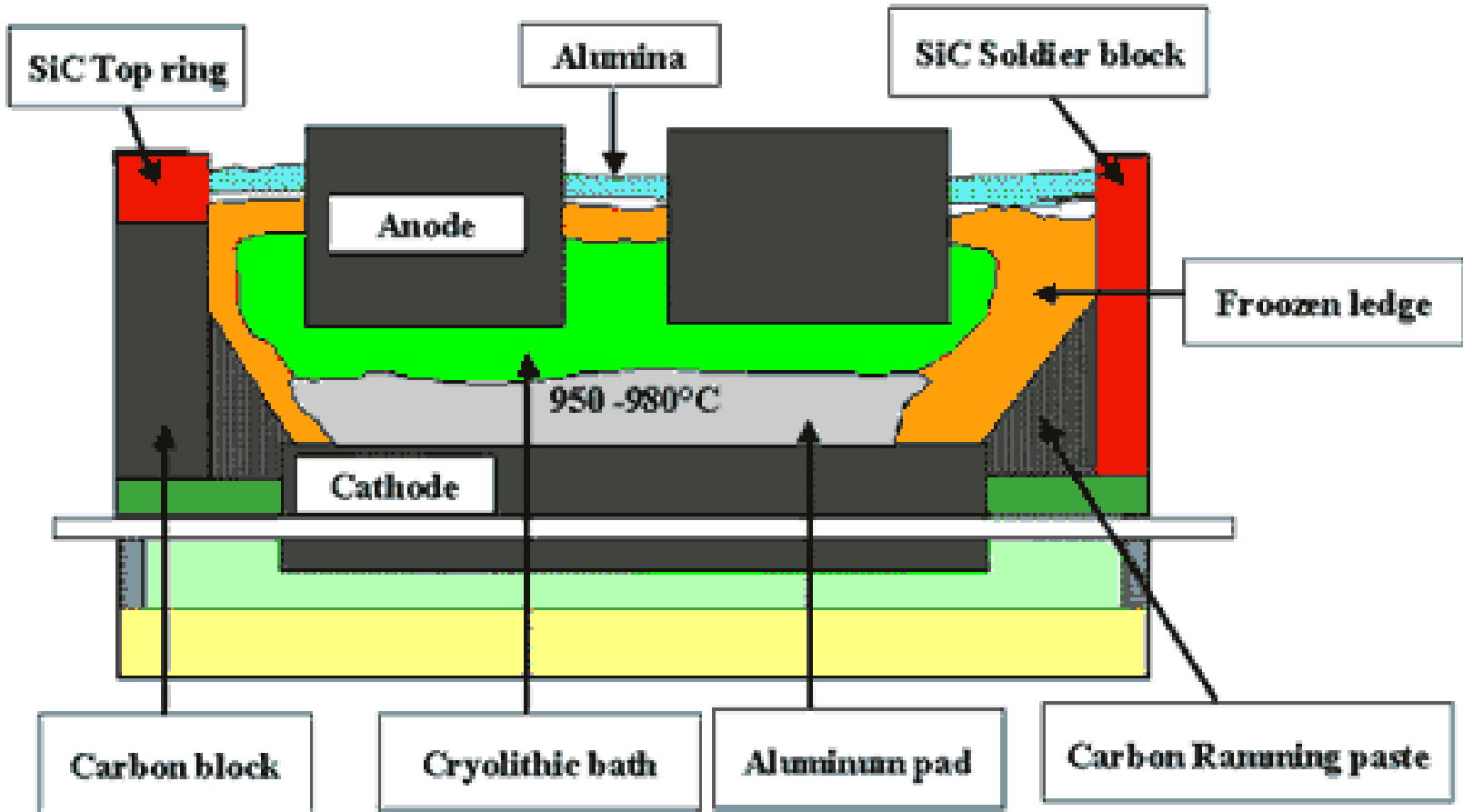
**Graphite blocks:** artificial graphite from oil/gas

**Carbon blocks:** from petro-coke .

No	Item	Unit	Partial Graphite	High Graphite	Graphite
1	Ash	≤ %	8	5	3
2	Resistance	≤ μΩ·m	43	35	32
3	Bulk Density	≈ g/cm <sup>3</sup>	1.54	1.52	1.52
4	Electrolytic Expansion	≤ %	1.0	—	—
5	Real Density	≈ g/cm <sup>3</sup>	1.88	1.90	2.06
6	Compressive Strength	≈ Mpa	30	28	26

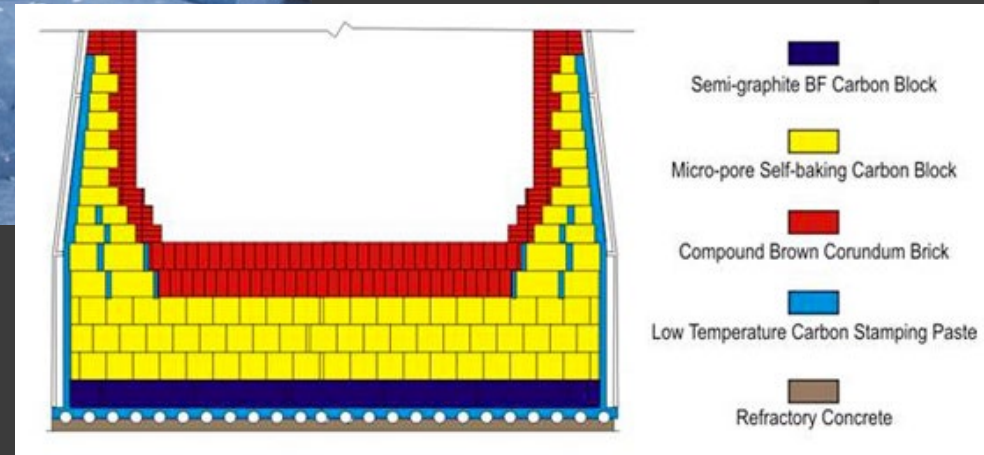
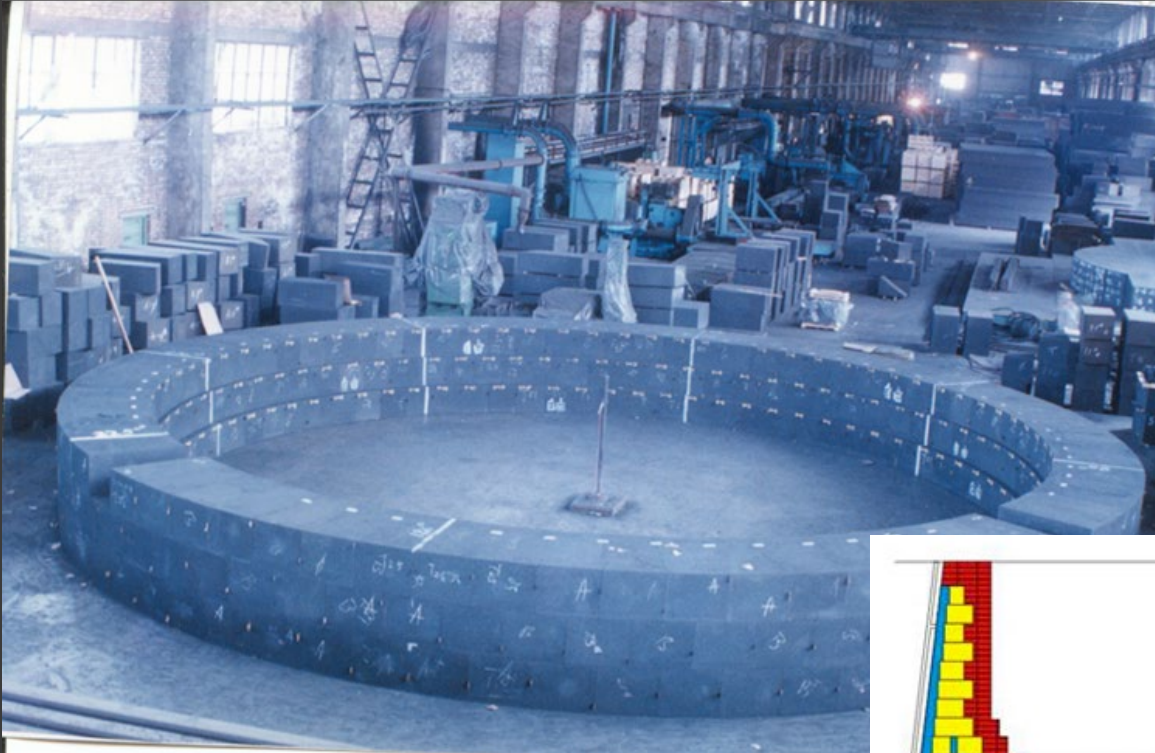


# Hall-Heroult alumina reduction cell



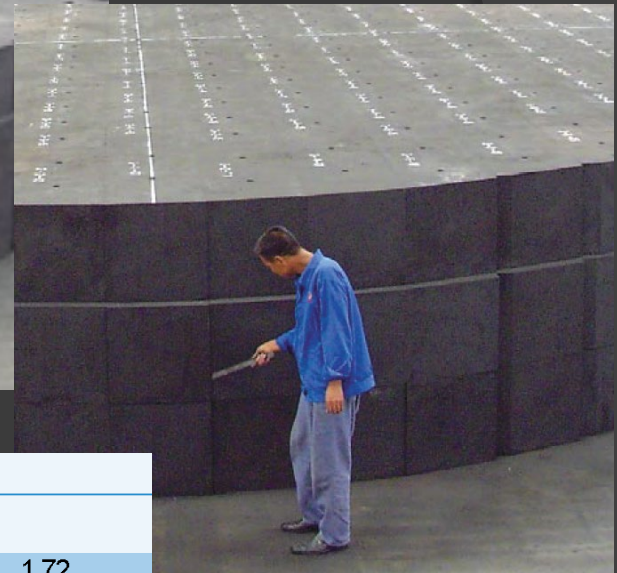
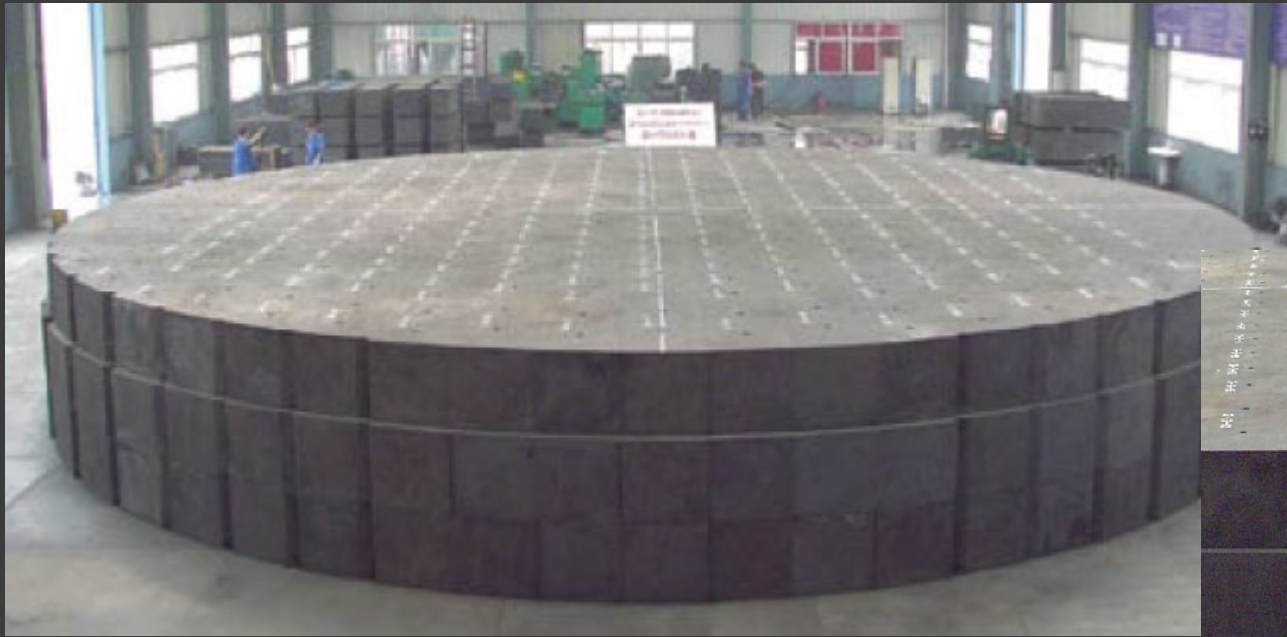
Source: <http://www.refractories.saint-gobain.com/Aluminium-Reduction.aspx>

# Carbon block – blast furnace for hot metal production



## Super Micro-pore Carbon Block

Fonte: <http://www.fangdacarbon.com/en/product/c/default.shtml>



Bulk density	g/cm <sup>3</sup>	1.61	1.65	1.72
Apparent porosity	%	24.70	19.10	16.5
Compression strength	MPa	25.70	30.25	35.80
Bending strength	MPa	7.80	10.51	12.50
Thermal conductivity (room temp.)	W/m-k	42	84	128
Ash	%	0.48	0.39	0.28

[http://www.furnacelining.com/FangYuan/Literature\\_files/FangYuan\\_Blast\\_Furnace.pdf](http://www.furnacelining.com/FangYuan/Literature_files/FangYuan_Blast_Furnace.pdf)

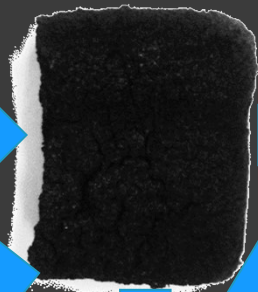
# Sustainable Porous Carbons Materials



**Kraft Black liquor**

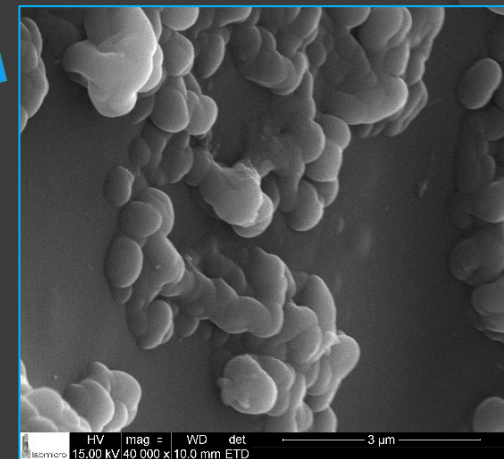
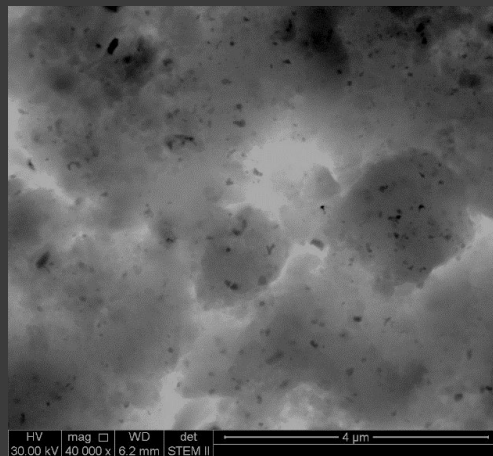
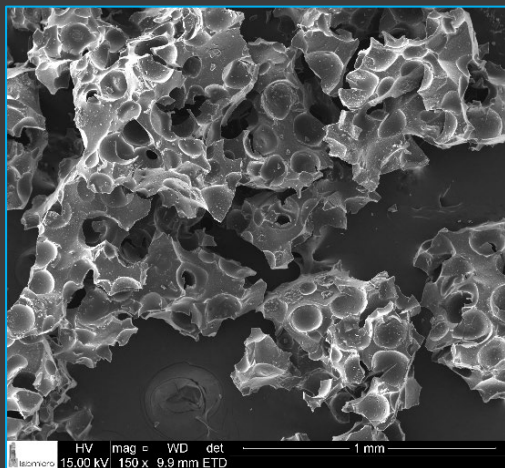
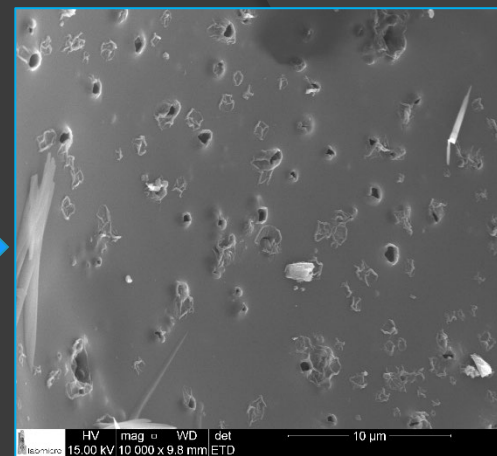
Crosslinker  
PMMA  
Resorcinol

**Polymerization/  
Carbonization**



$\text{Ni}(\text{NO}_3)_2 \cdot 5\text{H}_2\text{O}$

**Nanoparticles**



**Bulk density  
(g/cm<sup>3</sup>)**

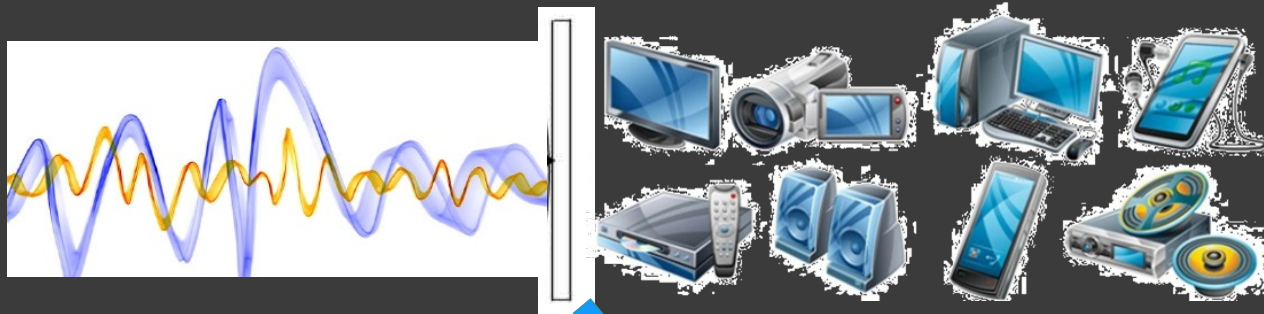
**Porosity  
(%)**

0.35-0.50

70-85

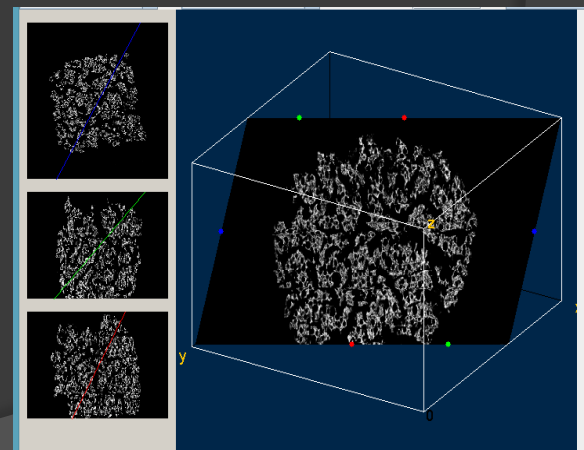
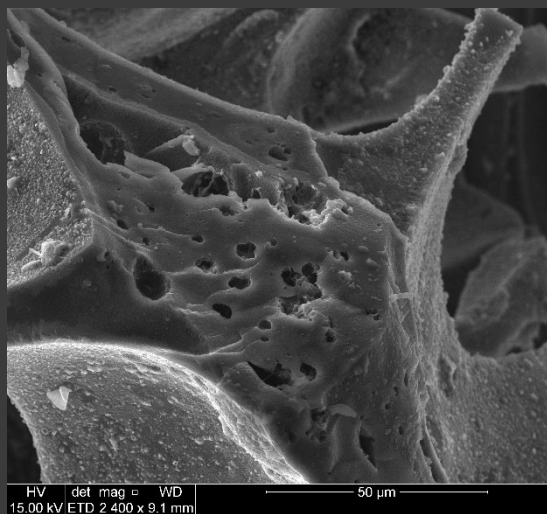
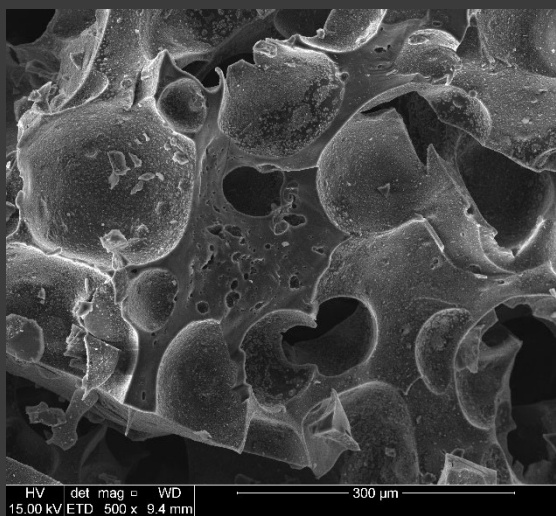
# Electromagnetic Shielding

Aerospace field- radar



- ✓ High porosity
- ✓ Pore size (micron scale)

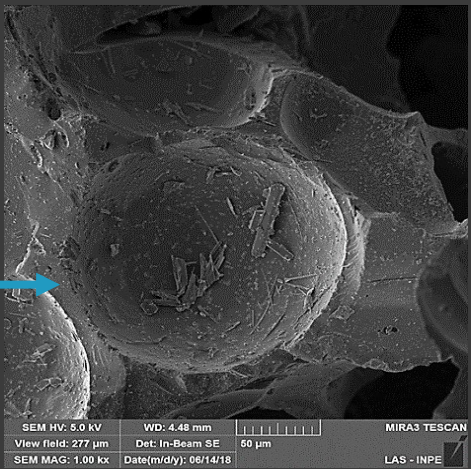
EMI- electromagnetic interference



# Shield by absorption

Vector network analyzer (VNA) in the frequency range from 8.2 GHz to 12.4 GHz (X-Band) for electromagnetic properties

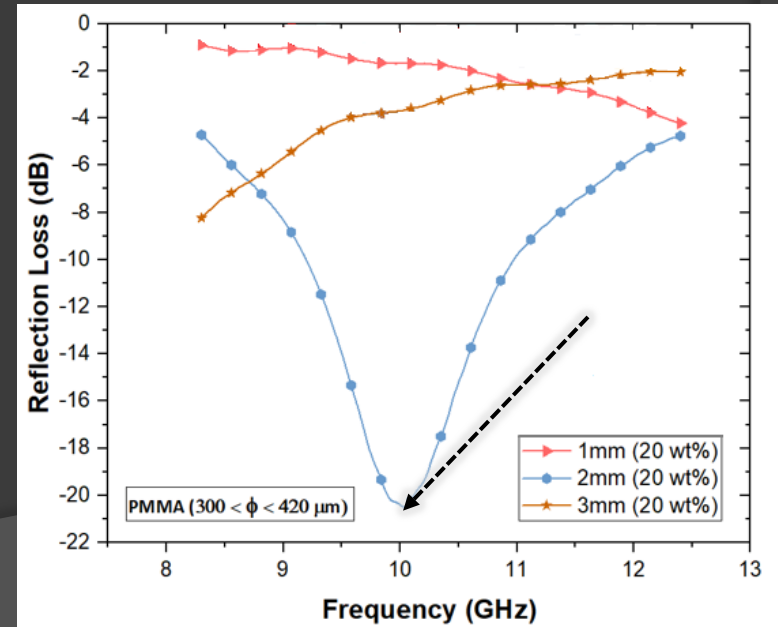
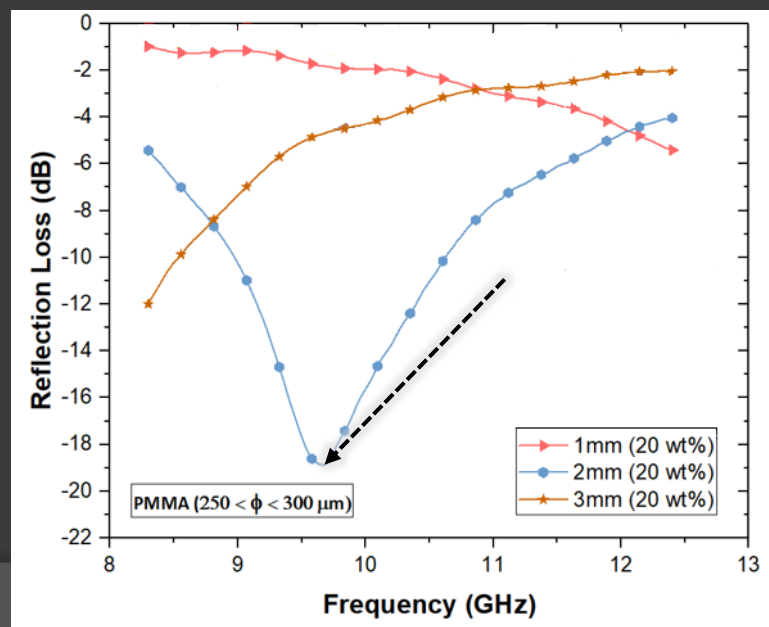
Dispersion in Silicon



Particle size 250-420  $\mu\text{m}$

$$RL(\text{dB}) = 10 \log_{10} \frac{P_i}{P_r}$$

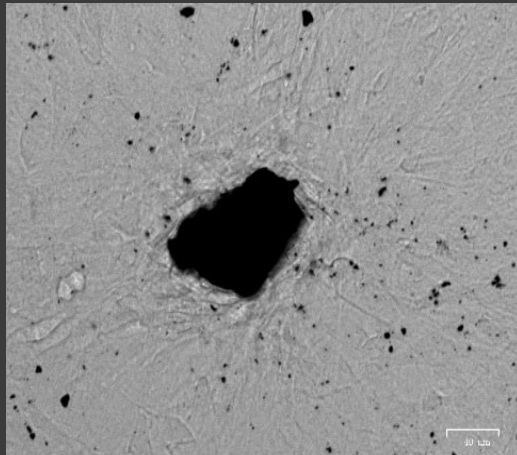
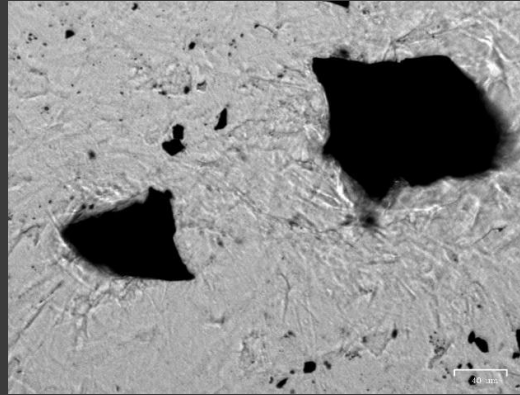
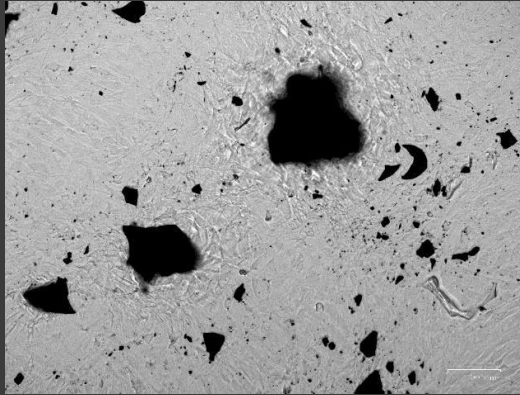
**99% of absorption**



# POROUS CARBON BIOMATERIAL (IN VITRO TESTS)

## Osteoblasts cells culture

### Biomaterial



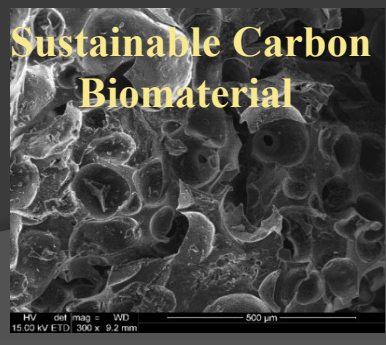
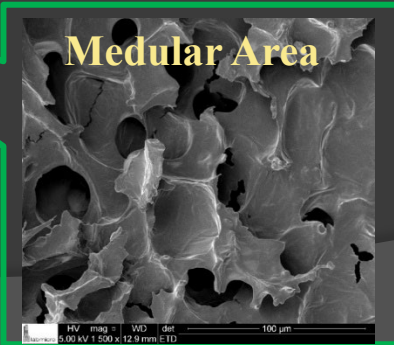
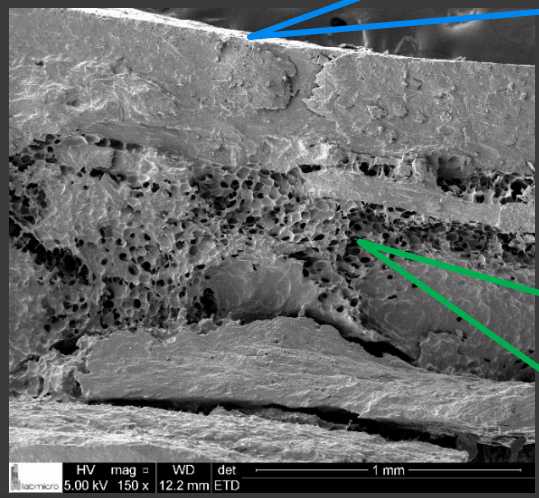
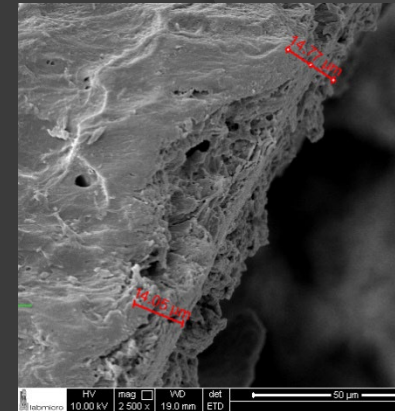
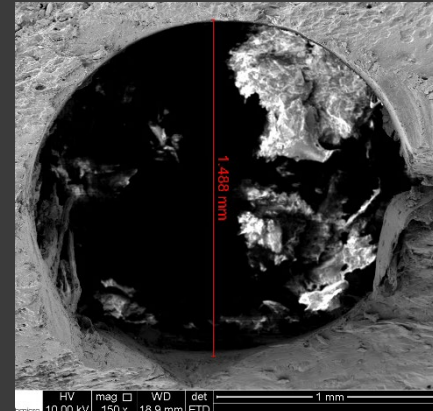
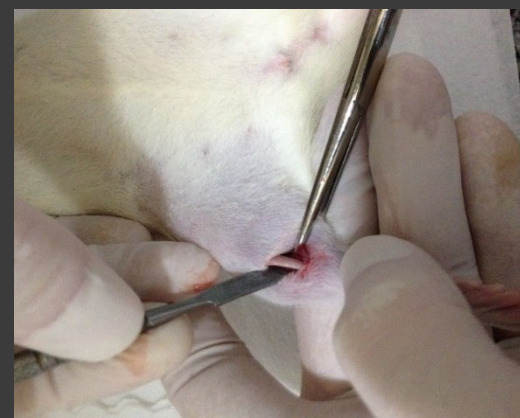
**Biomaterial + Silver  
nanoparticles (600 ppm)**

- Cell Adhesion

Optical Microscopy

# Bone X Biomaterial (structure)

## Tibia - Bone Defect (standard process)

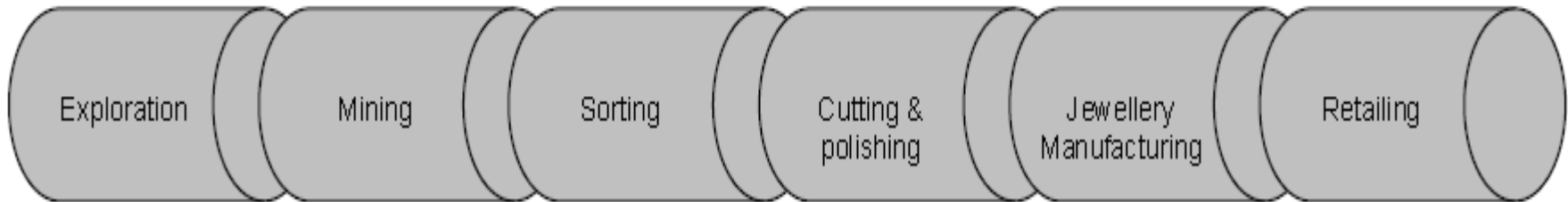


Dra. Patrícia Almeida  
(Posdoc at LM<sup>2</sup>C<sup>2</sup>-PMT POLI – USP)



# Synthetic and natural diamonds

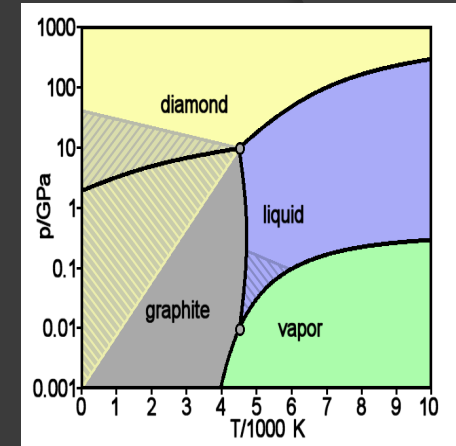
- Market size (estimated): US\$9 to 13 billions /year
- Jewelry: 30%
- Industrial grade: 70% (cut, drilling and polishing)



- |                 |            |                 |                     |
|-----------------|------------|-----------------|---------------------|
| - Angola;       |            |                 |                     |
| - Australia;    | - Color;   | - Antwerp;      | - USA (50%);        |
| - Botswana;     | - Size;    | - Mumbai;       | - Japan (15%);      |
| - Canada;       | - Quality; | - Tel Aviv;     | - Italy (5%);       |
| - Congo;        | - Shape.   | - New York;     | - India (3%);       |
| - Namibia;      |            | - China;        | - China (2%);       |
| - Russia;       |            | - Thailand;     | - Middle East (2%); |
| - South Africa. |            | - Johannesburg. | - Others (23%).     |

# Synthetic diamonds

2 main synthesis routes: The first one was discovery by GE at 1955 . The second is by CVD (Chemical Vapor Deposition). Nowadays 100 million of carats of industrial grade diamonds are produced by year.

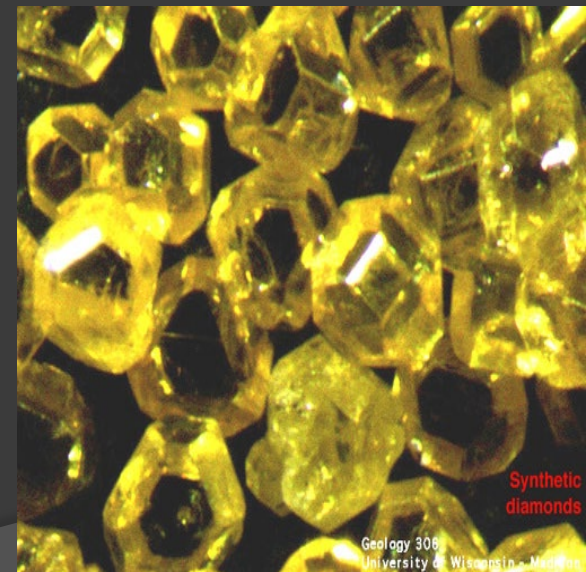


Carbon phase diagram



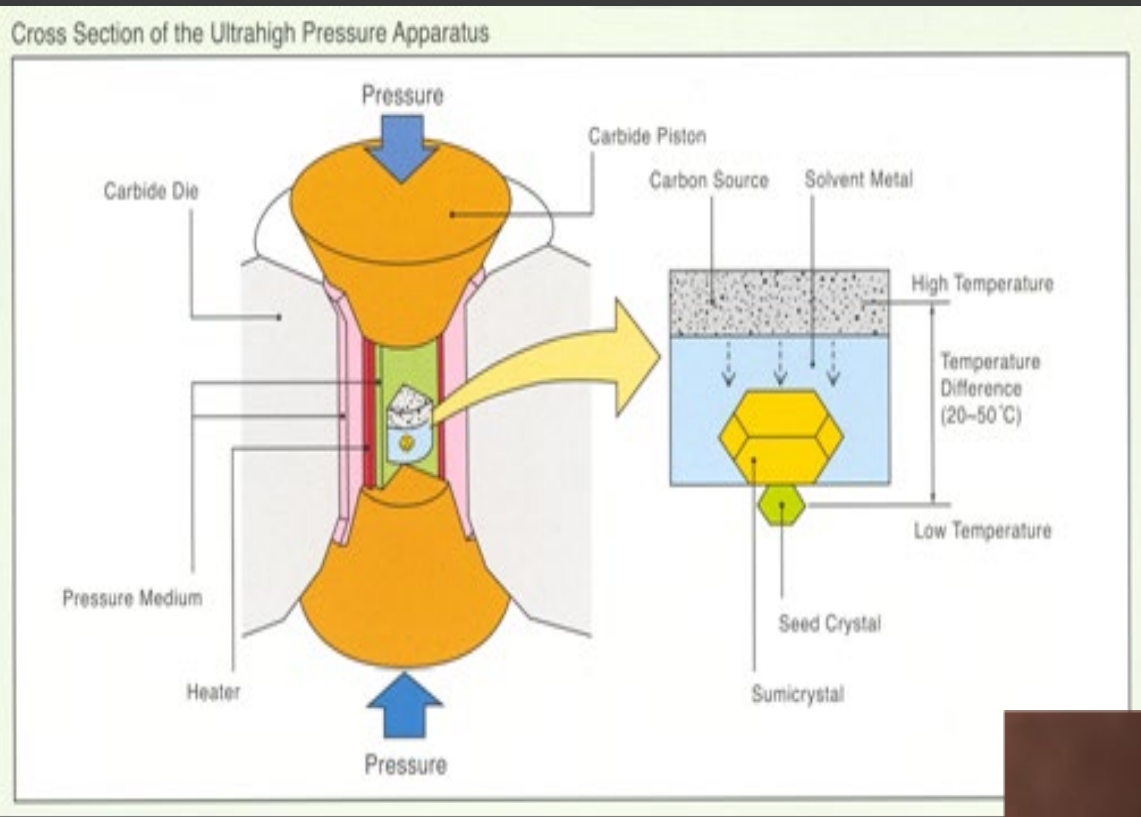
High Pressure High Temperature (HPHT)

Synthetic Diamond Hydraulic Press



Synthetic diamonds

Geology 306  
University of Wisconsin - Madison



Artificial diamond



# NOTHING LASTS FOREVER

**Nothing Lasts Forever**

2022 | A12 | 1h 27m | Documentary

Insiders cut through public perception to clarify the infiltration of synthetic gems into the diamond industry in this probing documentary.

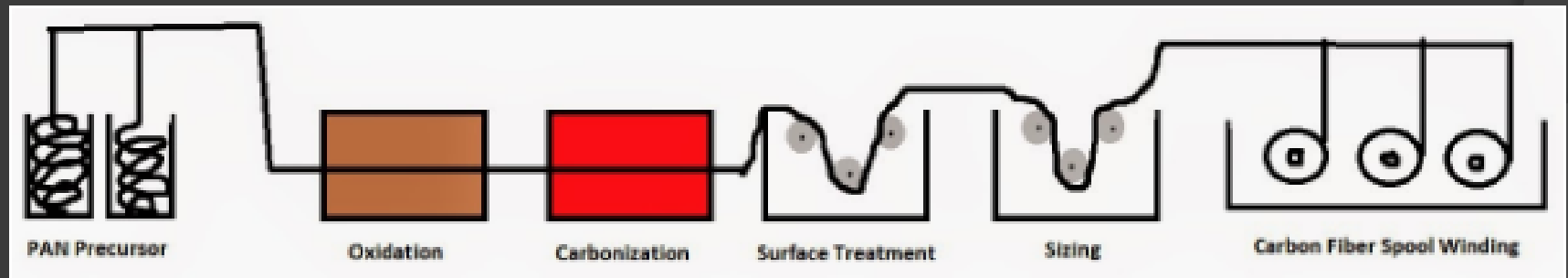
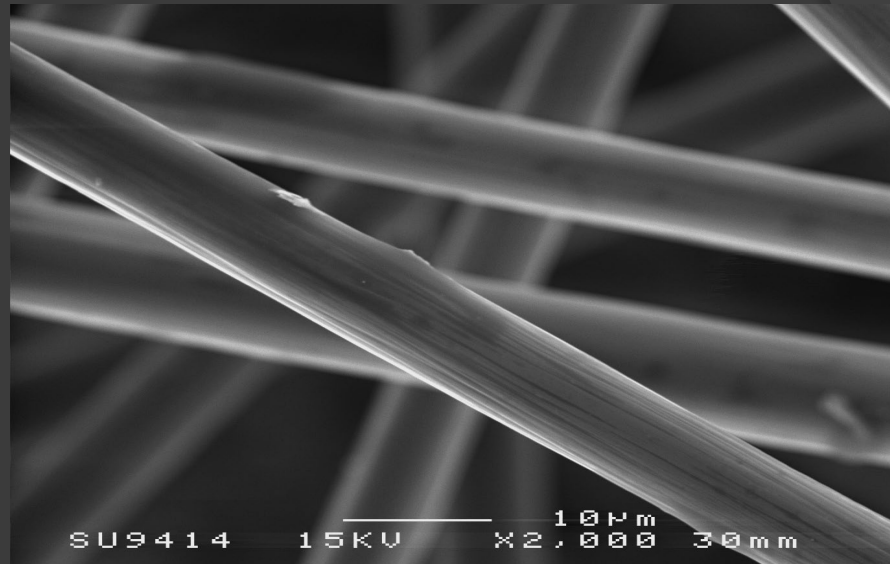
**NETFLIX**

# Carbon Fibers

Two dominant technologies: PAN and pitch.

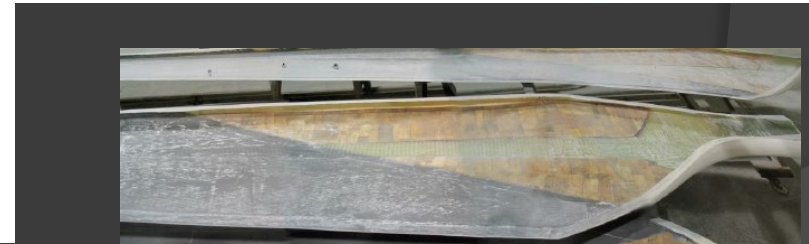
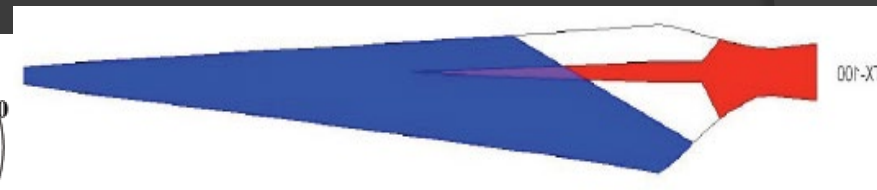
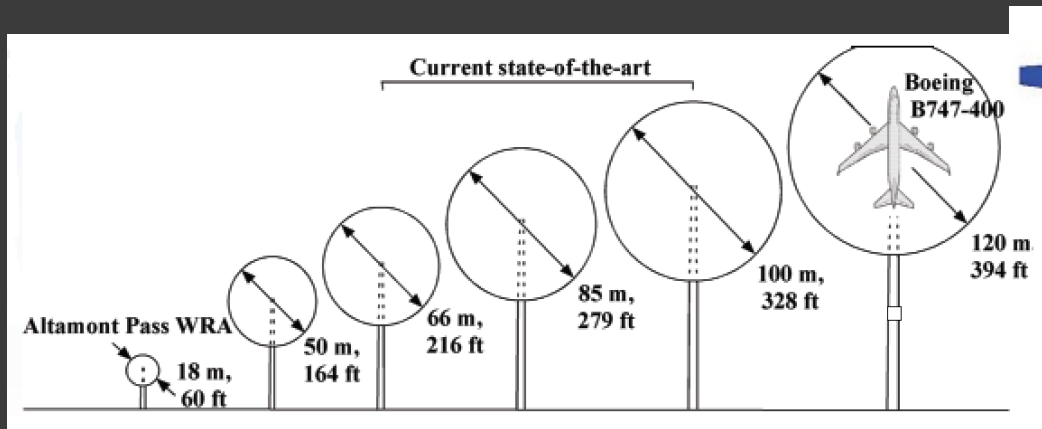
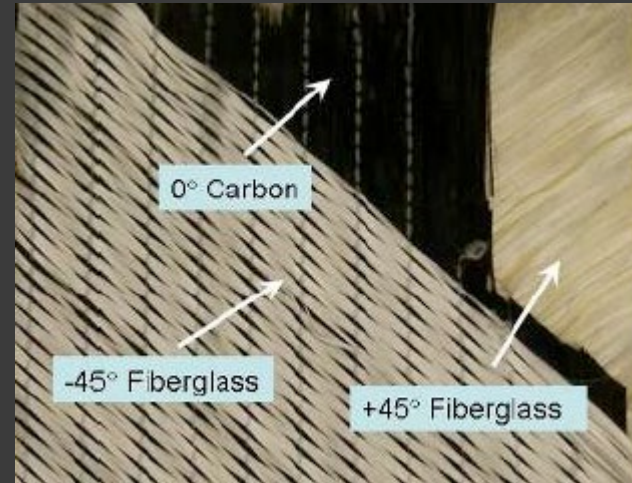
SEM Image SEI mode: 7 microns  
High resistance fiber from:  
Toray/France

MEV JEOL 6400-F



Typical carbon fiber production line from PAN precursor.

# Composites: carbon f& glass fibers



# Lamborghini – Sesto elemento

**Cost: 2,2 Millions US\$**  
(2<sup>nd</sup> more expensive car model in history !)



Boasting an extremely lightweight construction thanks to advanced carbon-fiber technology, the Lamborghini Sesto Elemento has an overall curb weight of just 999 kilograms (2,202 lb) - including V10 power unit and permanent all-wheel drive.

With its amazing output of 570 hp, sensational power-to-weight ratio of only 1.75 kilograms per hp and 0 to 100 km/h (0-62 mph) acceleration of only 2.5 seconds, the Sesto Elemento guarantees unparalleled driving fun.

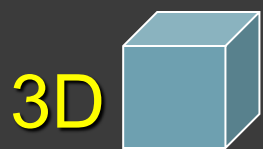
**Partnership: Lamborghini & Boeing**



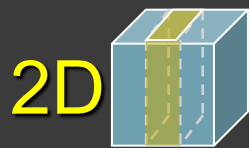
The carbon big “new” family:

... from micro to nanosize materials !

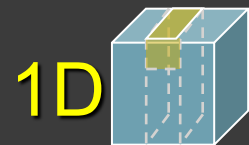
# What happens in the nano-size world?



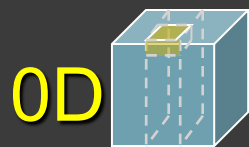
“Bulk”



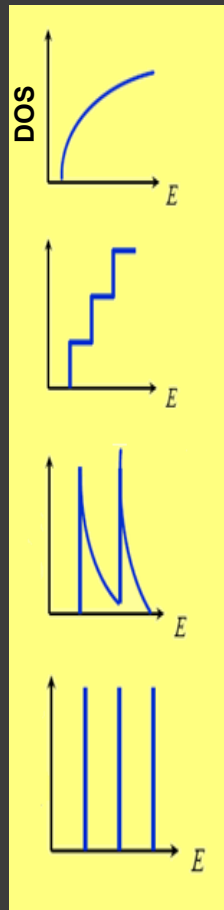
“Films”



CNT  
(SWCNT, DWCNT,  
MWCNT, etc.)



Nanoparticles  
Quantum dots  
etc.



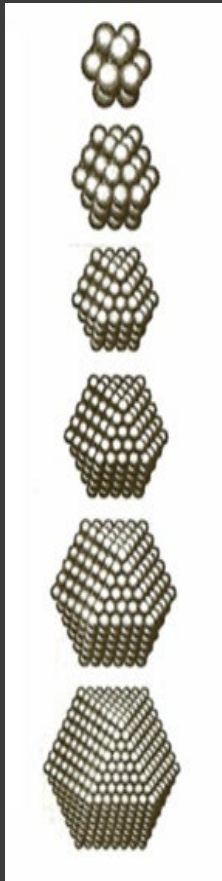
DOS : Density of states

- High surface area
  - Quantum size effects ;
  - Changing on total energy : less bulk and more superficial energy;
  - Defects of grains and nanoparticles;
  - High catalytic properties due high surface area: particles and agglomerates;
  - Optical properties;
  - Mechanical properties; etc.
- ... Nano is the scale at which surfaces and interfaces play a large role in materials properties and interfaces



Agglomerate	Total number of atoms (%)	Surface atoms (%)
-------------	---------------------------	-------------------

**1 Shell**



**13**

**92**

**2 Shells**



**55**

**76**

**3 Shells**



**147**

**63**

**4 Shells**



**309**

**52**

**5 Shells**



**561**

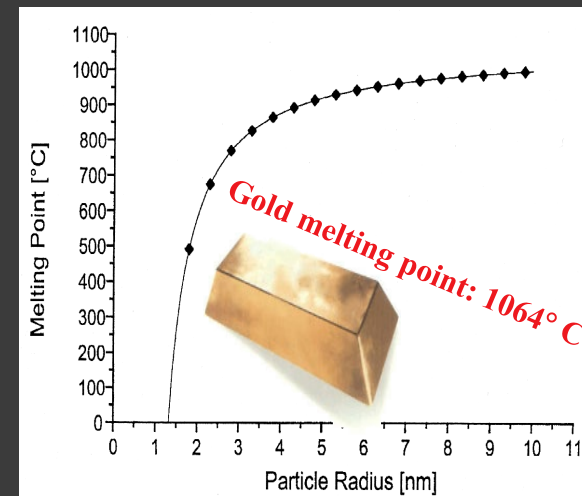
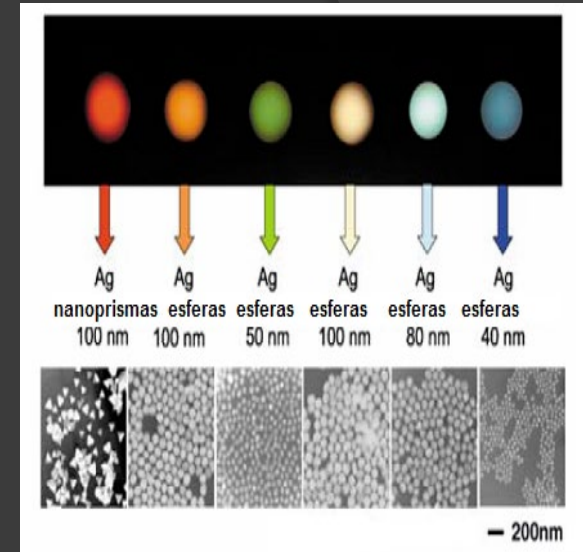
**45**

**6 Shells**



**1415**

**35**



Source: K.J. Klabunde, 2001

# Thermodynamics of surfaces

- $U(S, V, N)$
- $dU = TdS - PdV + \mu dN$  or
- $U = TS - PV + \mu N$ , Inserting the associated factor with superficial tension, we have:
- $U = TS - PV + \mu N + \gamma A$ , where:

U – Internal energy

T – Temperature

S – Entropy

P – Pressure

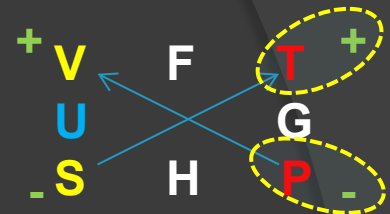
V – Volume

$\mu$  - Chemical potential

N – Number of moles

$\gamma$  - Superficial tension

A – Surface area

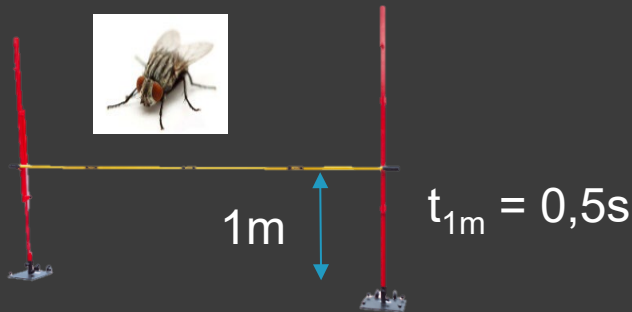


$U = f(S \text{ and } V)$ , then:  
 $U = +T dS - P dV$

“Valid Facts and Theoretical  
 Understanding Generates  
 Solutions of Hard Problems”

# Let's make an exercise calculation !

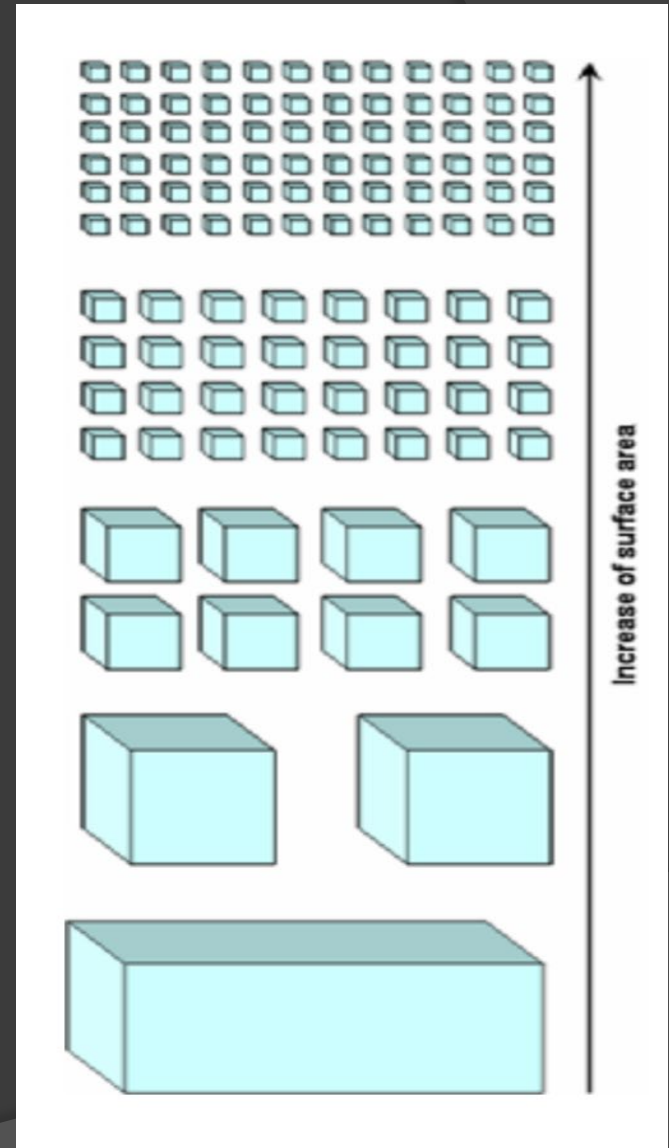
Potential energy  
(mgh) = g \* mass \* height ...  
 $9,8 * 12 * 10^{-6} * 1 = 1,176 * 10^{-7} \text{ J}$



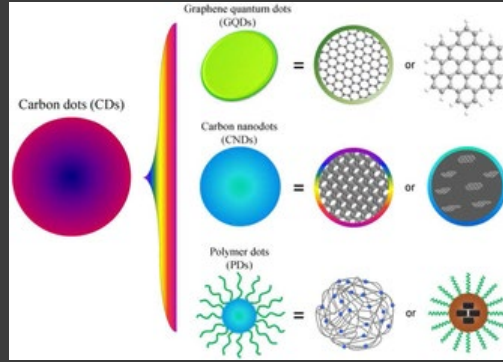
Fly (*Musca domestica*)  
Mass: 12 mg  
Velocity: 7.2 km/h = 2m/s  
[www.speedofanimals.com](http://www.speedofanimals.com)

Remembering ....  
 $1 \text{ erg} = 10^{-7} \text{ J}$  and  $1 \text{ erg/s} = 10^{-7} \text{ W}$

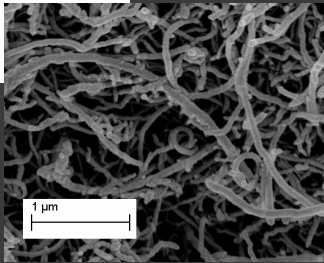
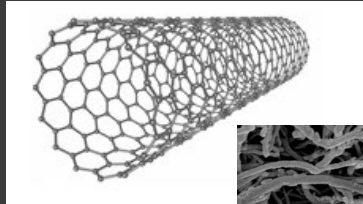
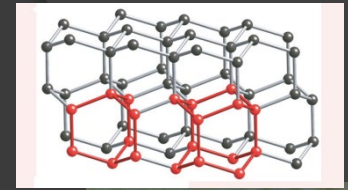
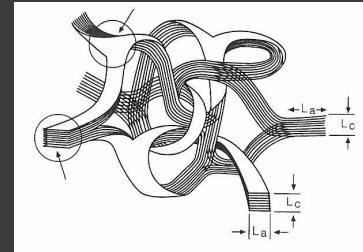
**Eg: Bonding rupture energy in a single crystal is about :  $\gamma \sim 1.200 \text{ erg.cm}^{-2}$**



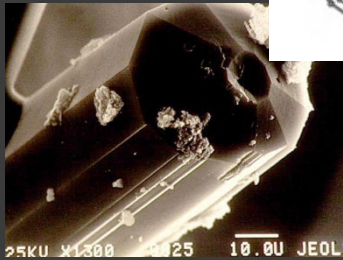
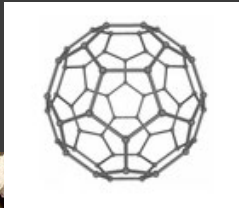
# The carbon family !



## Amorphous & carbon black

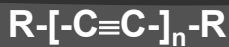


**CNT**  
**SWCNT**  
**DWCNT**  
**MWCNT**



<http://www.cite-sciences.fr>

$C_{60}$ ,  $C_{70}$ ,  $C_{540}$ , etc.

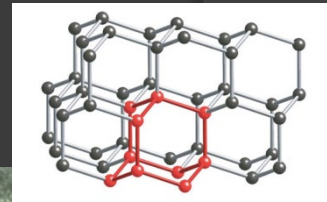


**Carbon dots**

**Lonsdaleite**

**Carbon Structures (Hybridization)**

**Diamond**



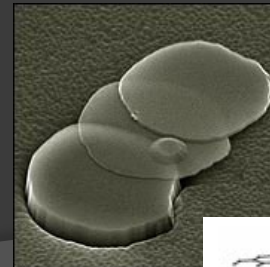
<http://www.nature.com>

<http://www.musee.ensmp.fr>

**Fullerenes**

**Graphene, r-GO, FLG**

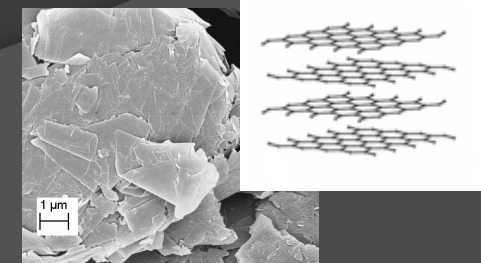
**Carbyns & carbolite**



**Carbon cones & onions**

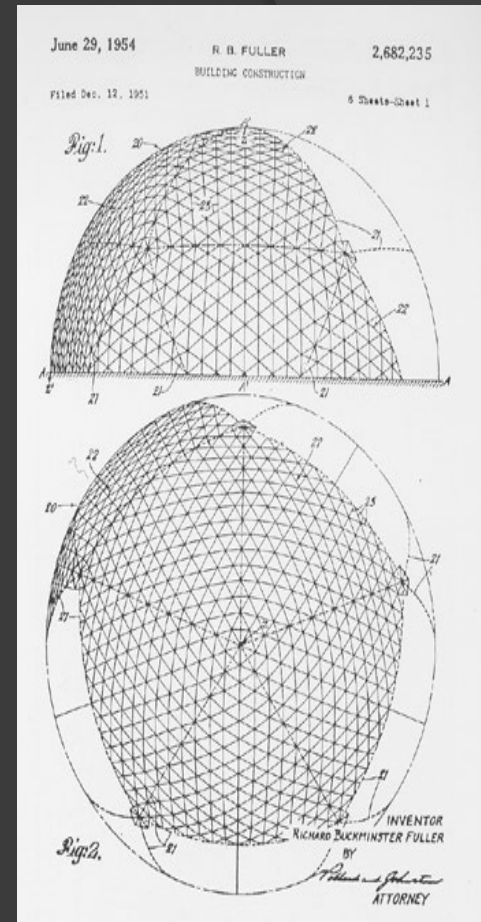
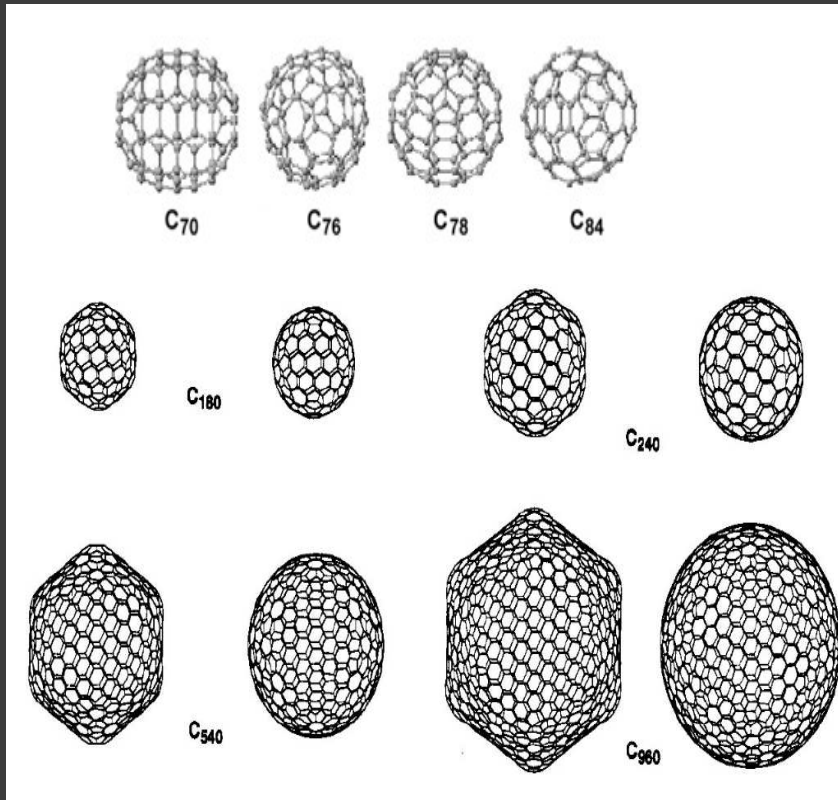


**Graphite**



<http://webmineral.com>

# Fullerenes



Buckminster Fuller,  
 Ford rotonde du dôme, 1953

# "The most symmetrical large molecule"

- Discovered in 1985
- Nobel prize Chemistry 1996, Curl, Kroto, and Smalley

The Nobel Prize in Chemistry 1996  
Robert F. Curl Jr., Sir Harold Kroto, Richard E. Smalley

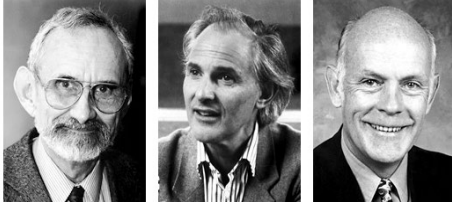
The Nobel Prize in Chemistry 1996

Nobel Prize Award Ceremony

Robert F. Curl Jr.

Sir Harold Kroto

Richard E. Smalley



Robert F. Curl Jr. Sir Harold W. Kroto Richard E. Smalley

The Nobel Prize in Chemistry 1996 was awarded jointly to Robert F. Curl Jr., Sir Harold W. Kroto and Richard E. Smalley "for their discovery of fullerenes".

*Pure Appl. Chem.*, Vol. 77, No. 5, pp. 843–923, 2005.  
DOI: 10.1351/pac200577050843  
© 2005 IUPAC

INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

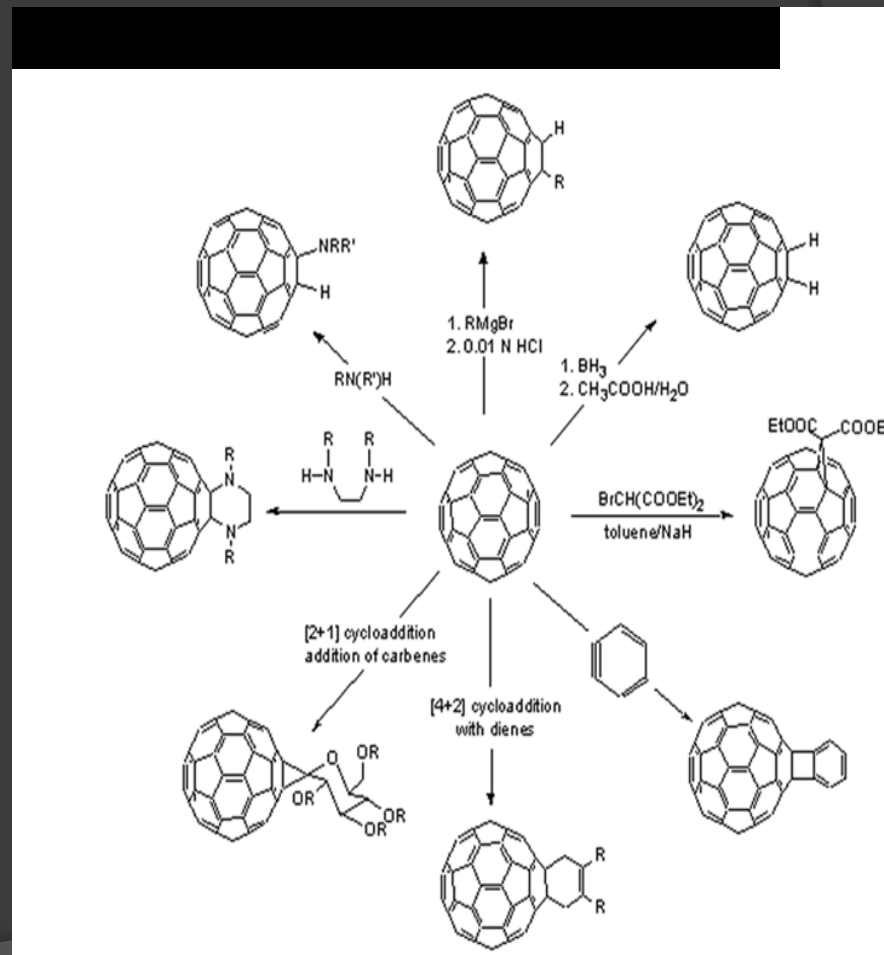
CHEMICAL NOMENCLATURE AND STRUCTURE REPRESENTATION DIVISION\*

## NUMBERING OF FULLERENES

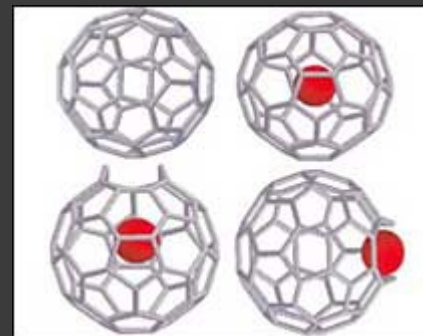
(IUPAC Recommendations 2005)

F. COZZI<sup>1</sup>, W. H. POWELL<sup>2,‡</sup>, AND C. THILGEN<sup>3</sup>

<sup>1</sup>Dipartimento di Chimica Organica e Industriale via Golgi, 19-I-20133, Milano, Italy; <sup>2</sup>1436  
Havencrest Ct., Columbus, OH 43220, USA; <sup>3</sup>Laboratorium für Organische Chemie - ETH Zürich,  
Wolfgang-Pauli-Strasse 10, CH-8093, Zürich, Switzerland

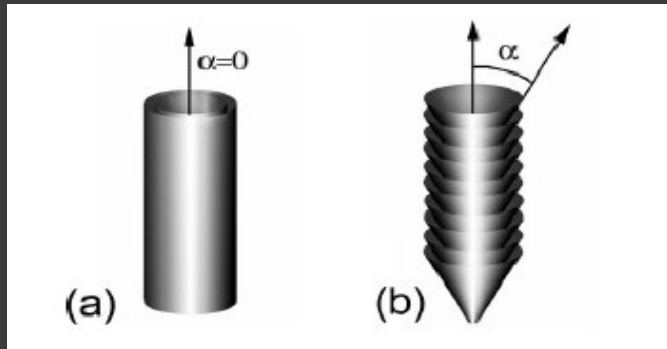


- Symmetric shape  
→ lubricant
- Large surface area  
→ catalyst
- High temperature (~500°C)
- High pressure
- Hollow  
→ caging particles
- Ferromagnet ...
  - polymerized C<sub>60</sub>
  - up to 220°C
- Chemically stable as graphite
  - most reactive at pentagons
- Crystal by weak van der Waals force
- Superconductivity:
  - K<sub>3</sub>C<sub>60</sub>: 19.2 K and
  - RbCs<sub>2</sub>C<sub>60</sub>: 33 K



Slide and images from:  
Kai de Lange Kristiansen

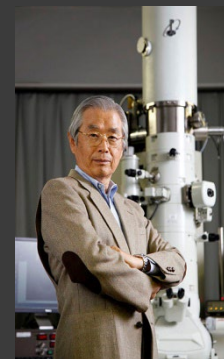
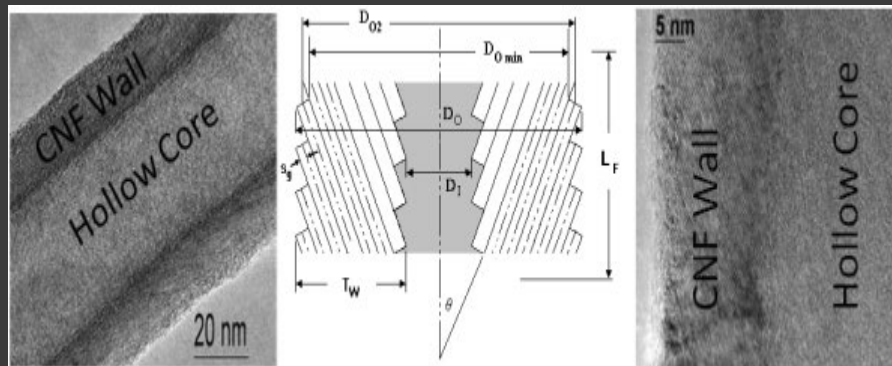
# Carbon nanotubes and carbon nanofibers ...



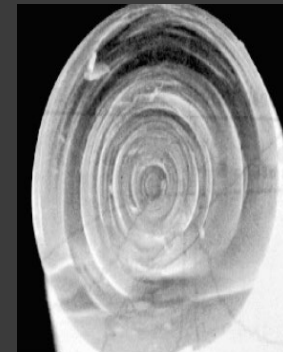
CNT first discovered by L. V. Radushkevich and V. M. Lukyanovich at: **1952** (*Journal of Physical Chemistry*, 62:88-95);

- Observed by Endo, Koyama and Oberlin in **1976** and;
- Finally described by Iijima at **1991**.

Nanofibers: **discovered in 1889**, US patent number: 450 e 480, by Hughes and Chambers.



Iijima



VGCF: Vapor growth carbon fibers

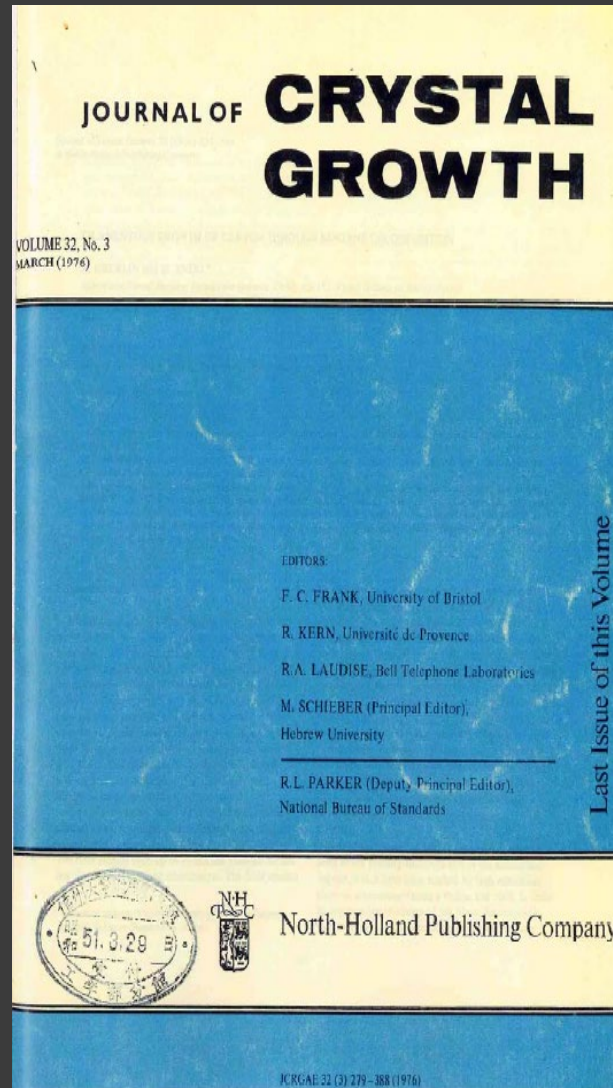


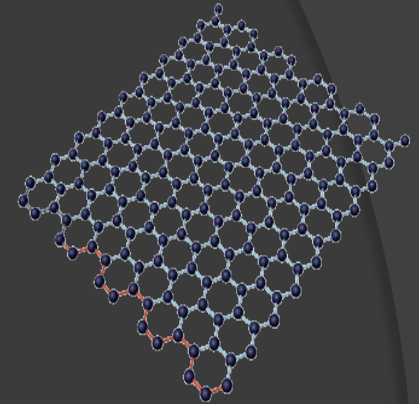
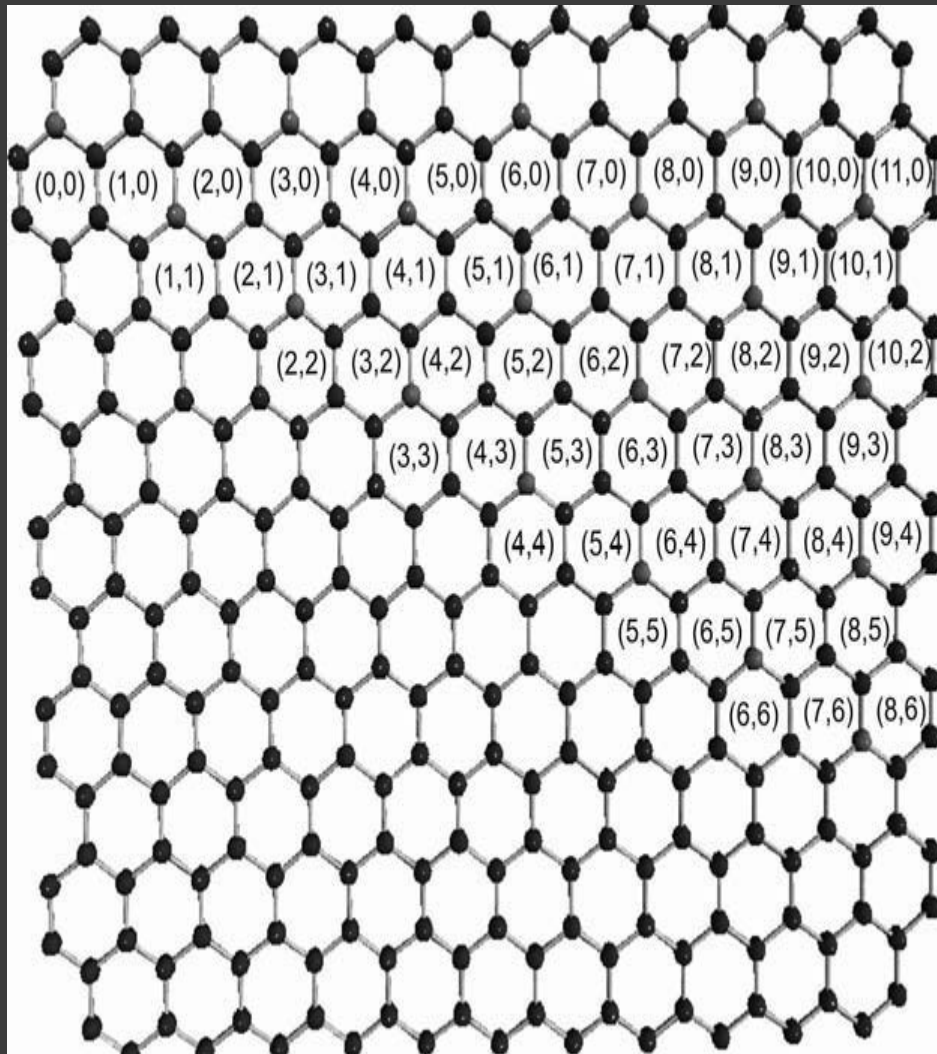
**J. Cryst. Grow. 32  
(1976) 335-349**

**A. Oberlin,  
M. Endo, e  
A. T. Koyama**



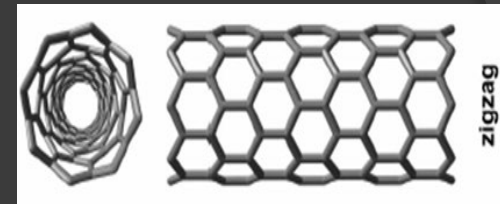
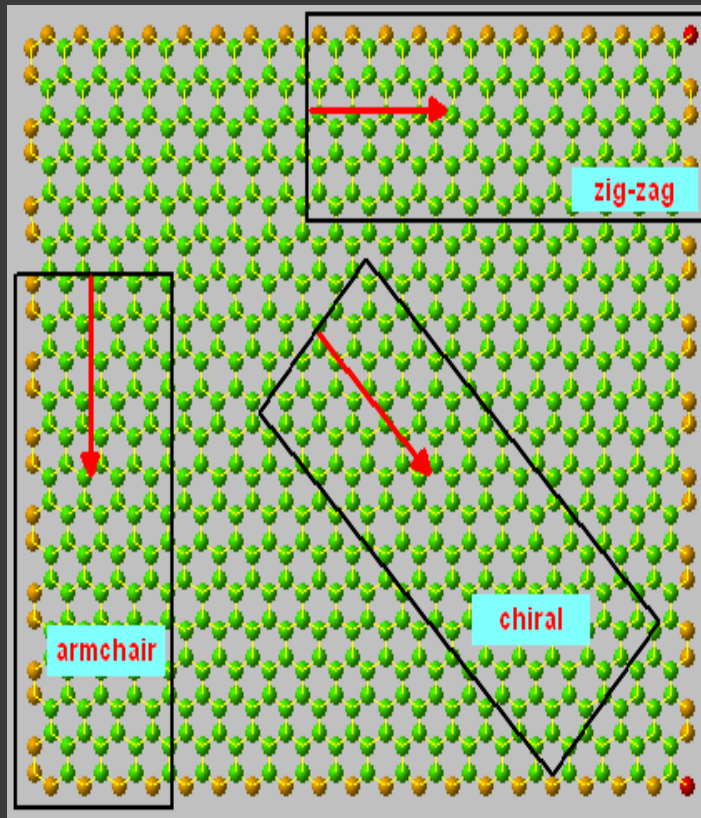
**Morinobu Endo**



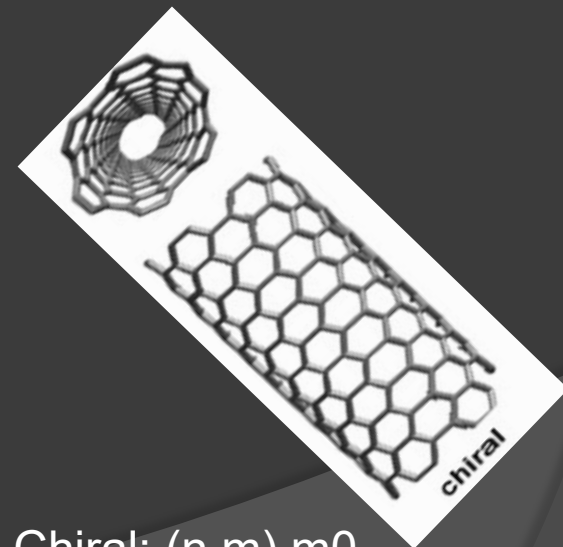


*Animation courtesy H. Nakahara and Saito Lab*

M. Endo, Ph.D. Thesis (1975)

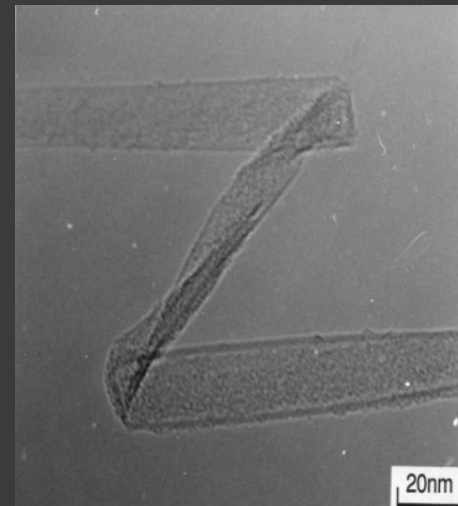
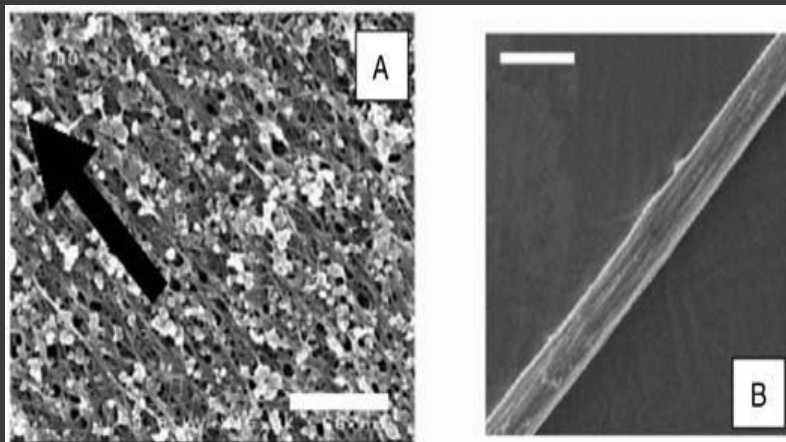


Zigzag:  $(n,0)$



Chiral:  $(n,m) m \neq 0$

Armchair:  $n = m$



M. Endo, K. Takeuchi, K. Kobori, K. Takahashi, H. W. Kroto and A. Sarkar, *Carbon*, 33, 873-881 (1995).

Scanning electron micrographs (SEMs) of a SWNT ribbon (scale bar = 667 nm)

(a) and a SWNT fiber (scale bar = 25  $\mu\text{m}$ )  
(b) each showing the alignment of the SWNTs within the structure.

(c) An optical micrograph of a SWNT fiber tied in a knot showing the high flexibility and resistance to torsion (fiber diameter = 15  $\mu\text{m}$ ).

From B. Vigolo, A. Pénicaud, C. Coulon, C. Sauder, R. Pailler, C. Journet, P. Bernier, and P. Poulin. Macroscopic fibers and ribbons of oriented carbon nanotubes, *Science*, 290, 1331–1334, 2000.

- **Young Modulus (stiffness):**

<b>Carbon nanotubes</b>	<b>1250 GPa</b>
Carbon fibers	425 GPa (max.)
High strength steel	200 GPa

- **Tensile strength (breaking strength)**

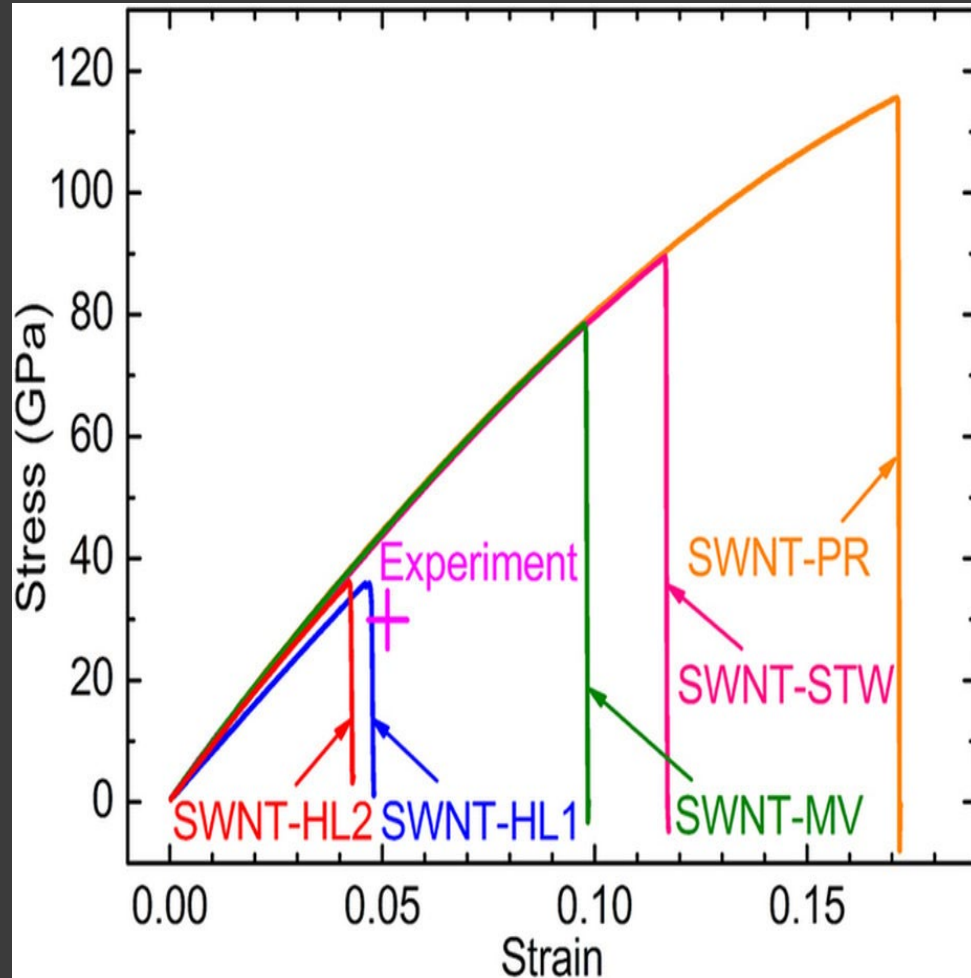
<b>Carbon nanotubes</b>	<b>11- 63 GPa</b>
Carbon fibers	3.5 - 6 GPa
High strength steel	~ 2 GPa

- **Elongation to failure : ~ 20-30 %**

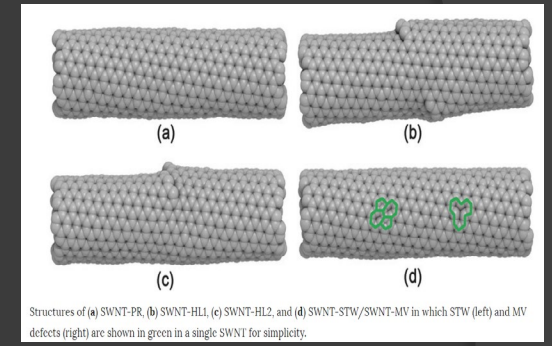
- **Density:**

<b>Carbon nanotube (SW)</b>	<b>1.33 – 1.40 gram / cm<sup>3</sup></b>
Aluminium	2.7 gram / cm <sup>3</sup>

# Yield strength limit: real (experimental) versus theory (simulated)



(2106)



MV - monovacancy;  
STW - Stone-Thrower-Wale defects  
(created during processing)  
PR - "Pristine" (without defects)

<http://www.nature.com/articles/srep20324/figures/3>  
<http://www.nature.com/articles/srep20324/figures/2>

# Mechanical properties of materials:

- Steel: 0,2 a 2 GPa
- Spider silk: 1 GPa ( one specimen ) : <https://phys.org/news/2013-06-spider-silk-nature-stronger-steel.html>
- Carbon nanotubes filaments: 1,3 GPa
- Single wall carbon nanotubes (with helicoidally defects ): 30-45 GPa ,  
(<http://www.nature.com/articles/srep20324>)



<https://phys.org/news/2013-01-nanotube-fibers-unmatched-combination-strength.html>



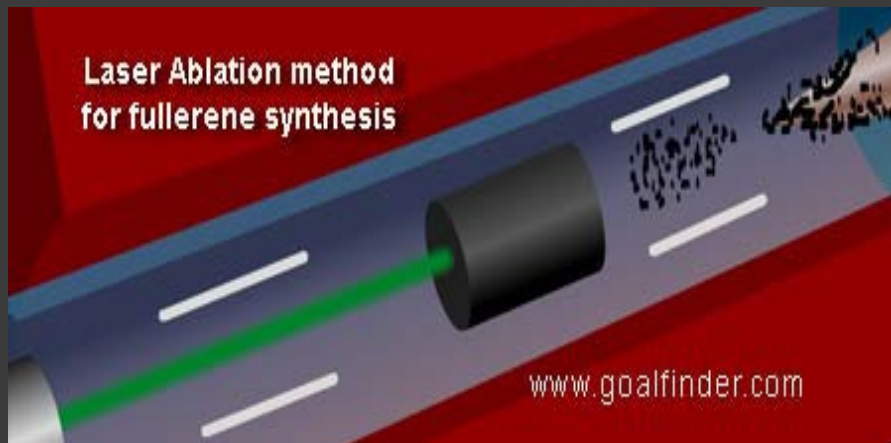
<https://i.ytimg.com/vi/AyYH0LNW8yo/hqdefault.jpg>

# Synthesis methods

Large number of methods:

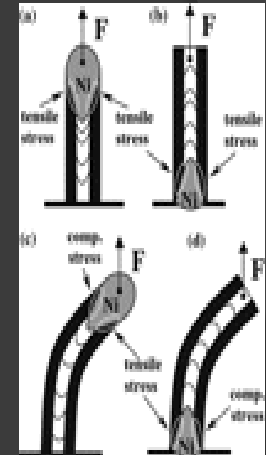
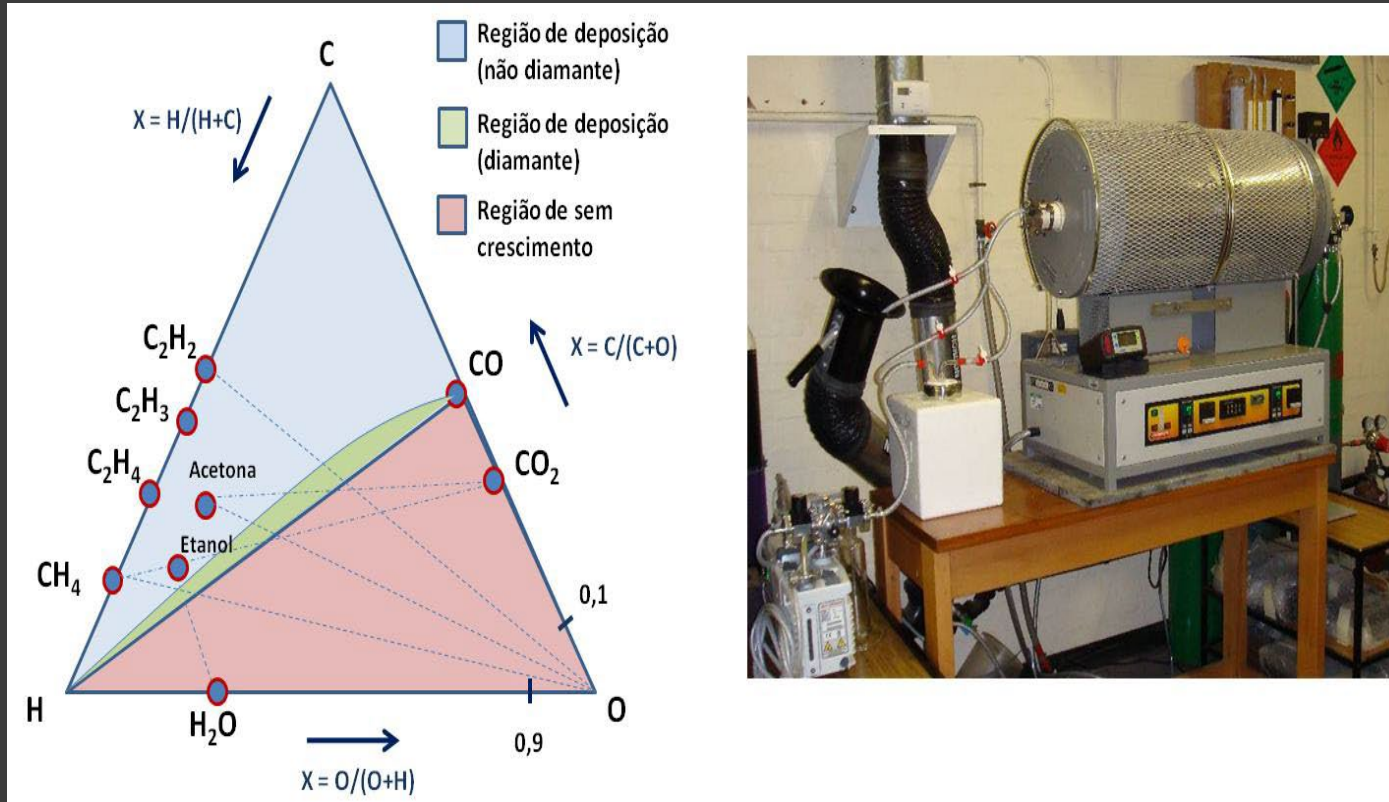
- CVD
- Catalytic combustion;
- Laser ablation
- Plasma

...





# CVD – Chemical Vapor Deposition



- Reactor atmosphere
- Thermal gradient (temperature)
- Type of catalyzers
- Source of carbon (chemical compounds)
- Pressure ...

# Catalyzers

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba		72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra		104 Rf	105 Db	106 : Sg	107 Bh	108 Hs	109 Mt	110 Uun	111 Uuu	112 Uub		114 Uuq		116 Uuh		118 Uuo
			57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
			89 Ac	90 Th	91 : Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

# “Volcano curve”: Catalyzer and CO (g) bonding - (Fisher-Tropsch reaction)

## Catalyzers

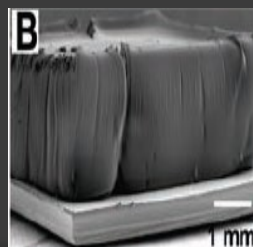
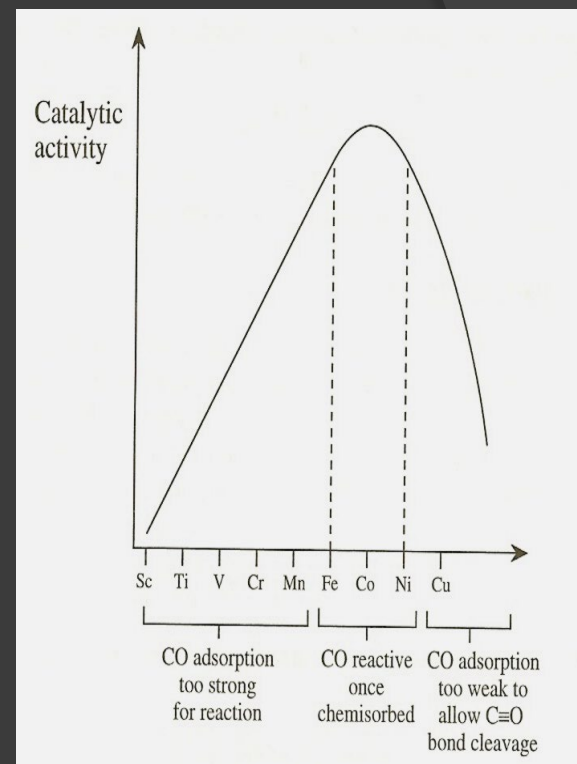
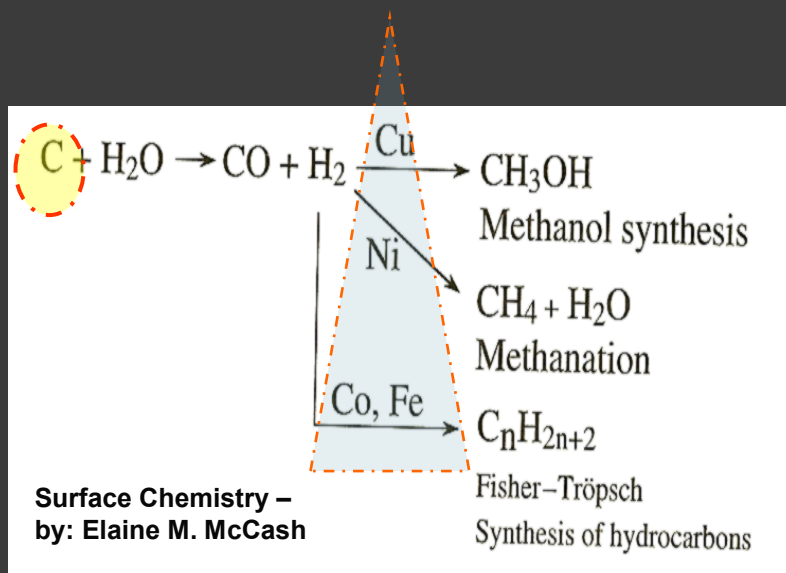
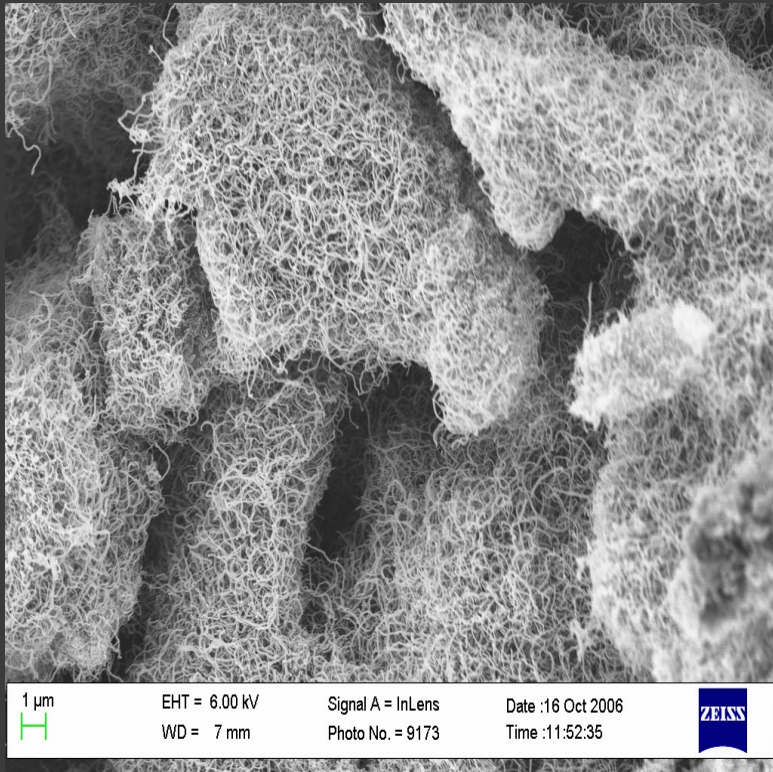


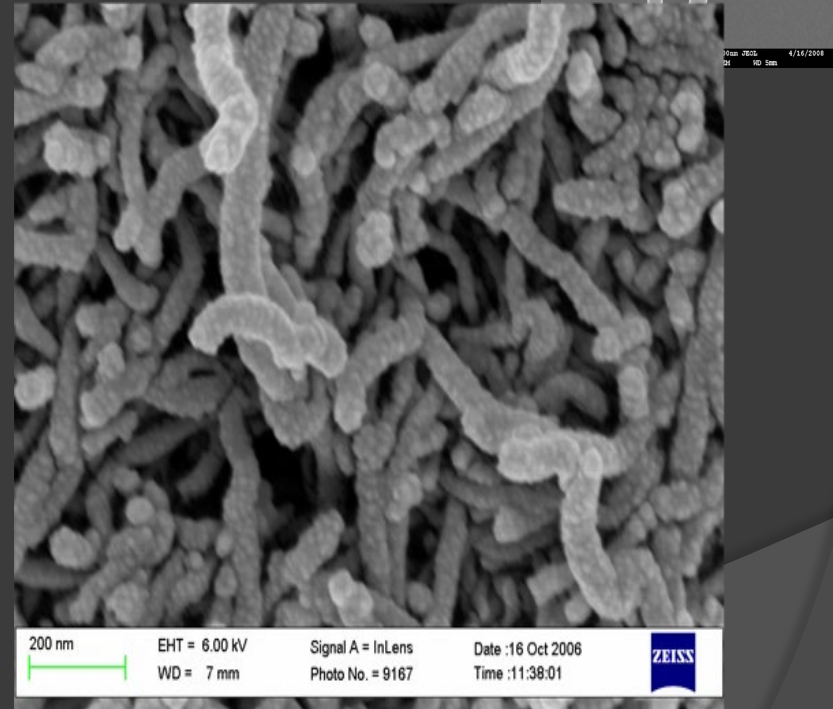
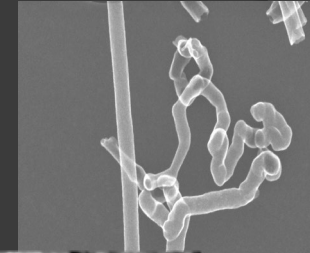
Fig. 1. SWNT forest grown with water-assisted CVD. (A) Picture of a 2.5-mm-tall SWNT forest on a 7-mm by 7-mm silicon wafer. A matchstick on the left and ruler with millimeter markings on the right is for size reference. (B) Scanning electron microscopy (SEM) image of the same SWNT forest. Scale bar, 1 mm.

Science 306, 1362, Hata et al.

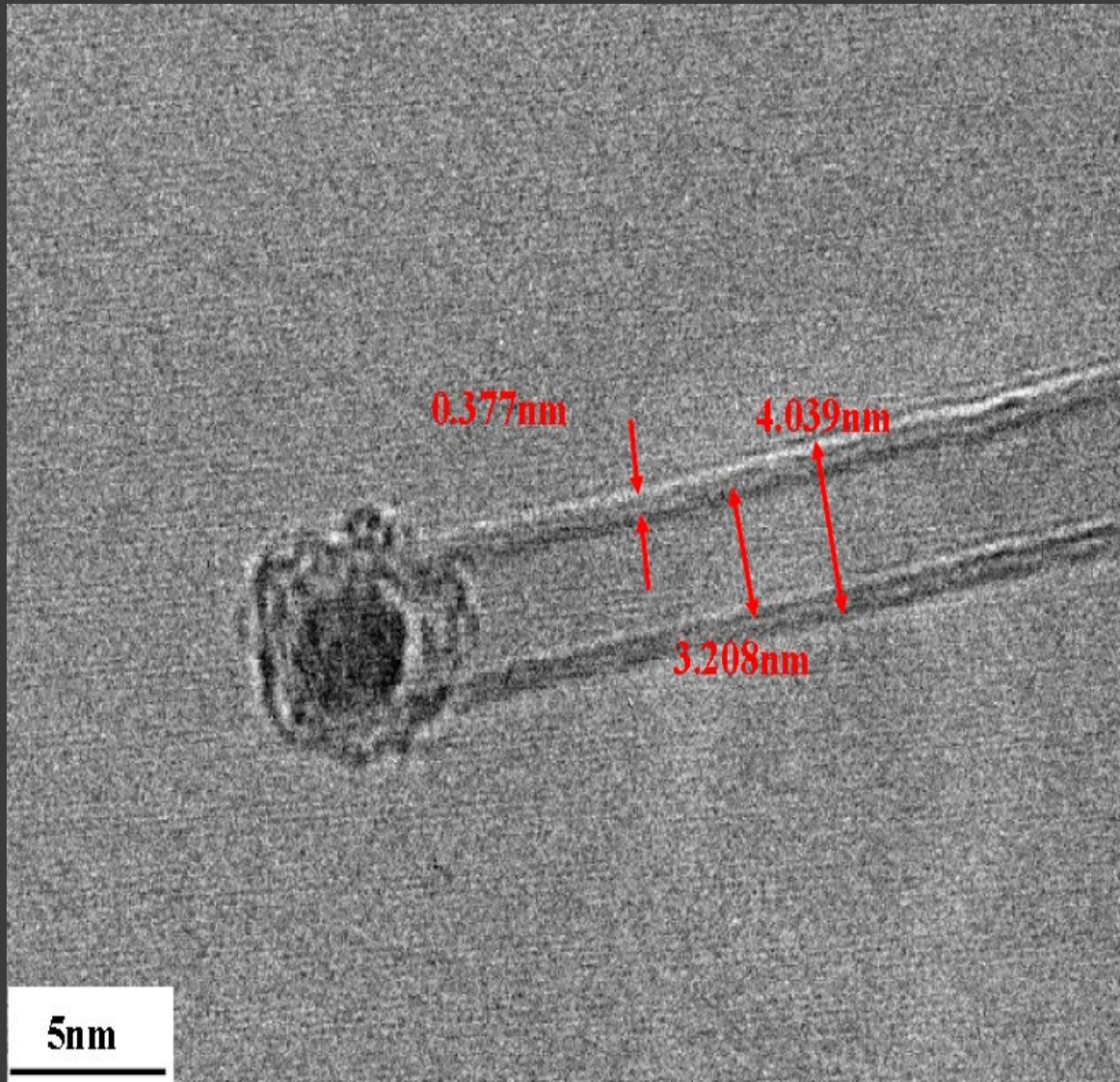
# Mutiwall carbon nanotubes



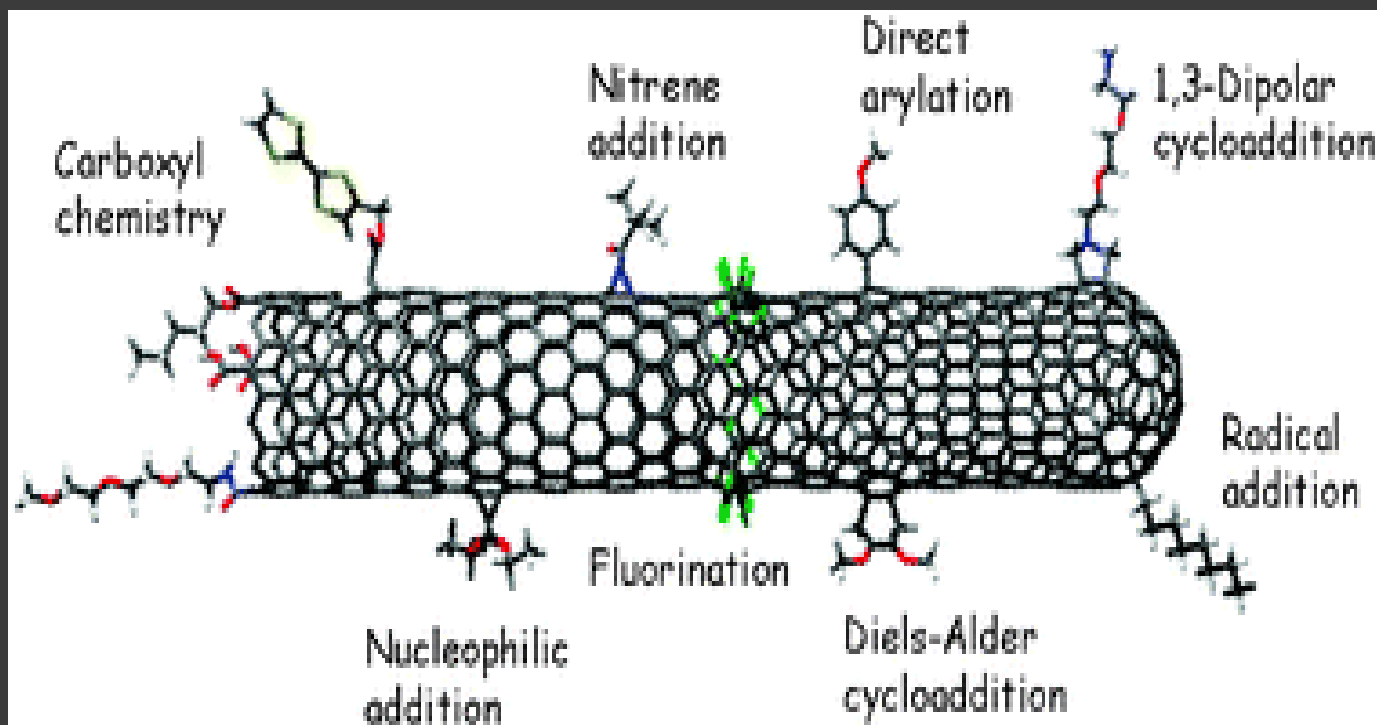
CVD – MWCNT  
(Industrial grade): +/- 95% purity



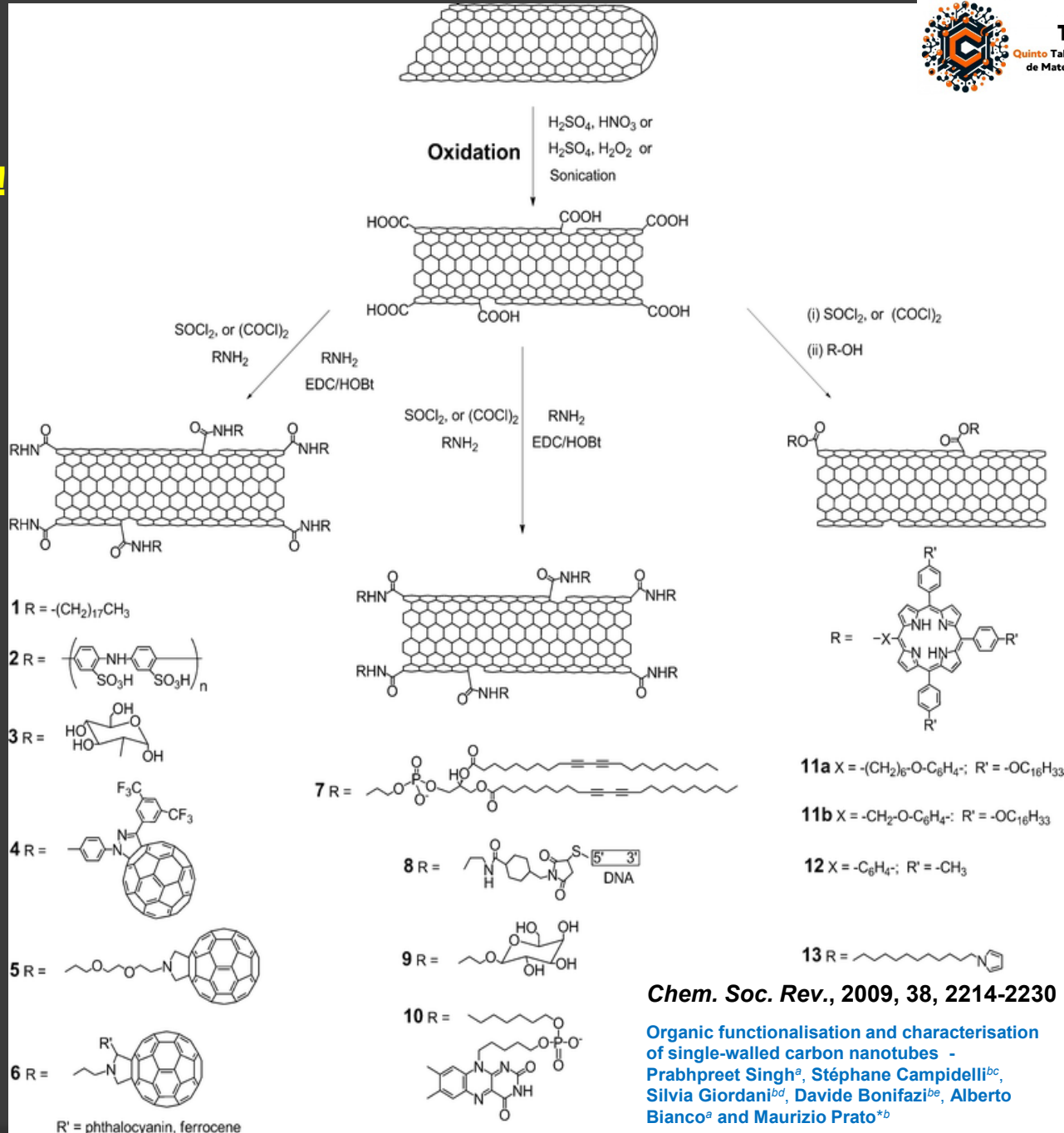
# Purification step !!!



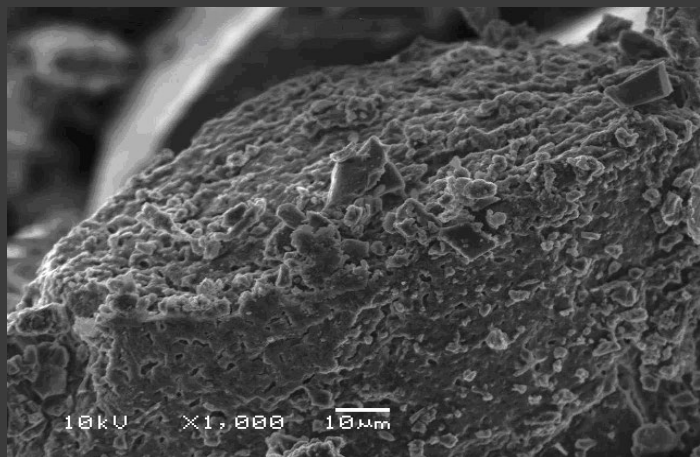
# Functionalization:



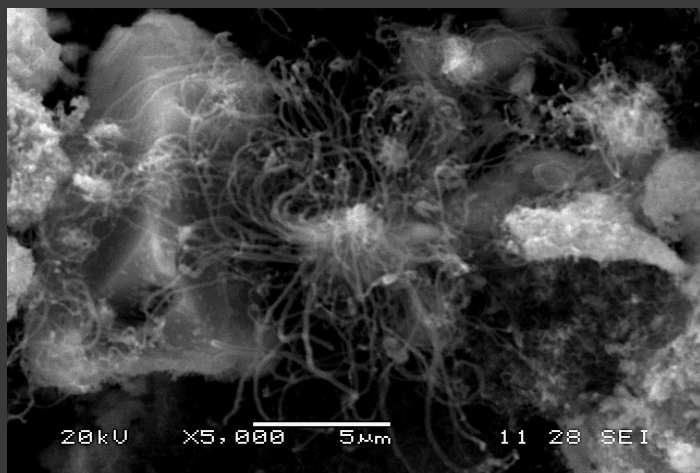
... More than 35 examples !



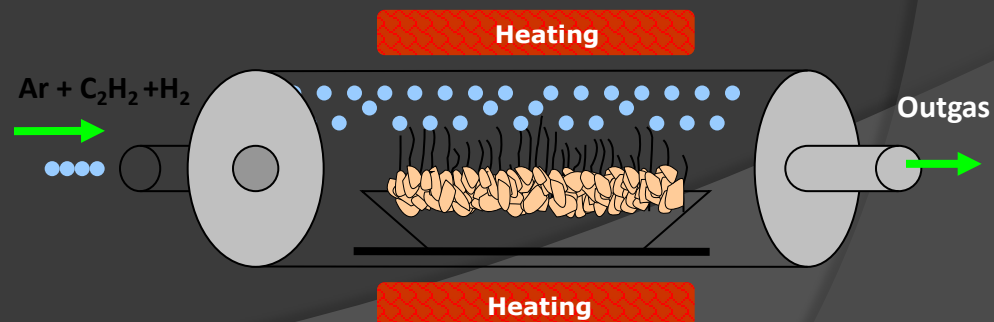
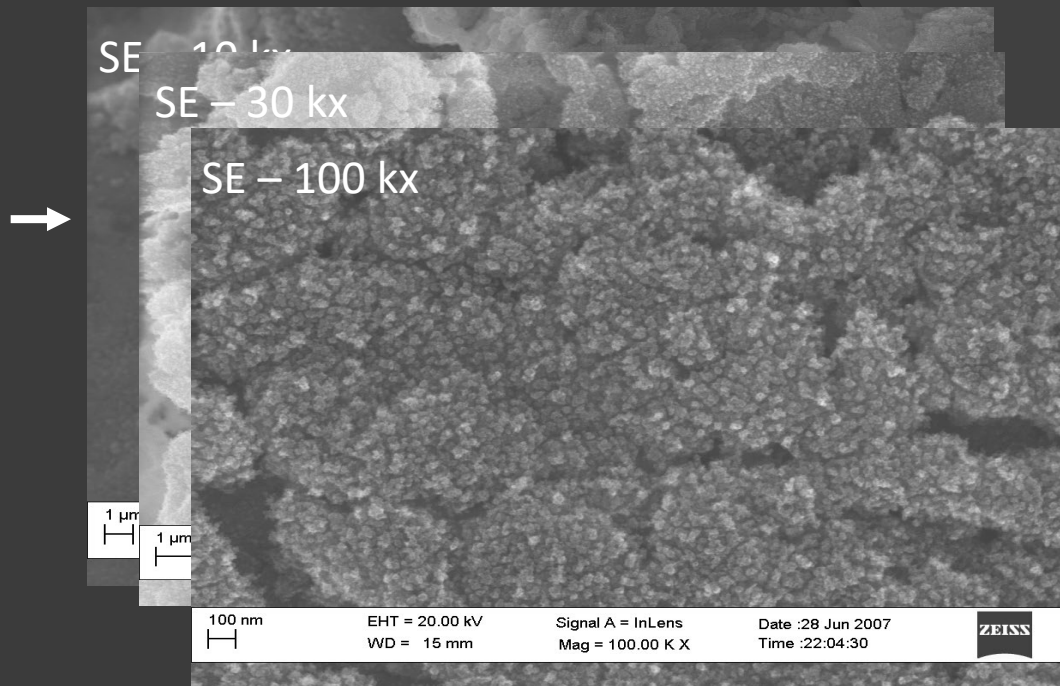
# Chemical Vapour DEPOSITION



SE image of a MgO grain coated with catalyst



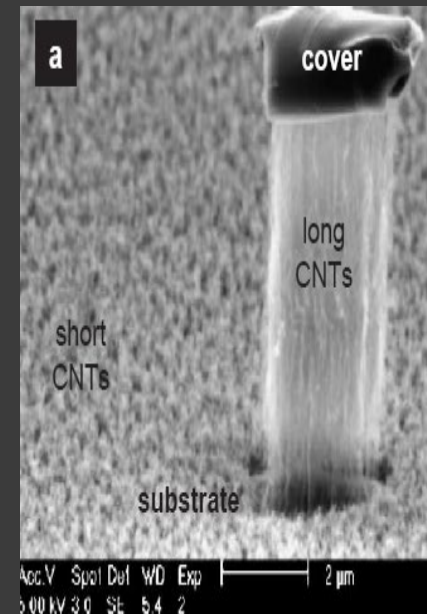
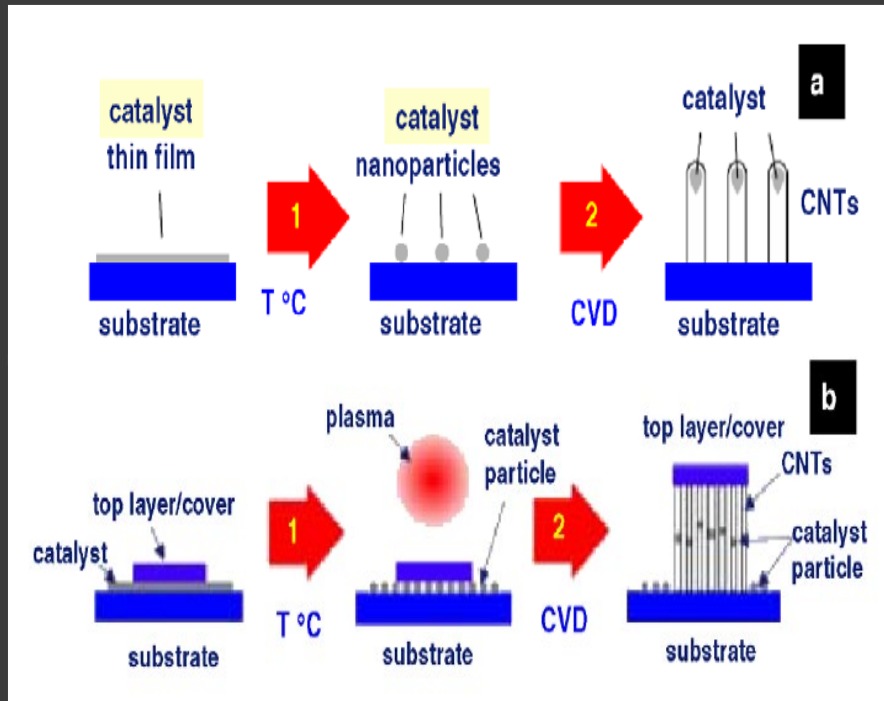
SE image of catalyst particles and CNTs



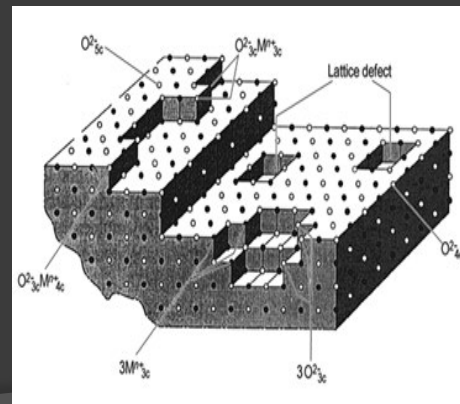
Guilherme Lenz (Pos-doc Program) & Rafael Sá, PhD Thesis (IC London) - 2006/2009



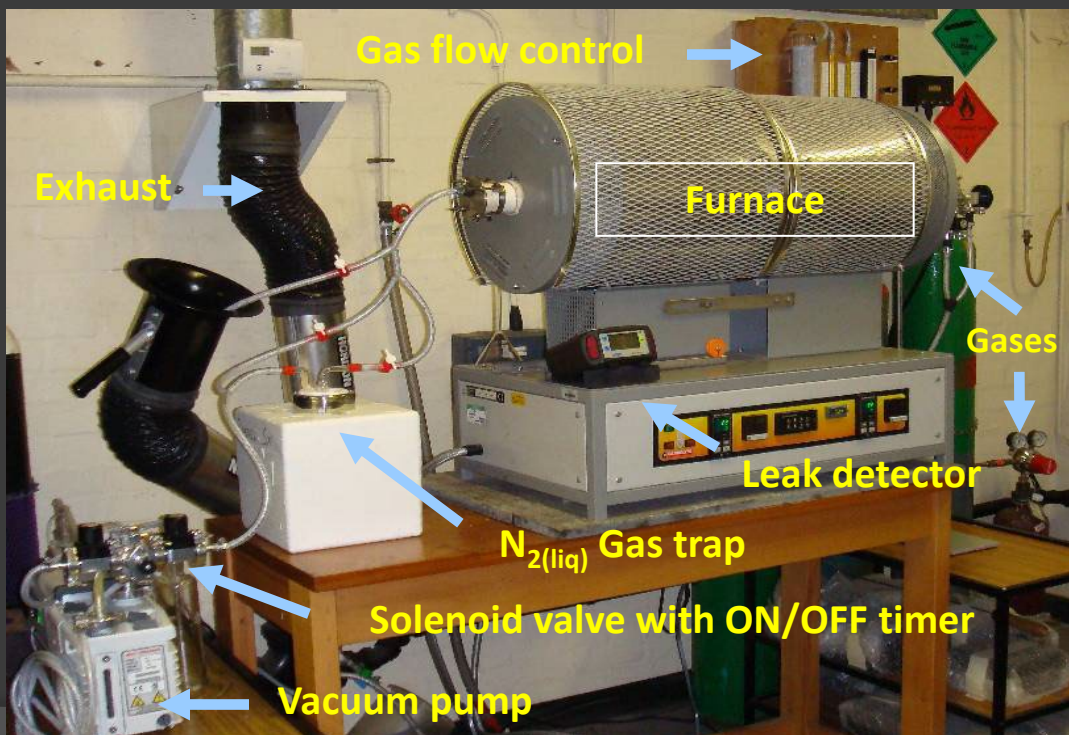
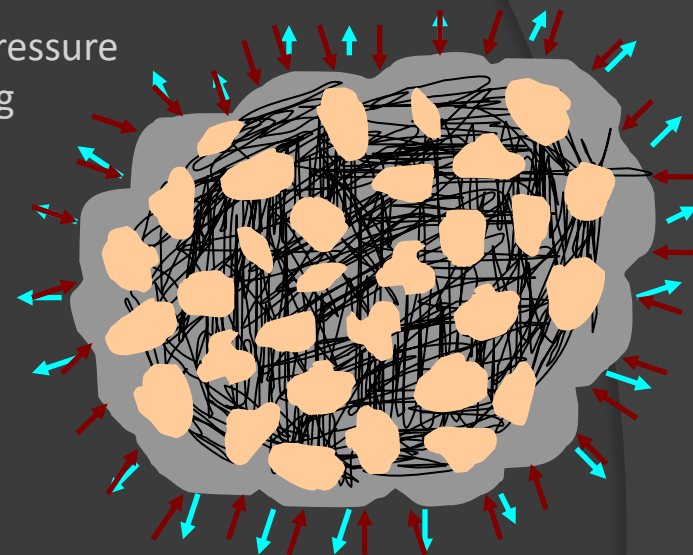
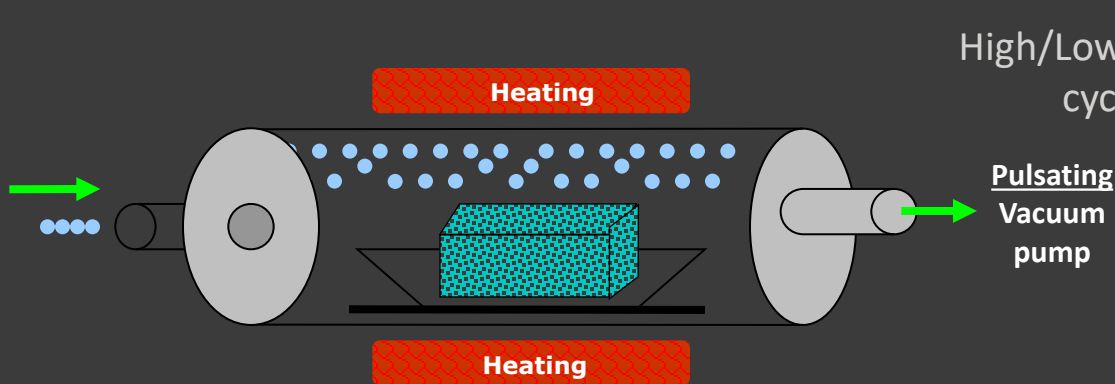
# Substrate a “sandwich” of surfaces



Z. Chen et al. / *Diamond & Related Materials* 15 (2006) 104–108

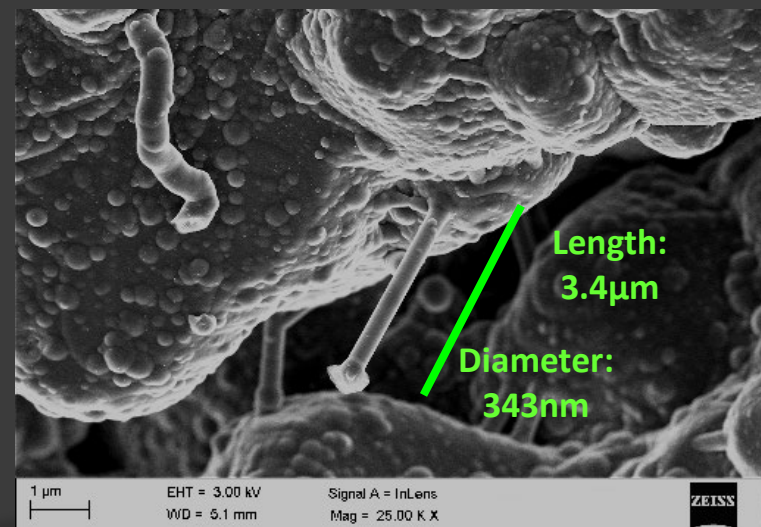
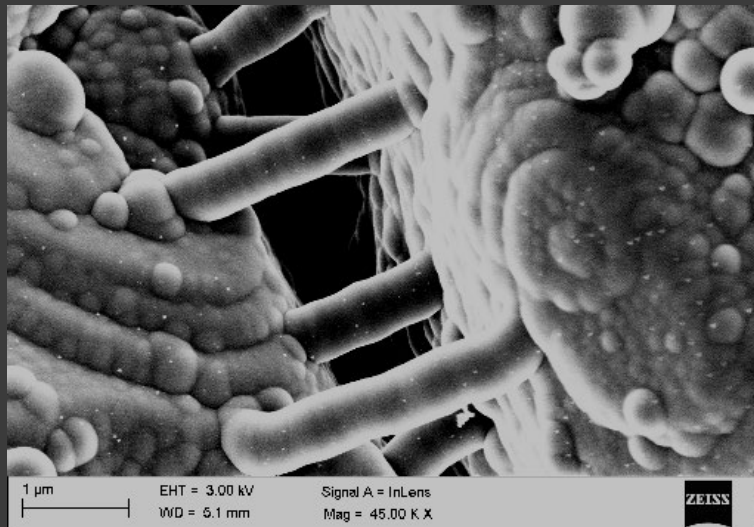
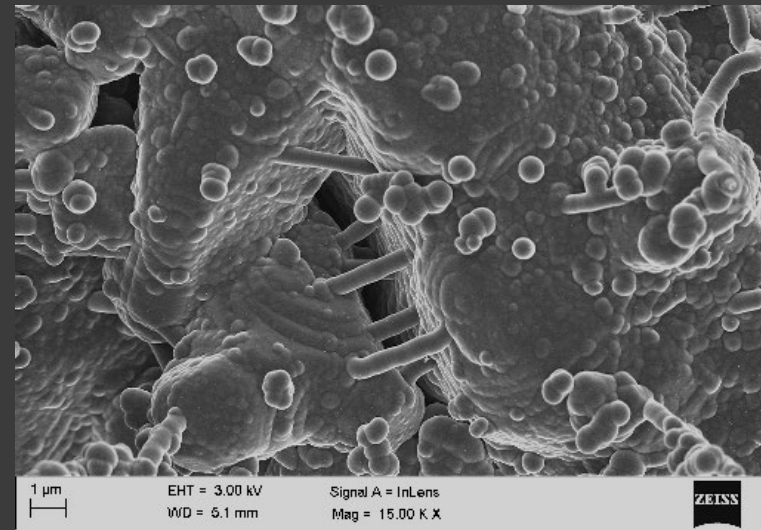
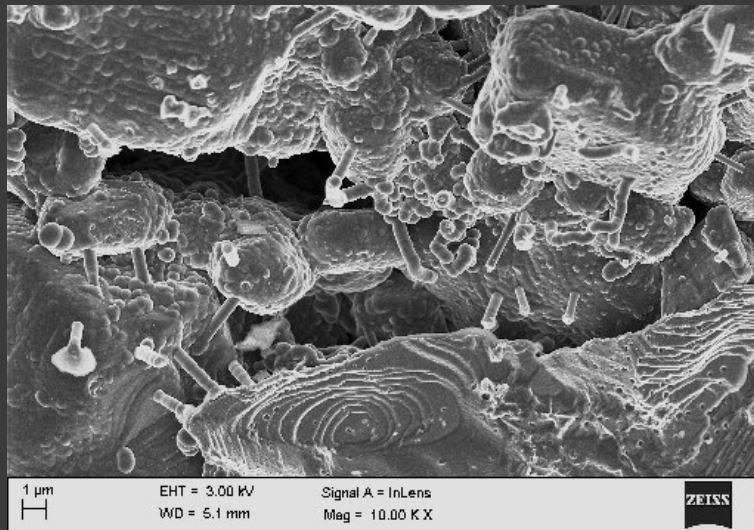


# In-situ Formation: CVI



Detail of gas mixture and flow controls

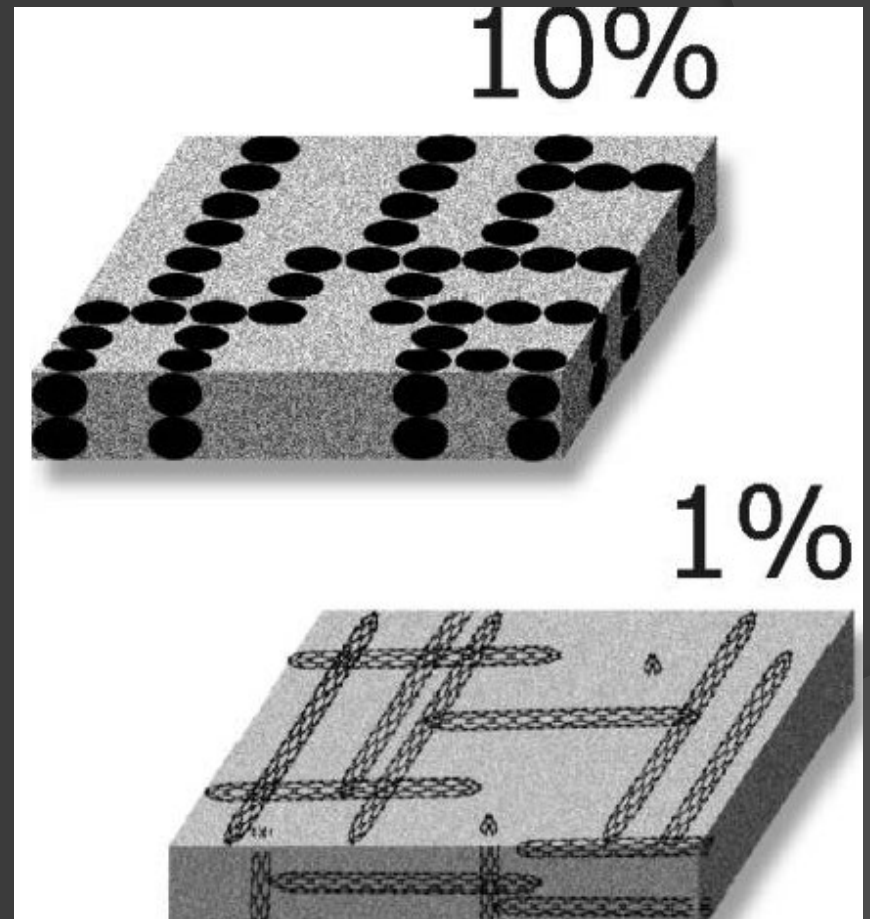
# In situ: MgO – MWCNT (CVI process)

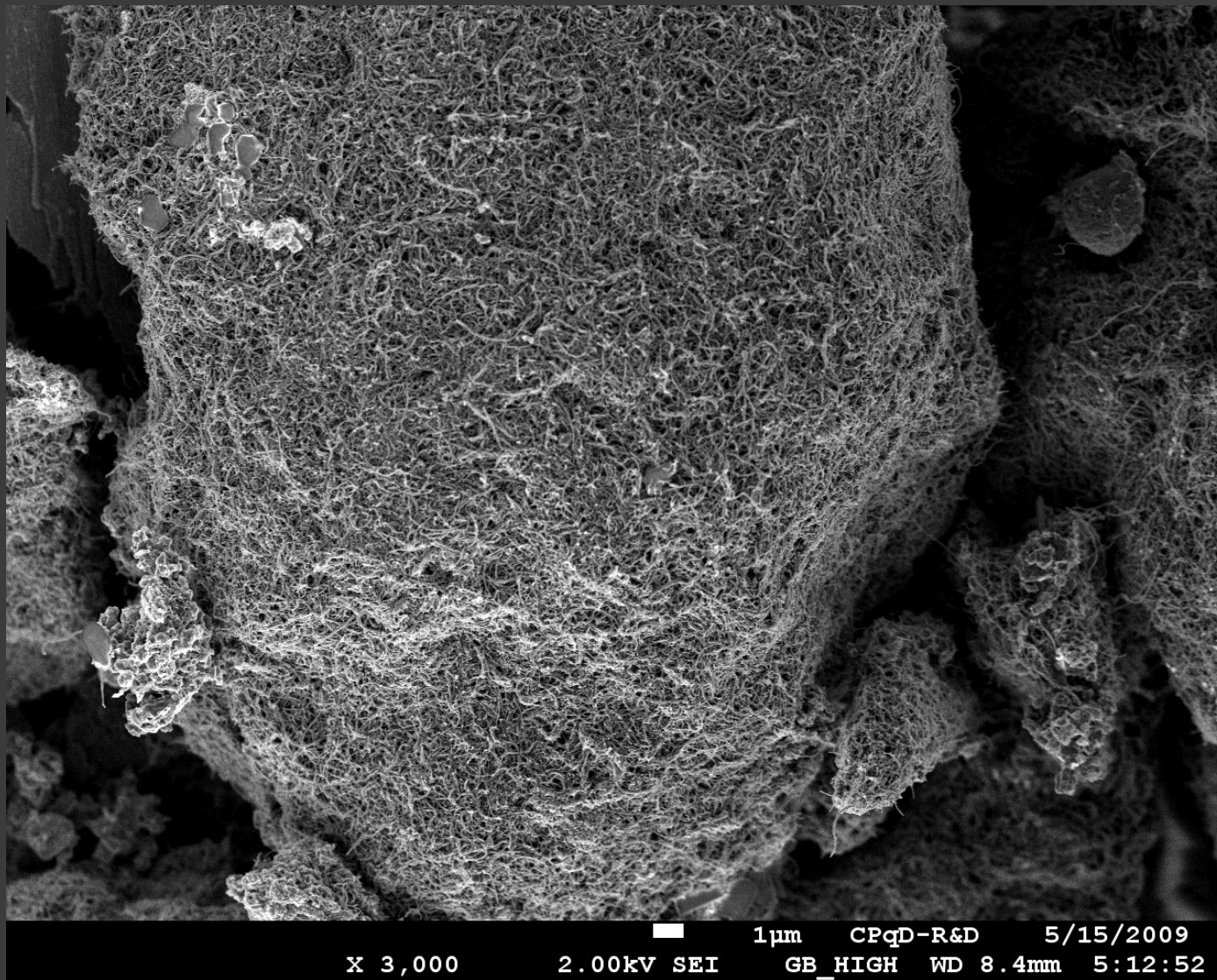


Thanks to Dr. Rafael G. de Sá

# CNT / polymer composite

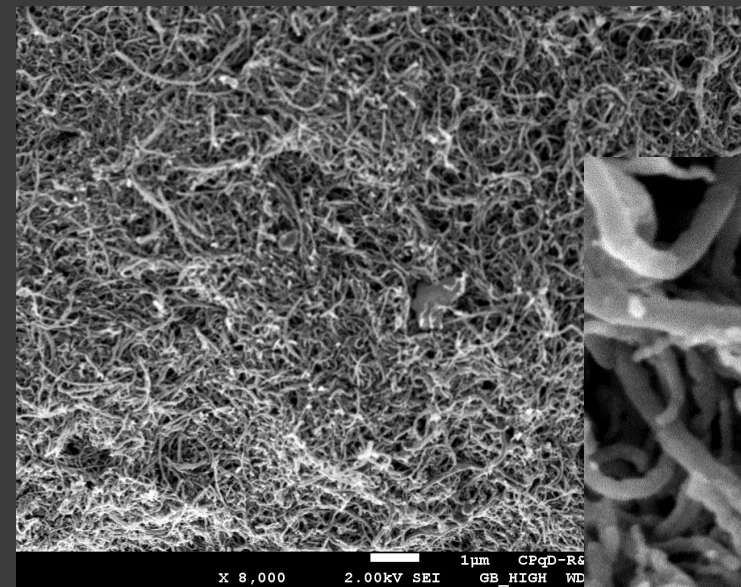
- Current technology
  - carbon black
  - 10 – 15 wt% loading
  - loss of mechanical properties
  
- CNT composites
  - 0.1 – 1 wt% loading
  - low percolation threshold





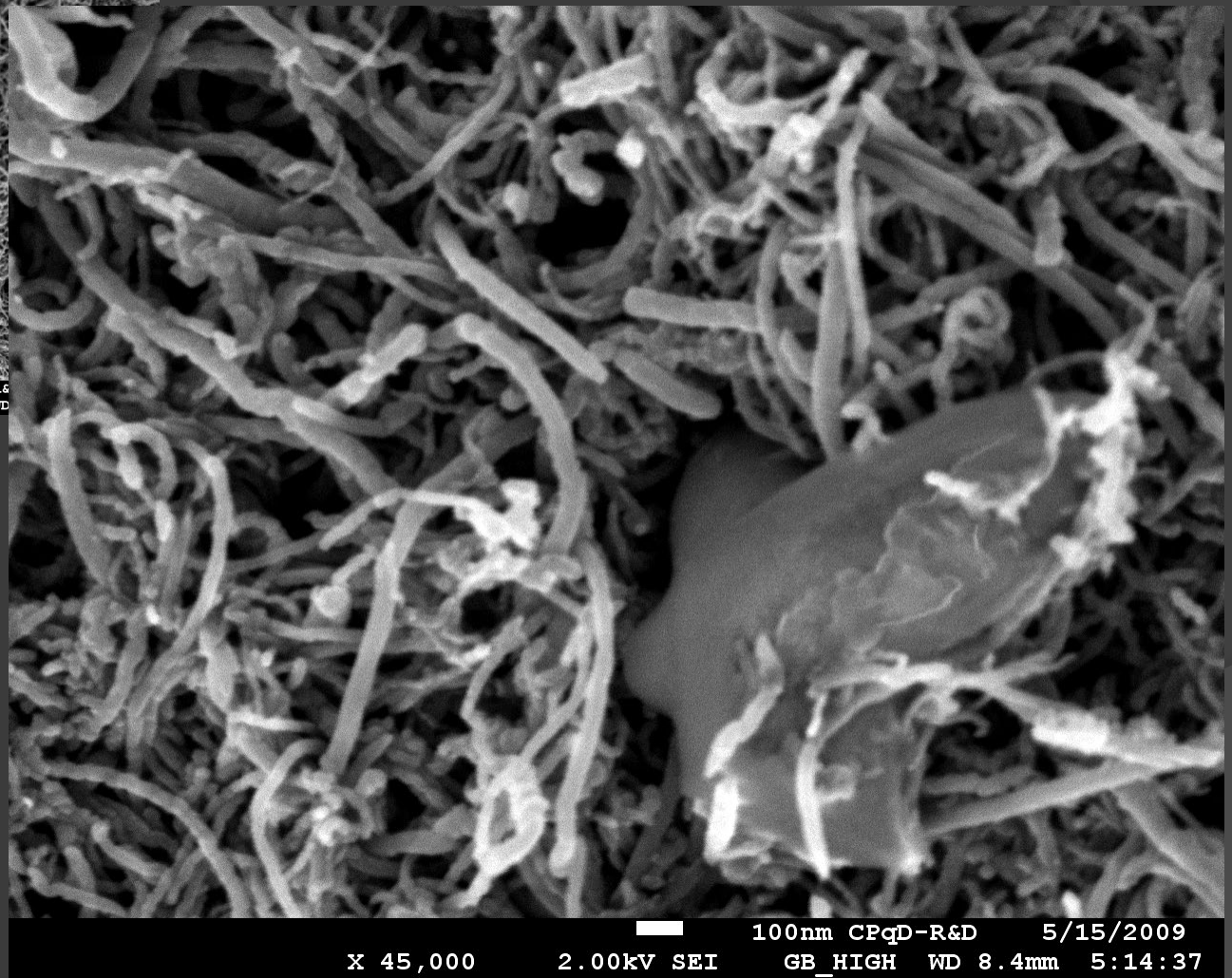
MWCNT  
(50 nm)

Surface treated  
(-OH)



MWCNT  
(50 nm)

Surface treated  
(-OH)



# CNT + phenolic resin + dispersants

## Key aspects:

viscosity;

dispersant  
chemistry;

temperature  
control;

& mixing  
equipment

A3

1%wt CNT



A4

1%wt CNT



A5

1%wt CNT



A6

2,5 %wt CNT



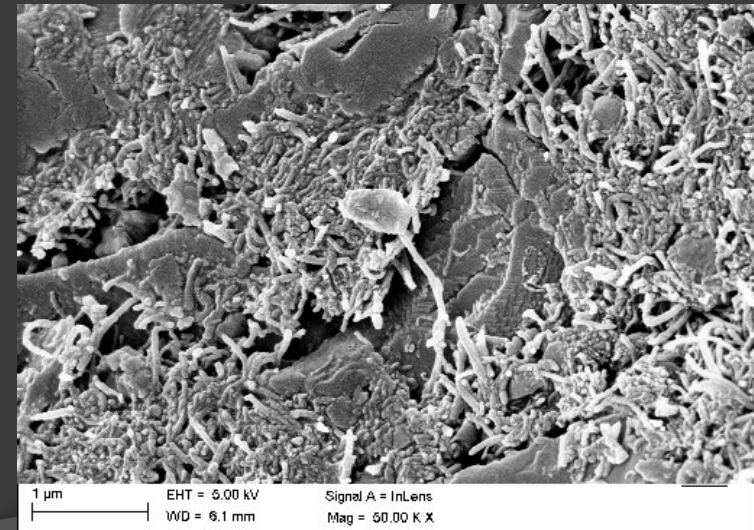
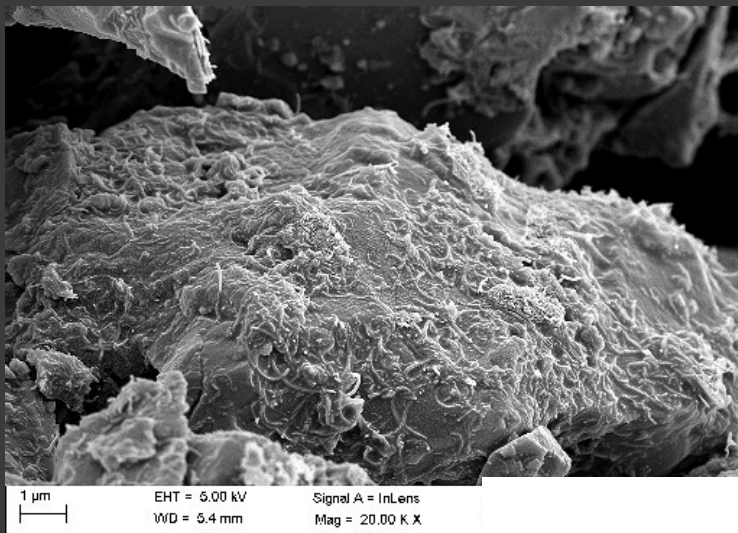
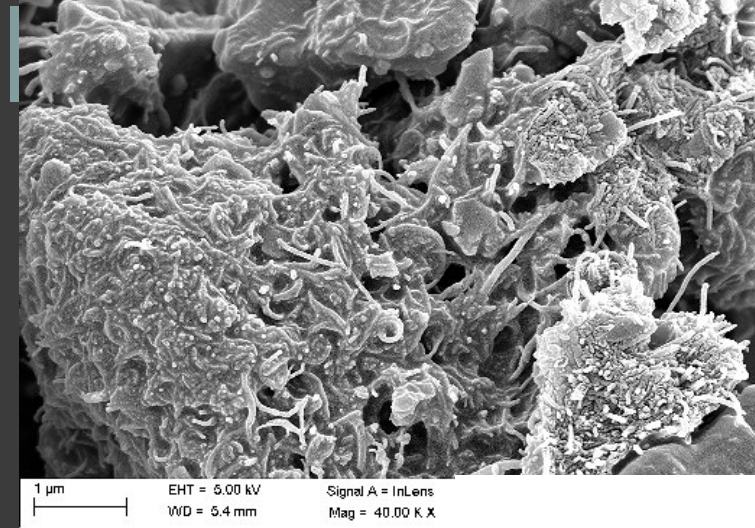
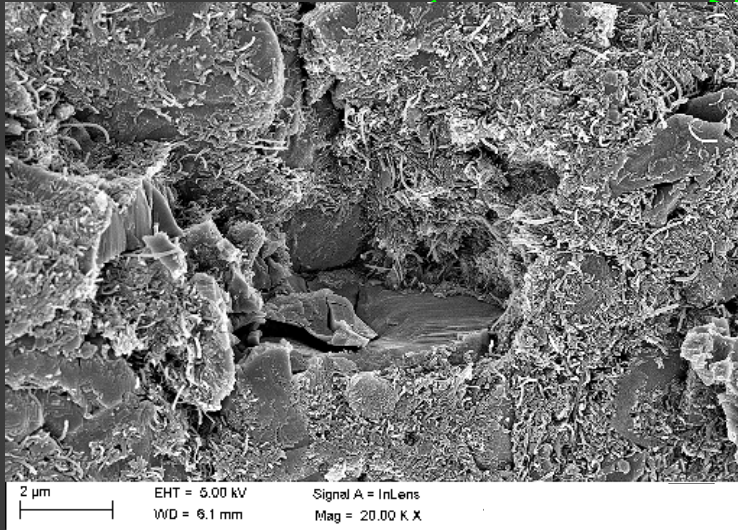
A7

2,5%wt CNT



# CNT Dispersed in Phenolic Resin

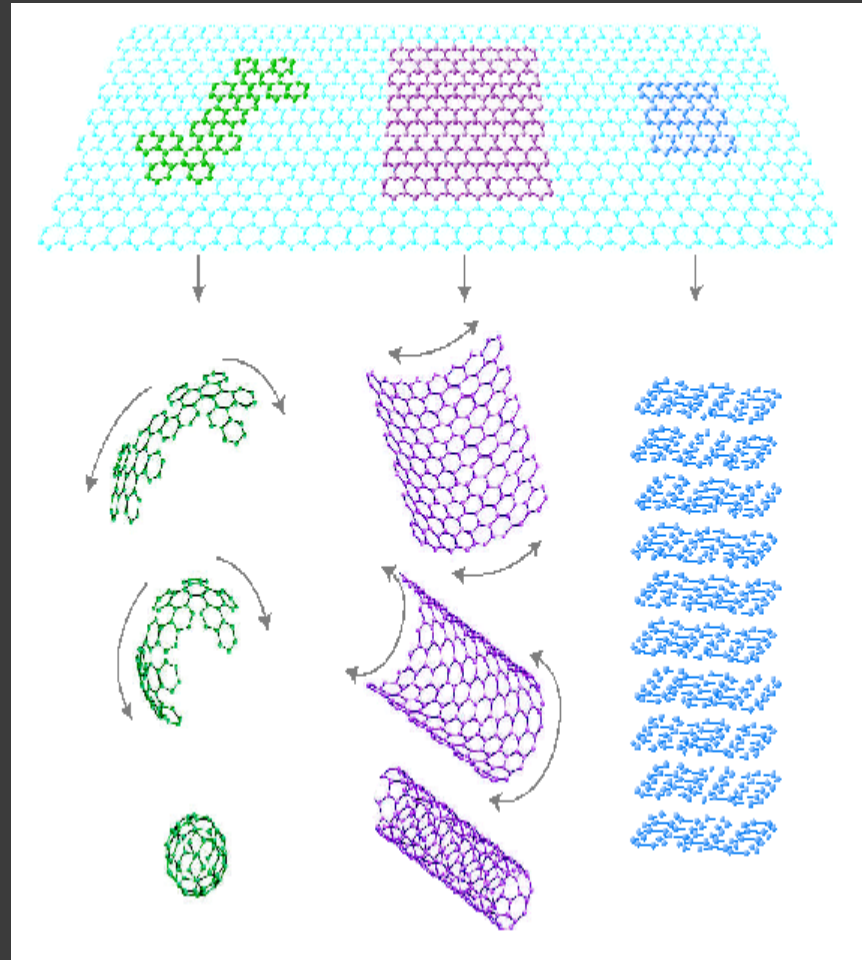
CNTs in binder



Thanks to Dr. Rafael G. de Sá



# Graphene



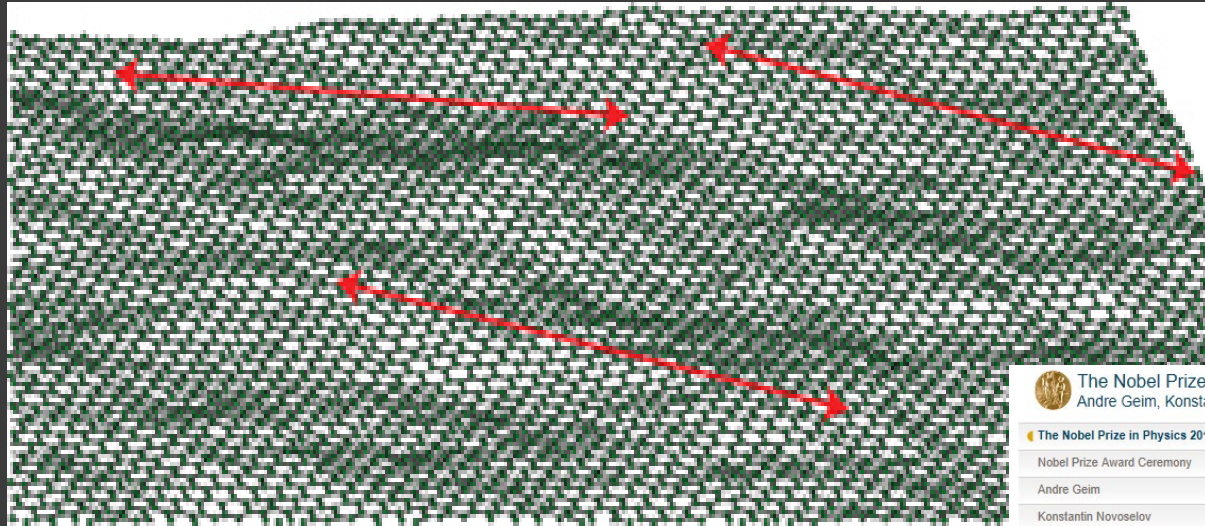
Graphene

Grafite

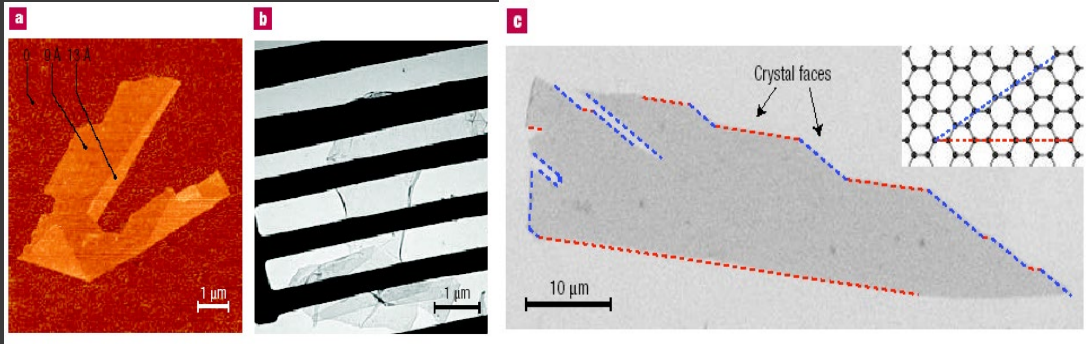
Fulerenos


Nanotubos

# The stability of graphene 2D? ... Came from the surface curvature !!



Red size:  
~800 nm



 The Nobel Prize in Physics 2010  
Andre Geim, Konstantin Novoselov

**The Nobel Prize in Physics 2010**

Nobel Prize Award Ceremony

Andre Geim

Konstantin Novoselov




Photo: U. Montan  
**Andre Geim**




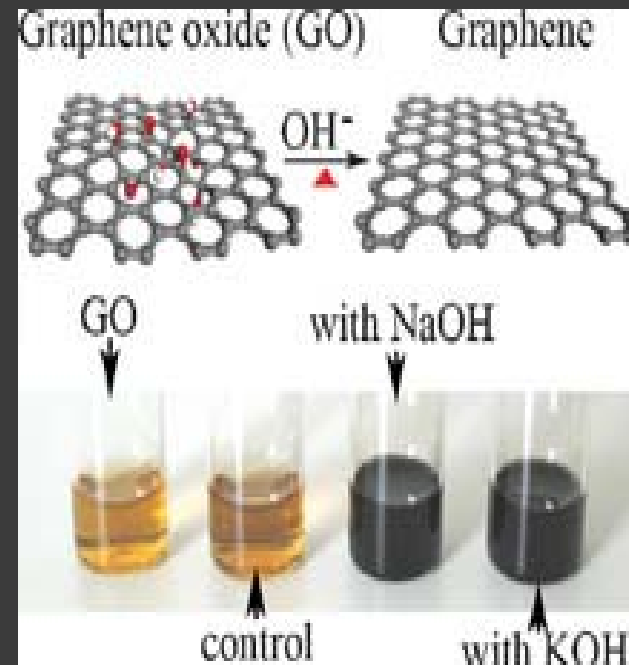
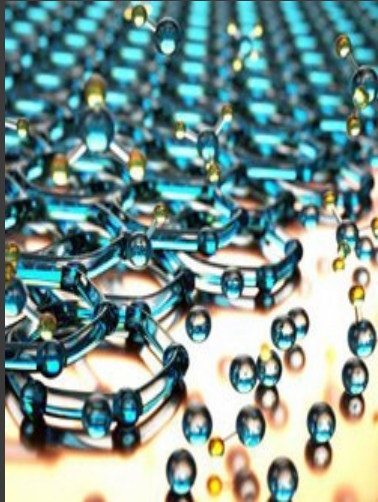
Photo: U. Montan  
**Konstantin Novoselov**

The Nobel Prize in Physics 2010 was awarded jointly to Andre Geim and Konstantin Novoselov "for groundbreaking experiments regarding the two-dimensional material graphene"

[http://www.nature.com/nmat/journal/v6/n11/fig\\_tab/nmat2011\\_F1.html#figure-title](http://www.nature.com/nmat/journal/v6/n11/fig_tab/nmat2011_F1.html#figure-title)

A. K. Geim & K. S. Novoselov. The rise of graphene. *Nature Materials* Vol 6 183-191 (March 2007)

# Graphene synthesis

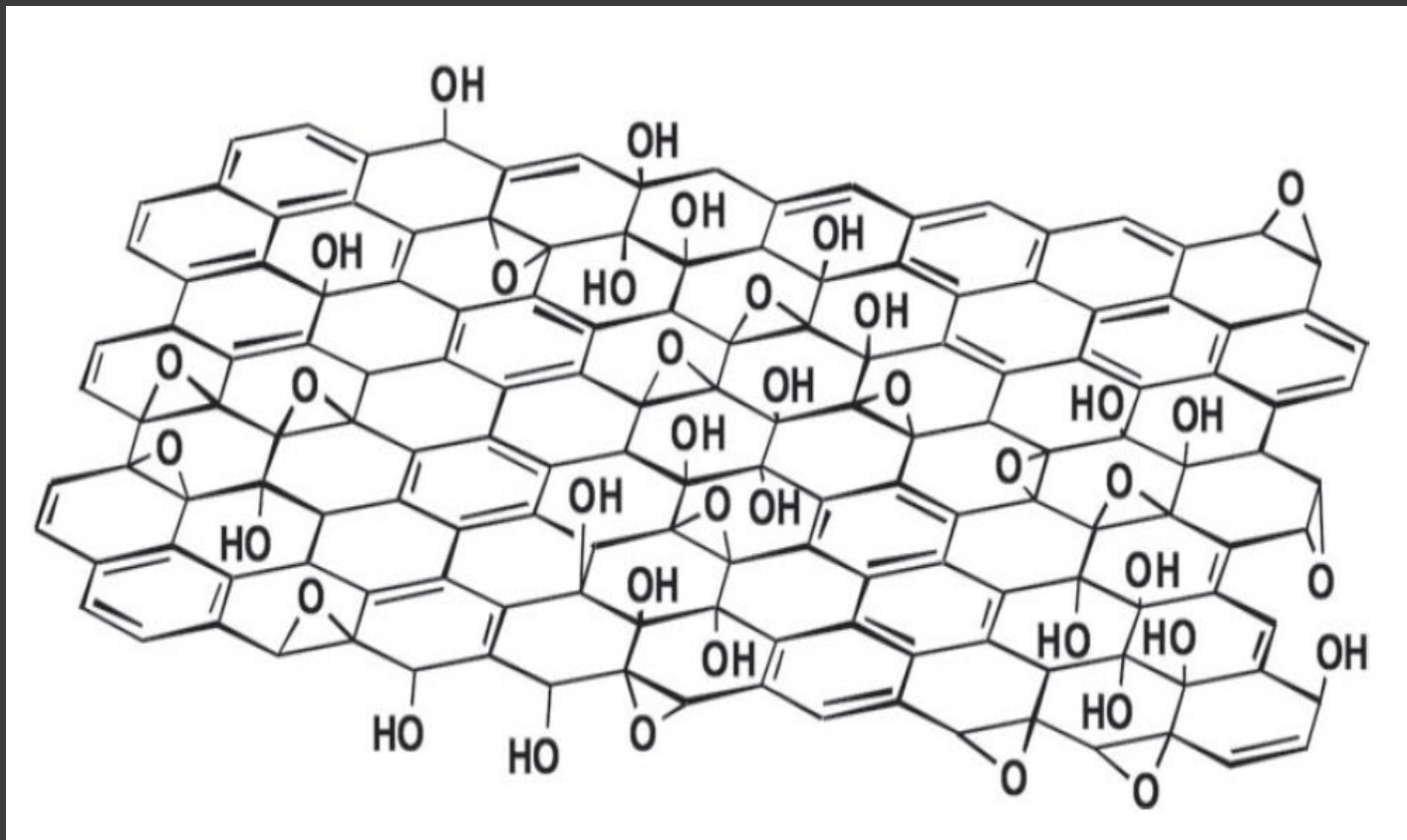


Fan, X., Peng, W., Li, Y., Li, X., Wang, S., Zhang, G. and Zhang, F. (2008), Deoxygenation of Exfoliated Graphite Oxide under Alkaline Conditions: A Green Route to Graphene Preparation. *Advanced Materials*, 20: 4490–4493

- 1) Exfoliação mecânica
- 2) Exfoliação química
- 3) CVD (sobre cobre)
- 4) Remoção do silício do carbeto de silício

Sais de potássio: inflamabilidade ao GO

# GO chemical model

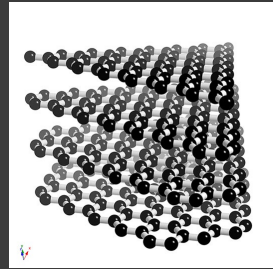
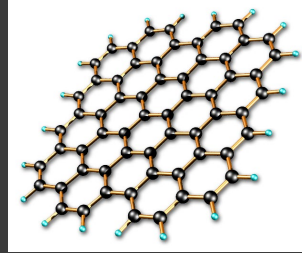


Lerf-Klinowski model

Chem. Phys. Lett., 1998, 287, 53–56

# Graphene & GO

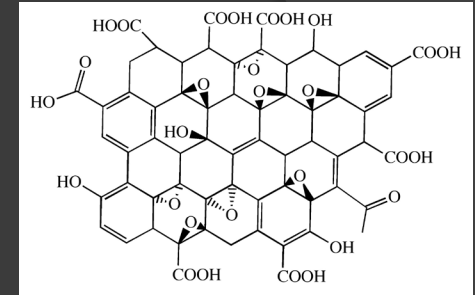
Natural Flake graphite



Exfoliated Graphite

-Chemical

-Thermochemical



Schematic of GO monolayer

Strong acids, nitrides, perchlorates, alkaline neutralization, etc...

-Graphene oxide (-OH, COO, COOH, etc...)

- Single-atom-thick sheet of carbon with oxygen containing functional groups decorating its basal plane and edges sites

-Water soluble/monolayer (hydrophilic)

-Widely used as precursor for graphene synthesis

-Area/mass: 2650 m<sup>2</sup>/g

where  $\rho \sim 1.8 \text{ g/cm}^3$  (GO density)  
 $d \sim 0.42 \text{ nm}$  (GO thickness)

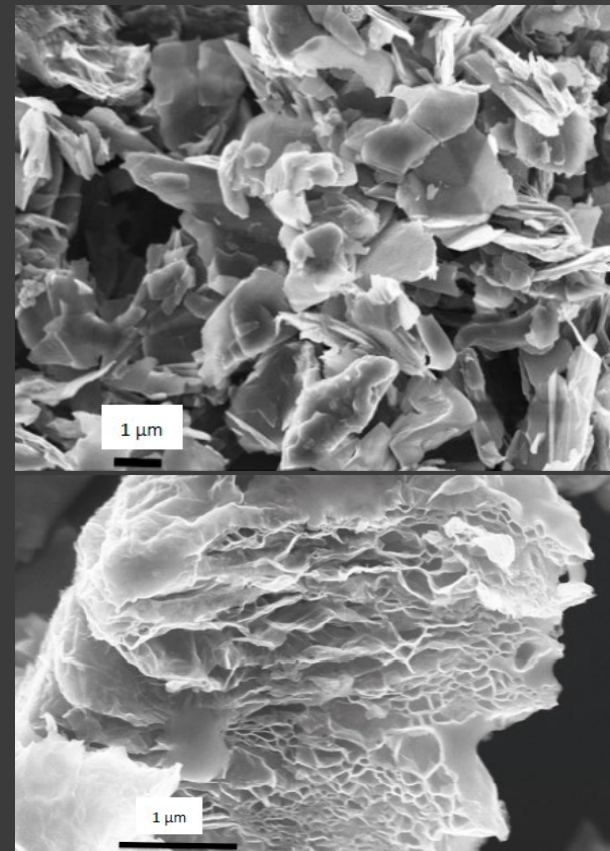
$$\frac{\text{Area}}{\text{mass}} = \frac{2LW}{\rho LWd} = \frac{2}{\rho d}$$

Dinkin et al. Nature 448(26),2007

New opportunities for castables, resin integration, pre-mixing, etc.

New opportunities for Improvements on thermal conduction of carbon-oxides refractories.

Material/Chemical	Chemical application	Decomposition enthalpy, [Jg <sup>-1</sup> ]
GO	Precursor for rGO	1,400-1,700
Benzoyl peroxide	Industrial chemical	1,602
Cumene hydroperoxide	Industrial chemical	1,219
Ethyl oleate ozonide	Industrial chemical	684
Trinitrotoluene (TNT)	Monopropellant and explosive	2,305
Hydrazine	Monopropellant and explosive	2,980-4,906
Nitrocellulose	Monopropellant and explosive	1,900-2,400



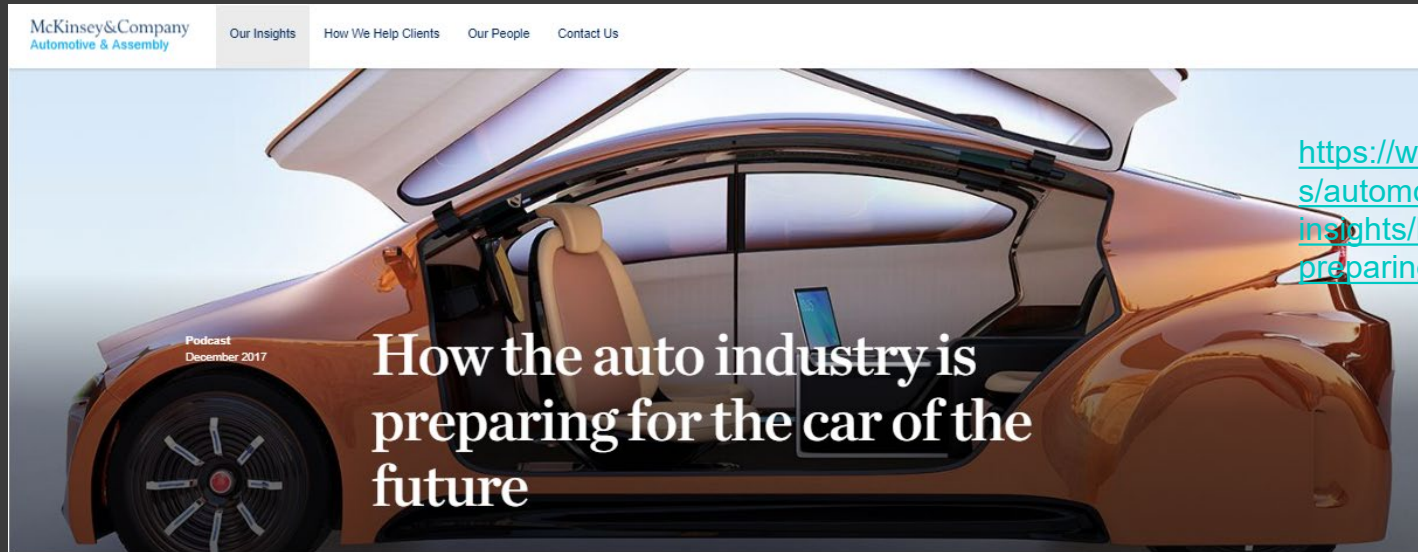
Qiu et al., Explosive thermal reduction of graphene oxide-based materials: Mechanism and safety implications, *Carbon* 72; 215-223, 2014.

r-GO before and after thermal explosive reduction.

Thanks to Dr. Indrek Kulaots and Yang Qiu

# Automotive technology ...

## And the future of automobilist industry

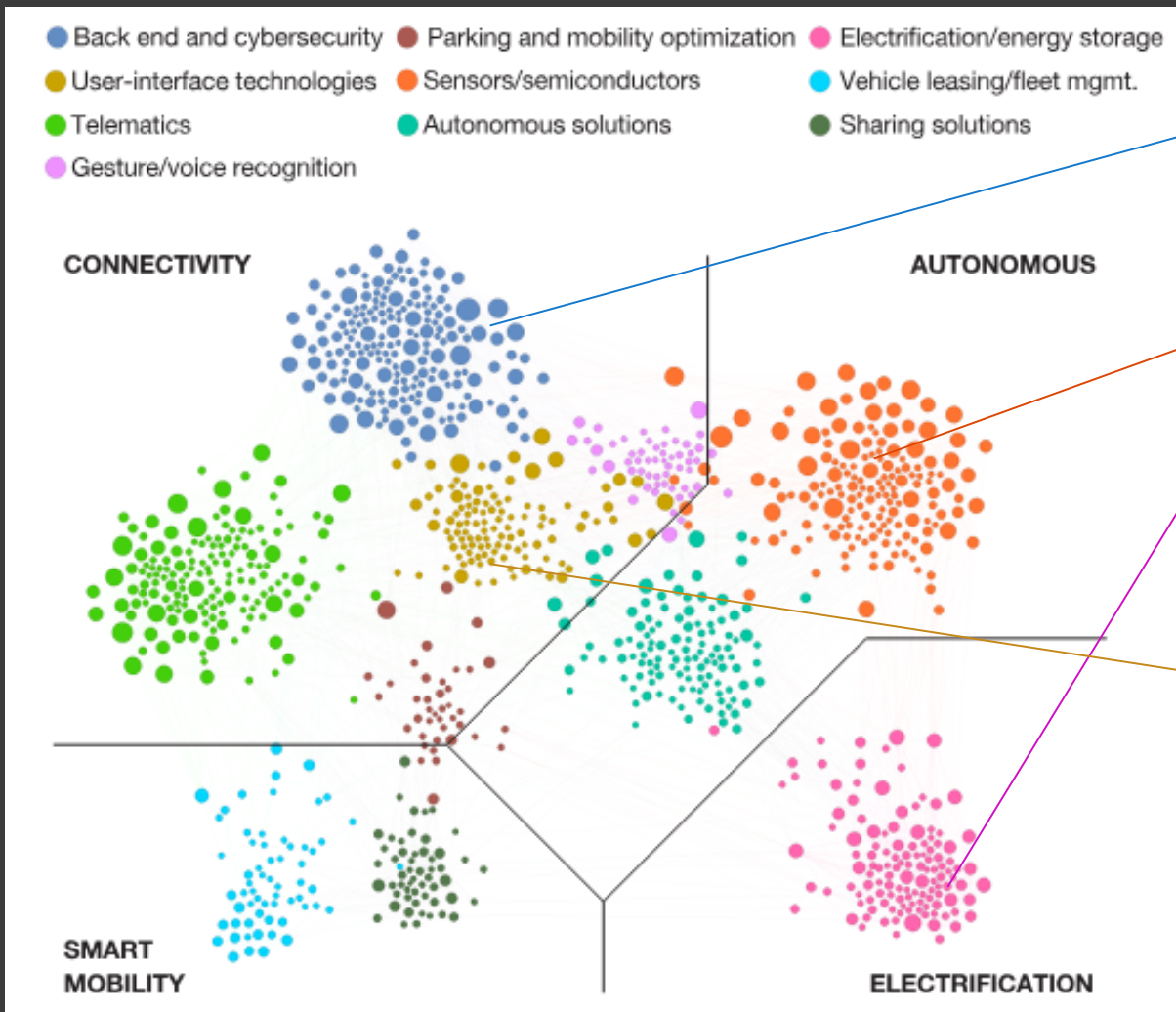


### Autonomous / Connectivity / Electrification / Smart Mobility

- Fuel economy & autonomy: Better engine + Better electronic control systems
- Combustion engines are changing to hybrids and electrical (Flexibility)
- Smart and software control
- Self driving technologies: more specialized + fast control (sensors & software)
- Safety

Hardware  
&  
Materials

Software



**New materials:**

**Reliable materials:**

- electromagnetic shield;
- fast devices;

**New sensors:**

**Recovery energy, autonomy & power:**

- batteries & supercapacitors;

**Smart coatings: touch technology**

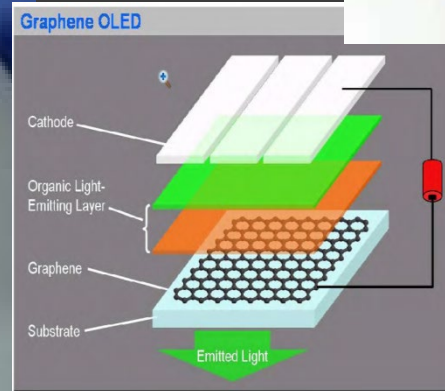
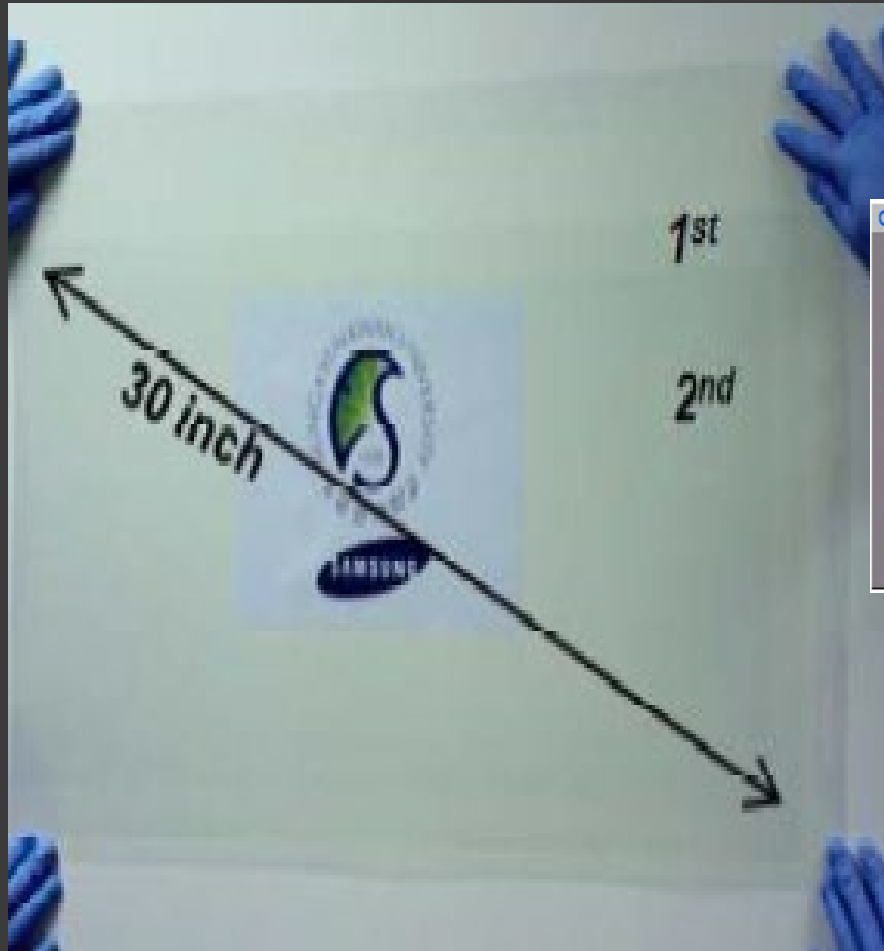
- smart glass (Man-Machine interfaces)



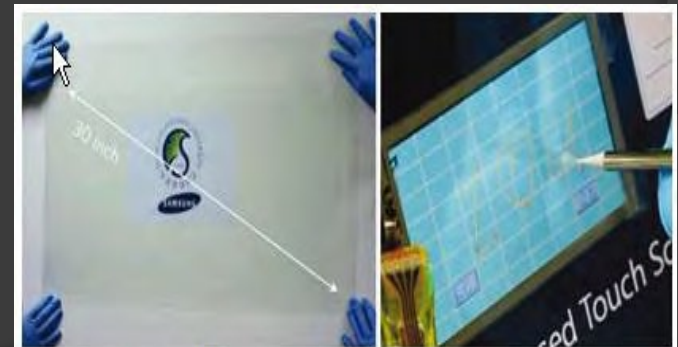
**(nano)-carbon based materials !**



# Touch capacity and sensibility

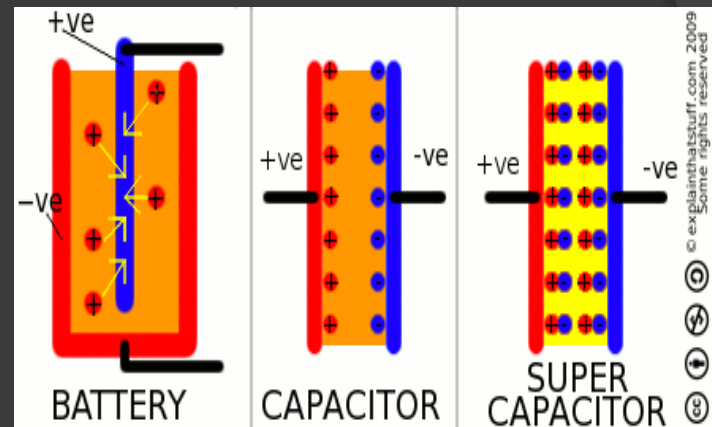
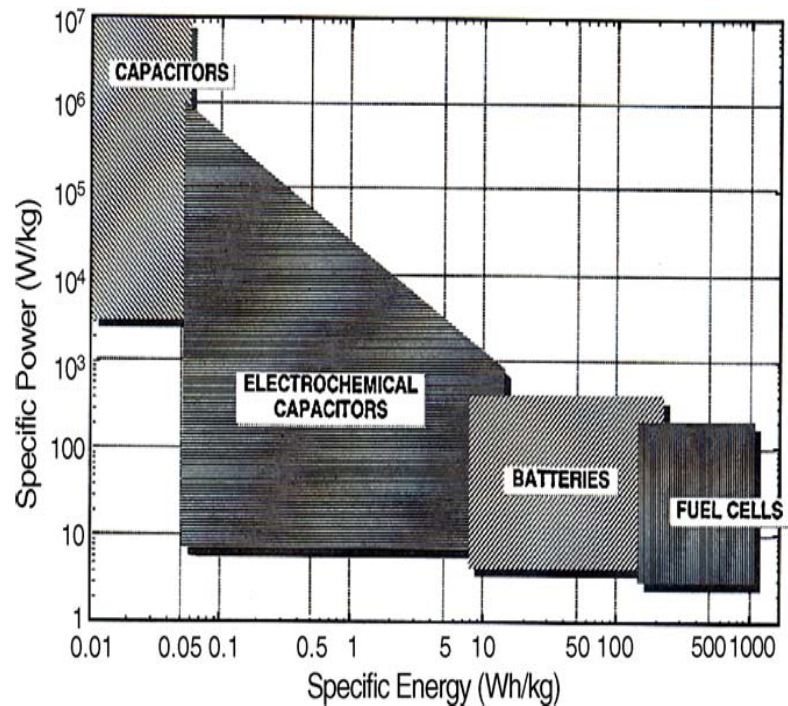


**Carbon-Based Supercapacitors  
 Produced by Activation of Graphene**  
*Science* 24 June 2011: 1537-  
 1541. DOI:10.1126/science.1200770



Left: A transparent graphene film transferred on a 35-inch PET sheet. Right: A graphene-based touchscreen panel connected to a computer

# Batteries, capacitors and supercapacitors



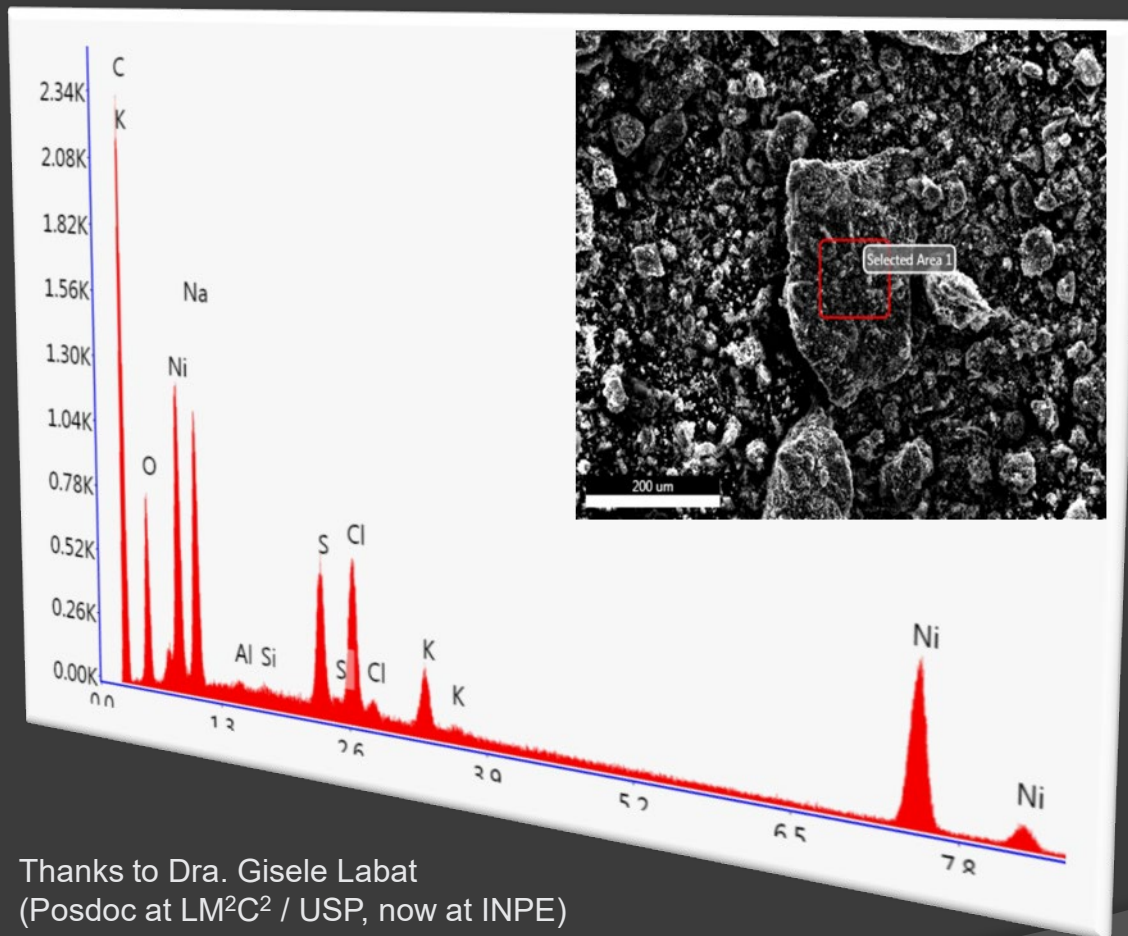
Energy density

Power density

Characteristic	Electrolytic capacitor	Carbon supercapacitor	Battery
Specific energy (Wh kg <sup>-1</sup> )	<0.1	1–10	10–100
Specific power (W kg <sup>-1</sup> )	≫10000	500–10000	<1000
Discharge time	10 <sup>-6</sup> to 10 <sup>-3</sup> s	s to min	0.3–3 h
Charging time	10 <sup>-6</sup> to 10 <sup>-3</sup> s	s to min	1–5 h
Charge/discharge efficiency (%)	~100	85–98	70–85
Cycle-life (cycles)	Infinite	>500000	~1000
Max. voltage (V <sub>max</sub> ) determinants	Dielectric thickness and strength	Electrode and electrolyte stability window	Thermodynamics of phase reactions
Charge stored determinants	Electrode area and dielectric	Electrode microstructure and electrolyte	Active mass and thermodynamics

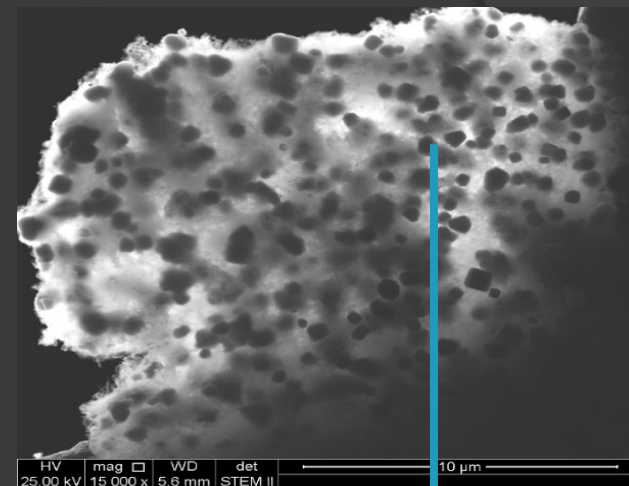
# Catalyst support (Fuel Cells)

## SEM + EDX

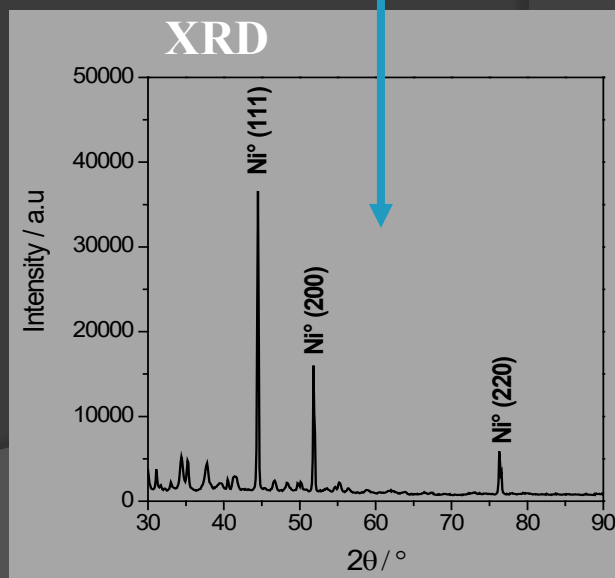


Thanks to Dra. Gisele Labat  
(Posdoc at LM<sup>2</sup>C<sup>2</sup> / USP, now at INPE)

## STEM

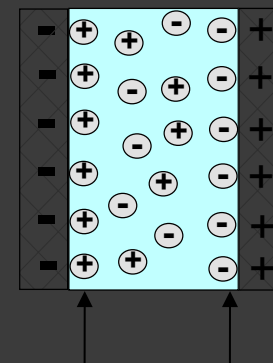
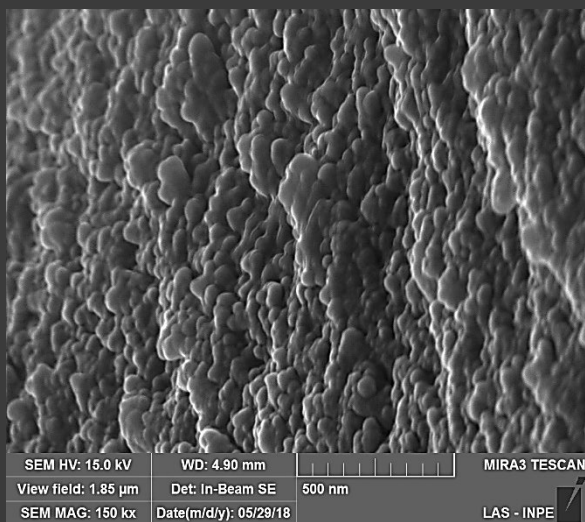
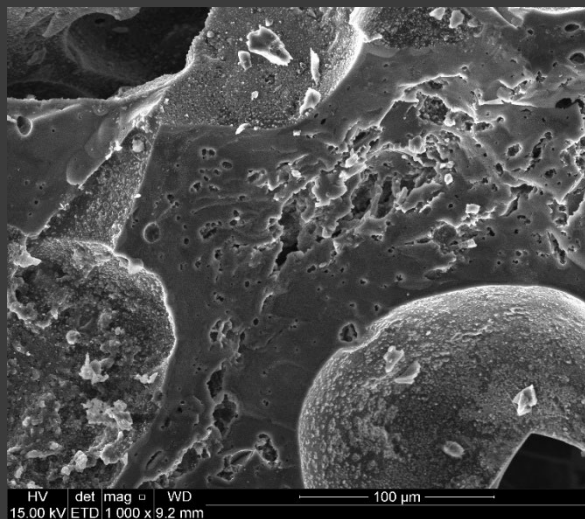


## XRD



# Energy storage (supercapacitors)

Structured Activated Carbon from Black liquor-lignin carbonization



Electric double-layer

**80 – 110 F/g**

\*Commercial activated carbon (28-100 F/g)  
 Carbon xerogels from resorcinol (10-200 F/g)

\*Obreja *et al*, International Review of Electrical Engineering, 2010 5(1):272-282

# Carbon application and future

Smart glasses



Energy storage



Structural composites



Sensors

# Graphene in automotive industry ...

## Lightweight graphene based components by Fiat

Lightweight Graphene based materials for structural applications

Graphene nanocomposites for integrated sensors

Functional textile

Sensors for pollutants detection and safety



Smart adhesives

Nanostructured TE materials (cooling/heat recovery)

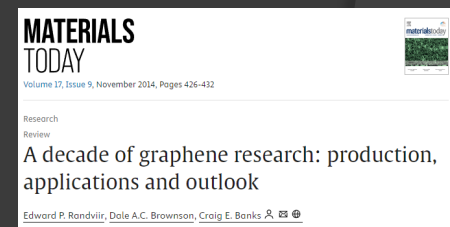
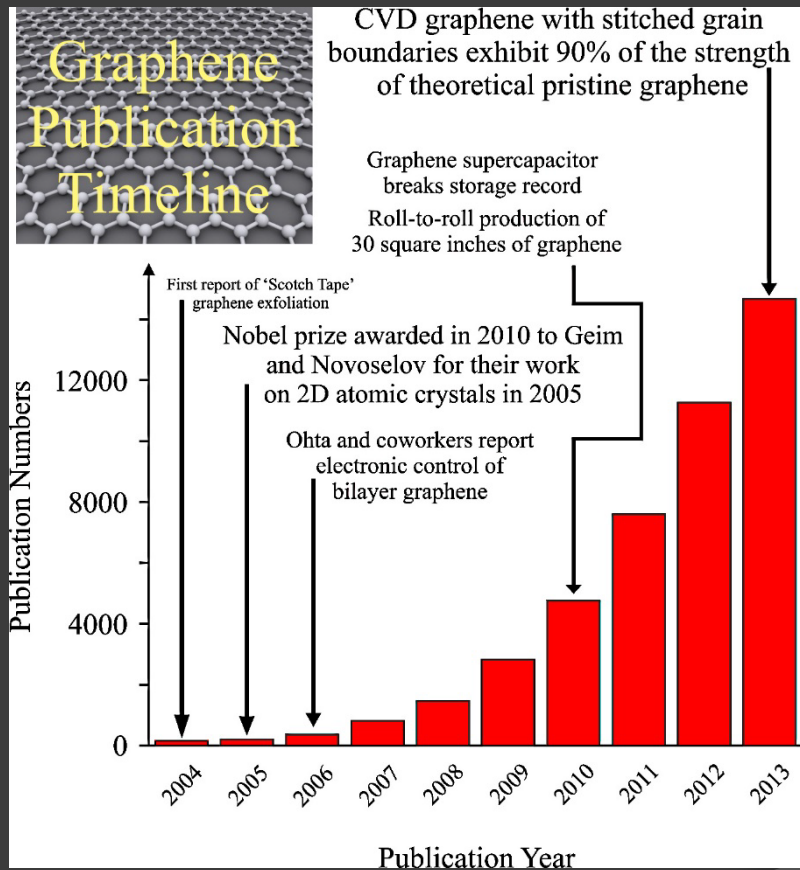
Nanofluids (friction/thermal management)

# Graphene ... 10 years ago

10 year ago ...

## Applications:

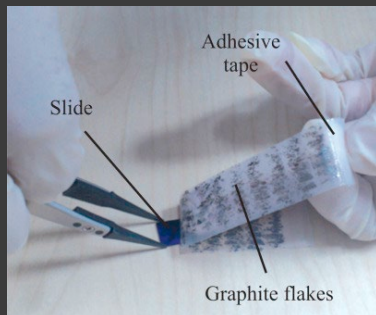
- High speed electronics
- Data storage
- LCD smart windows and OLED displays
- Supercapacitors
- Solar cells
- Electrochemical sensing



<https://www.sciencedirect.com/science/article/pii/S1369702114002144>



# Graphene ... now



R. Van Noorden  
Nature, 483 (2012), p. S32

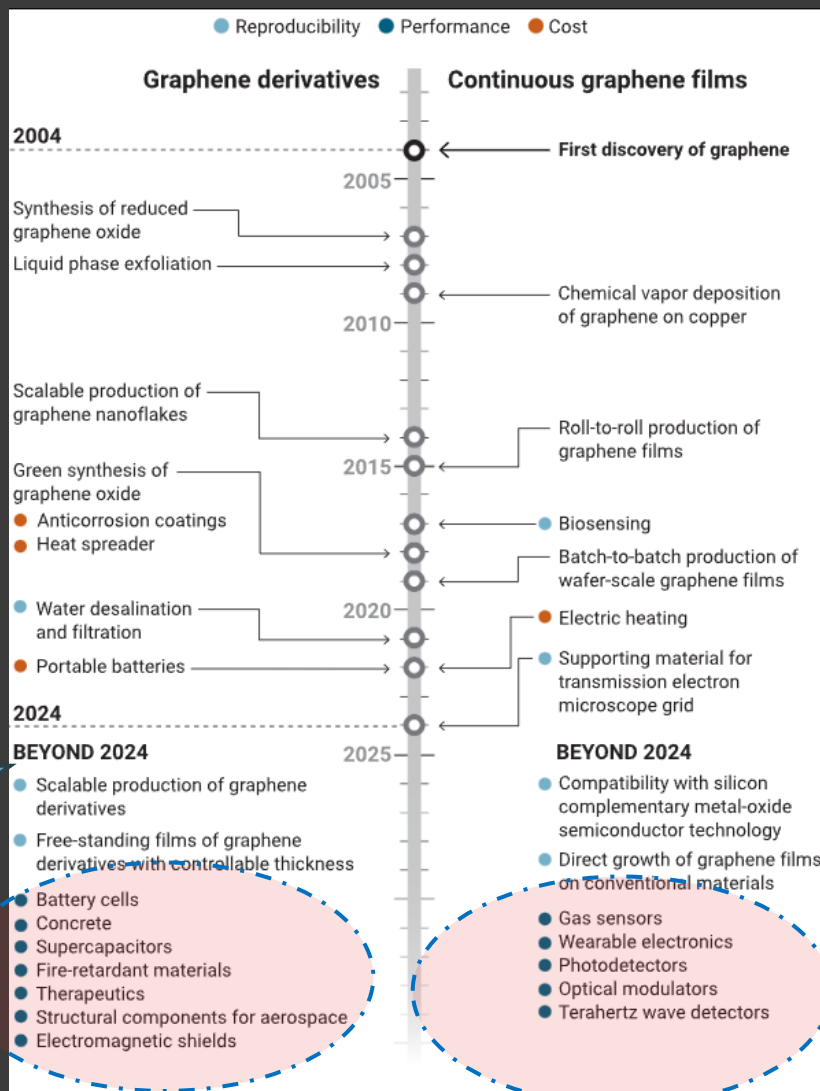
Creativity,  
Ingenuity,  
Serendipity

~ 10 years of hard work on science & technology approach to scale up and industrial ramp-up !



New materials and increasing market share of graphene-based products ... ~20 years !

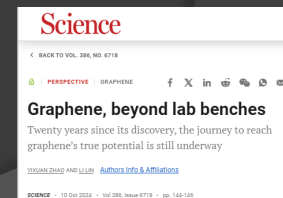
Wide range of new products compared 10 years ago !



Now ...

From lab to the market ...

- Universities
- Center of technology for industrial applications
- Spin offs
- Start ups
- New companies
- Big techs ...



<https://www.science.org/doi/10.1126/science.ads4149>



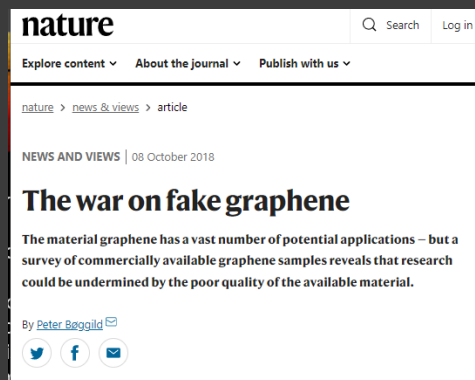
# What changed ... (after 20 years of graphene discovery !)

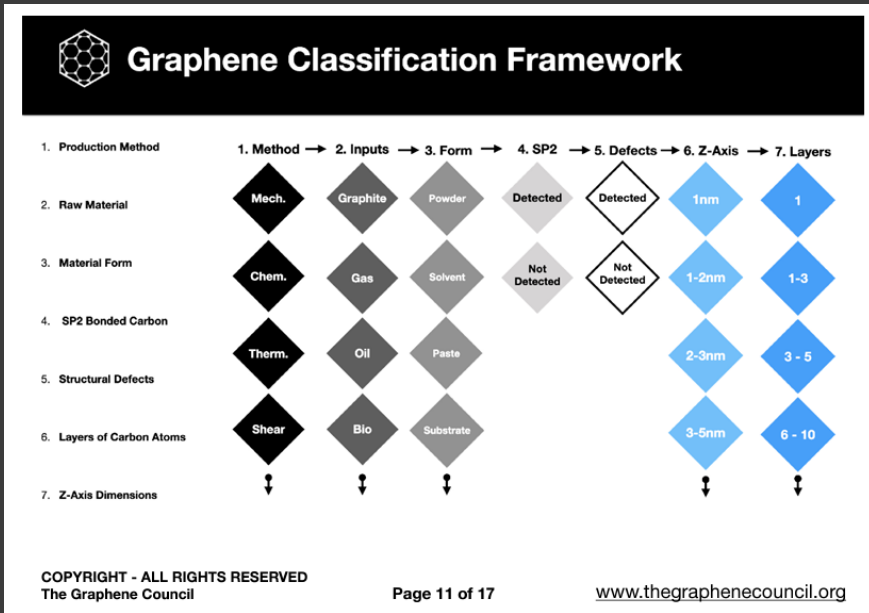
From lab to industry: new products and new production capability ! .... Lowering price ...new applications !

Understanding of graphene properties & applications ... “New graphene” material types/specifications

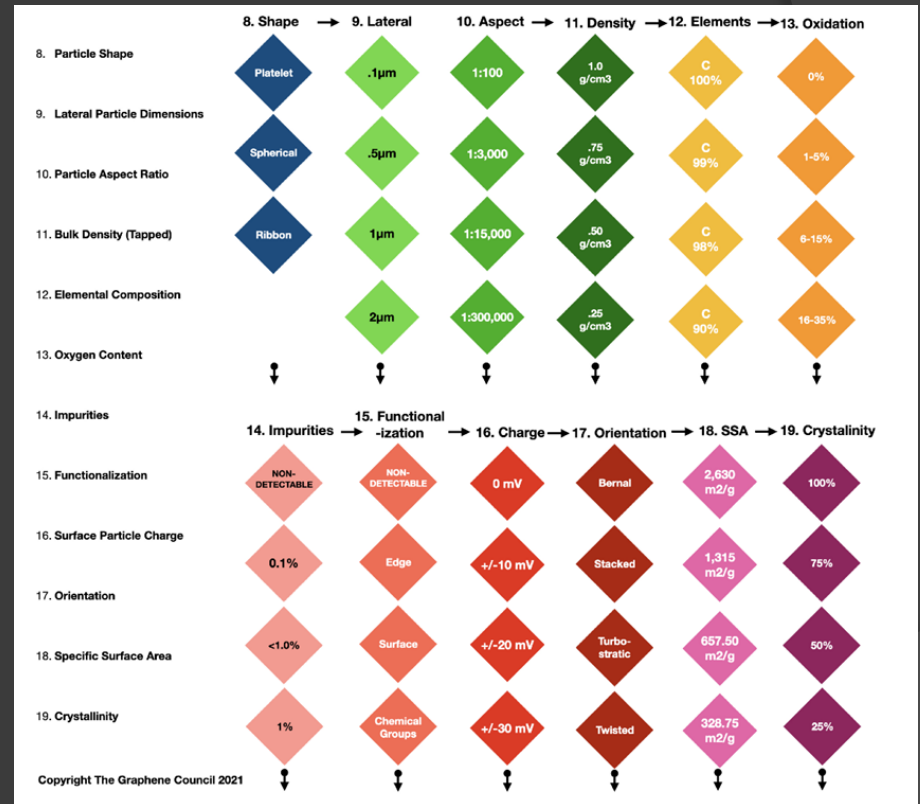
From pristine monolayers materials to dozen of “graphene family” materials ...mono, bi, last then 10 layers, more than 10 layers, nano graphite , oxidized graphite (many oxidation grades), GO, reduced GO, and so on !

<https://www.nature.com/articles/d41586-018-06939-4>





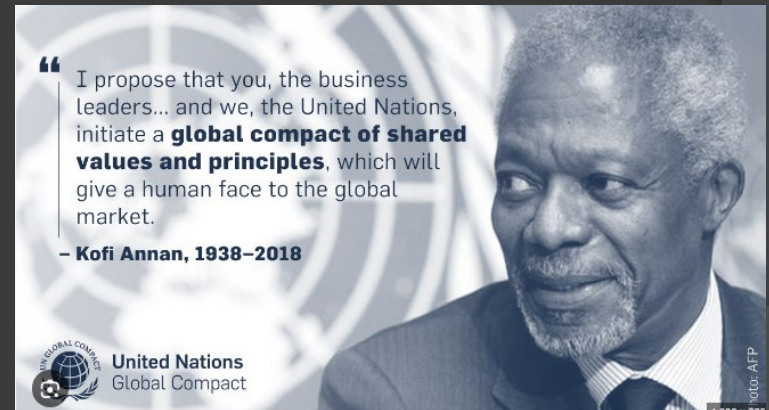
chrome-extension://efaidnbmnnnibpcajpcglclefindmkaj/https://cdn.ymaws.com/www.thegraphenecouncil.org/resource/resmgr/standards/report/gcf\_public\_version0.1.pdf



# ESG : Environmental, Social & Governance

ESG is a 20 years old concept !

It's came from UN (Kofi Annan UN secretary general )



2004 report titled "Who Cares Wins",

ESG give to the business:

- Clear PURPOSES

- LONG TERM VISION

- INTREGRATED VISION

- VISIBILITY



People & Community

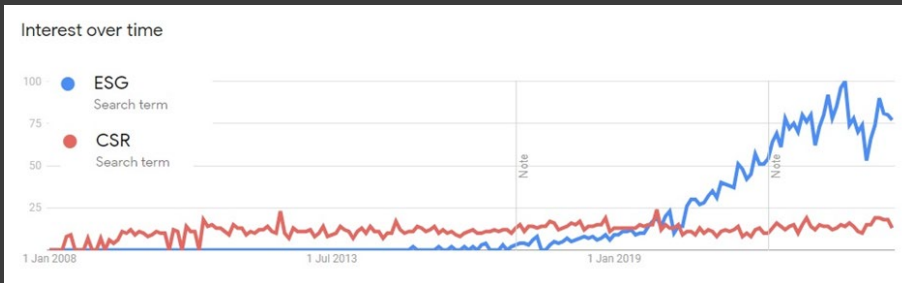


Planet



Business performance

Companies that perform better about these issues can increase shareholder value



ESG: Environmental, Social & Governance  
CSR: Corporate Social Responsibility



Financial Sector Initiative  
*Who Cares Wins*

### Graphical summary of key recommendations

#### Overall goals:

- Stronger and more resilient financial markets
- Contribution to sustainable development
- Awareness and mutual understanding of involved stakeholders
- Improved trust in financial institutions



## S - Social

- Race/Colour Diversity
- Women in Management
- Inclusion
- Health
- Equity
- Training
- Safety
- Inequality



## E - Environmental

- Environmental Education and Carbon Market
- Reforestation
- Renewable Energies

## G - Governance

- Government/Regulation
- Corporate Governance
- ESG Investments
- Innovation
- Compliance
- Risk Management
- Accountability

# The balance between E & S. The key point of G!

Remember ... ESG came from a vision of UN working together of most influential finance companies, bank and market people, such like: [ABN AMRO](#), [Bank N.V.](#), [AXA](#), [Banco do Brasil](#), [Sarasin & Partners LLP](#), [BNP Paribas](#), [Credit Suisse](#), [Deutsche Bank](#), [Goldman Sachs](#), [HSBC](#), [Morgan Stanley](#), [UBS](#) and [Westpac](#) with more than 500 representatives from business and civil society !

Governance is the key point to evaluate, control and give vision and directions for the ESG platform. It is the balance for a good implementation of S (Social) and E (Environmental) programs based on accountability, performance, measurements, goals and achievements !

Without metrics and vision, it is impossible to implement a coherent ESG program!





<https://www.un.org/sustainabledevelopment/news/communications-material/>



New electronic materials:  
Sensors, resistive memories,  
smart screens, etc.

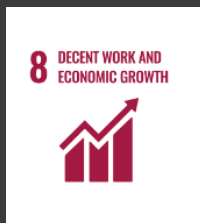


New plastic composites: high  
mechanical properties, less material  
use, recycling technologies based on  
graphene materials: chemical  
adsorbents for metal recovery



New medical treatments: graphene-  
based scaffolds (bone replacement  
with stem cell technology, graphene  
tissue engineering,  
Soft 3D printed human tissue, etc.

New materials: OPC (ordinary  
Portland cement) with graphene



New companies, tech business,  
new products, new materials & Applications

# ESG – SDG & Graphene Applications

Graphene for wind blades composites  
Graphene based solar cells  
(...)



High efficiency membranes for  
water filtration and treatment



Graphene for pollutes  
removing & treatment ,  
soil fertilizing, ...



Low carbon footprint  
technologies: solar cells,  
New batteries,  
carbon capture & storage,  
Synthetic photosynthesis,  
...



Water pollution control technologies

And here ? ... It is about good  
**Leadership & Vision !**

... Simple like that !!!!!



**Communication**  
**Ethics**  
**Responsibility**  
**Empathy**  
**Actions**  
**Integrity**



# There is only one Earth ! ...

You should know by now ... **There is no plan B !**



2013: Earth as seen through Saturn's rings (Cassini)

Image Credit: NASA/JPL-Caltech/Space Science Institute

Earth as seen  
from Saturn's  
rings looks like  
a blue  
luminescent  
carbon dot !!!



Image Credit: NASA/Goddard/Arizona State University

2015: Earthrise 2.0 (Lunar Reconnaissance Orbiter)

<https://explorer1.jpl.nasa.gov/galleries/earth-from-space/>



# TLMC<sub>5</sub>

## Quinto Taller Latinoamericano de Materiales de Carbono

*Thanks*

*धन्यवाद*

*Aloha*

*Merci*

*Gracias*

*谢谢*

*Danke*

*Grazie*

*Teşekkürler*

*Dziękuję*

*Спасибо (Spasibo)*

*Obrigado*

Let's talk with us about carbon ! <https://abcarb.org.br/>

Brazilian Carbon Association / ABCARB – President  
Senior Member of AIChE – American Institute of Chemical Engineers  
Strategic Decision and Risk Management Certificate Professional (Stanford University)  
Visiting Scholar at Brown University (LEHN / School of Engineering)  
Academic Visitor at Imperial College London (Department of Materials)  
Postdoctoral Fellow at the University of Sheffield/UK (Department of Materials Science & Engineering)  
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MBA in Environment Technologies & Management from Polytechnic School of Engineering (POLI-USP)  
Certified Safety Professional from Federal University of Minas Gerais (UFMG)  
Metallurgical Engineer from Federal University of Minas Gerais (UFMG)



Education changes people  
People change the world !